WO9203421

Publication Title:

ARTHROPODICIDAL PYRAZOLINES, PYRAZOLIDINES AND HYDRAZINES

Abstract:

Abstract of WO9203421

103f Arthropodicidal pyrazoline, pyrazolidine and hydrazine compounds, including all their geometric and stereoisomers, agriculturally suitable salts thereof and compositions containing them; and a method for controlling arthropods employing said compounds which are represented by formulae (I) and (II) wherein Q, X, X&It;1>, Y and G are as defined in the text. Data supplied from the esp@cenet database - Worldwide

Courtesy of http://v3.espacenet.com

This Patent PDF Generated by Patent Fetcher(TM), a service of Stroke of Color, Inc.



WORLD INTELLECTUAL PROPERTY ORGANIZATION International Bureau



INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification 5: C07D 231/54, 491/04, 487/04 C07D 491/06, 471/06, 495/16		(11) International Publication Number:	WO 92/03421
C07D 491/06, 471/06, 493/16 C07D 471/16, 403/12, 405/12 C07D 498/04, A01N 43/56 C07D 495/04, 471/04, 495/06	A2		
C07D 491/16, 495/16, 401/12 C07D 409/12, 513/04 C07F 9/6503		(43) International Publication Date:	5 March 1992 (05.03.92)

PCT/US91/05334 (21) International Application Number: 1 August 1991 (01.08.91) (22) International Filing Date:

(30) Priority data:

7

569,044 573,954 17 August 1990 (17.08.90) US US 27 August 1990 (27.08.90) 595,151 9 October 1990 (09.10.90)

(60) Parent Applications or Grants

(63) Related by Continuation US 569,044 (CIP) 17 August 1990 (17.08.90) 573,954 (CIP) 27 August 1990 (27.08.90) 595,151 (CIP) Filed on US Filed on Filed on 9 October 1990 (09.10.90)

(71) Applicant (for all designated States except US): E.I. DU PONT DE NEMOURS AND COMPANY [US/US]; 1007 Market Street, Wilmington, DE 19898 (US).

(72) Inventors; and

(72) Inventors; and
(75) Inventors/Applicants (for US only): HARRISON, Charles, Richard [US/US]; 137 Aspen Drive, Newark, DE 19702 (US). LETT, Renee, Marie [US/US]; 4758 Weatherhill Drive, Wilmington, DE 19808 (US). McCANN, Stephen Frederick [US/US]; 2012 Christiana Meadows, Bear, DE 19701 (US). SHAPIRO, Rafael [US/US]; 1415 Fresno Road, Wilmington, DE 19803 (US). STEVENSON, Thomas, Martin [US/US]; 103 Iroquois Court, Newark, DE 19702 (US) DE 19702 (US).

(74) Agents: COSTELLO, James, A. et al.; E.I. du Pont de Nemours and Company, Legal/Patent Records Center, 1007 Market Street, Wilmington, DE 19898 (US).

(81) Designated States: AT (European patent), BE (European patent), CH (European patent), DE (European patent), DK (European patent), ES (European patent), FR (European patent), GB (European patent), GR (European patent), IT (European patent), JP, LU (European patent), NL (European patent), SE (European patent), US.

Without international search report and to be republished upon receipt of that report.

(54) Title: ARTHROPODICIDAL PYRAZOLINES, PYRAZOLIDINES AND HYDRAZINES

(I)

(II)

(57) Abstract

Arthropodicidal pyrazoline, pyrazolidine and hydrazine compounds, including all their geometric and stereoisomers, agriculturally suitable salts thereof and compositions containing them; and a method for controlling arthropods employing said compounds which are represented by formulae (I) and (II) wherein Q, X, XI, Y and G are as defined in the text.

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

TA	Austria	ES	Spain	MG	Madagascar
AU	Australia	F)	Finland	ML	Mali
BB	Barbados	FR	France	MN	Mongolia
BE	Belgium	GA	Gabon	MR	Mauritania
BF	Burkina Faso	GB	United Kingdom	MW	Malawi
BG	Bulgaria	CN	Guinca	NL	Netherlands
BJ	Benin	GR	Greece	NO	Norway
BR	Brazil	HU	Hungary	PL	Poland
CA	Canada	IT	Italy	RO	Romania
CF	Central African Republic	JР	Japan	SD	Sudan
CG	Congo	KP	Democratic People's Republic	SE	Sweden
CH	Switzerland		of Korea	SN	Senegal
CI	Côte d'Ivoire	KR	Republic of Korea	su+	Soviet Union
CM	Cameroon	L.1	Liechtenstein	TD	Chad
cs	Czechoslovakia	LK	Sri Lanka	TG	Togo
DE*	Germany	LU	Luxembourg	US	United States of America
DK	Denmark	MC	Monaco		
			•		

⁺ Any designation of "SU" has effect in the Russian Federation. It is not yet known whether any such designation has effect in other States of the former Soviet Union.

15

20

TITLE

ARTHROPODICIDAL PYRAZOLINES, PYRAZOLIDINES AND HYDRAZINES BACKGROUND OF THE INVENTION

10 Field of the Invention

This invention concerns arthropodicidal pyrazolines, pyrazolidines and hydrazines, compositions containing them, and a method for their use to control arthropods.

State of the Art

WO 88/07,994 and EPA 330,678 disclose insecticidal pyrazolines. U.S. 4,547,524 discloses benzoyl hydrazone derivatives as insecticides. WO 88/00197 discloses substituted semicarbazones derived from chromanones and thiochromanones as insecticide intermediates. EP-3,913, EP 26,040 and EP 254,461 disclose substituted hydrazone insecticides. J. Org. Chem., 1987, 52, 2277 discloses pyrazolidines as does Chem. Soc., Jap., 55, 2450 (1982).

J. Ind. Chem. Soc., 37, pages 443 to 450 (1960) discloses a compound of the formula:

25

30

but no utility therefor.

Vaughn, <u>J. Org. Chem.</u>, <u>20</u> (1955), pages 1619 to 1626, discloses 1,5-diphenyl-2-pyrazoline-3-carboxamide. No utility is given for the disclosed compound which, in any event, does not suggest a compound of the instant invention.

Harhash et al., <u>J. Heterocyclic Chem.</u>, <u>21</u> (1984), at page 1013, discloses the preparation of five pyrazoline compounds, none of which is

5 disclosed in the instant application. No utility is given for any of said compounds:

where R/Ar are C_6H_5/C_6H_5 ; $CO_2C_2H_5/C_6H_5$; $C(O)NHC_6H_5/C_5H_5$; $CH=CHC_6H_5/C_6H_5$; and $CH_3/4-NO_2-C_6H_4$.

U.S. 4,070,365 discloses insecticidal compounds of the formula

15

wherein X is halogen; and Y is halogen, NO2 or alkyl.

EP 153,127 discloses insecticidal compounds of the formula

20

15

5 wherein

A is unsubstituted or substituted phenyl; B is unsubstituted or substituted phenyl; U is O, S or NR; Y is alkyl, unsubstituted or substituted phenyl, or C(X)G; Z is H, cycloalkyl, unsubstituted or substituted phenyl R⁴-Q; X is O or S; and G and R⁴-Q are broadly defined.

SUMMARY OF THE INVENTION

The invention pertains to compounds of Formula I and II, including all geometric and stereoisomers, agriculturally suitable salts thereof, agricultural compositions containing them and their use for the control of arthropods in both agronomic and non-agronomic uses. The compounds are:

20

wherein:

Q is selected from the group

Q-1
$$(R^2)_n$$

$$(R^2)_n$$

$$(R^2)_n$$

$$Q-2$$

$$(R^2)_n \qquad \qquad \begin{array}{c} A \qquad \qquad R^3 \\ R^4 \qquad \qquad \\ N-N \qquad \qquad \\ Y^1 \end{array}$$

Q-3

$$(\mathbb{R}^2)_{n} \longrightarrow \mathbb{N}$$

$$\mathbb{N}$$

$$\mathbb{N}$$

. 4

Q-4

$$(R^2)_n \xrightarrow{H} N - N$$

Q-5

$$(\mathbb{R}^2)_n \bigvee_{\mathbb{R}^{18}}^{\mathbb{N}} \mathbb{R}^3$$

Q-6

$$\begin{array}{c|c} & & & & R^3 \\ & & & & \\ & & & & \\ R^{16} & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ \end{array}$$

Q-7

$$(\mathbb{R}^2)_n \xrightarrow{\mathbb{R}^3} \mathbb{R}^4$$

Q-8

WO 92/03421

5

PCT/US91/05334

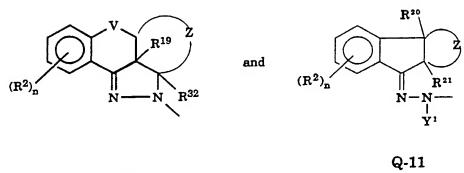
$$(\mathbb{R}^2)_n \xrightarrow[]{\mathbb{R}^3} \mathbb{R}^3$$

5

Q-9

10

15



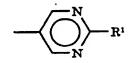
Q-10

A is selected from the group CH_2 , CH_2CH_2 , O, S, S(O), $S(O)_2$,

 $N(R^5)$, OCH_2 , SCH_2 , $N(R^5)CH_2$, substituted CH_2 , and substituted CH_2CH_2 , the substituents independently selected from 1-2 halogen and methyl;

G is selected from the group

$$\left\langle \begin{array}{c} \\ \\ \\ \end{array} \right\rangle$$
 $\left\langle \begin{array}{c} \\ \\ \end{array} \right\rangle$



G-1

G-2

G-3

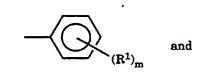
$$\langle \sum_{N-N} - R^1 \rangle$$

G-4

G-5

G-6

$$s \sim R^1$$



G-7

G-8

G-9

5

G-10

10

X is selected from the group O, S and N-X2;

 X^{1} is selected from the group Cl, Br, OR6, SR6 and N(R6)R7;

 X^2 is selected from the group R^6 , OH, OR⁶, CN, SO₂R⁶, SO₂Ph,

5	optionally substituted with R^9 ;
	Y and Y 1 are independently selected from the group H, $^{\mathrm{C}_{1}\text{-}\mathrm{C}_{6}}$
10	alkyl, benzyl, C_2 - C_6 alkoxyalkyl, CHO, C_2 - C_6 alkylcarbonyl, C_2 - C_6 alkoxycarbonyl, C_2 - C_6 haloalkylcarbonyl, C_1 - C_6 alkylthio, C_1 - C_6 haloalkylthio, phenylthio, R^{10} OC(O)N(R^{11})S-and R^{12} (R^{13})NS-;
	$ m R^{1}$, $ m R^{2}$ and $ m R^{14}$ are independently selected from the group H,
15	C ₁ -C ₆ alkyl, C ₁ -C ₆ haloalkyl, C ₂ -C ₆ alkenyl, C ₂ -C ₆ haloalkenyl, C ₂ -C ₆ alkynyl, C ₃ -C ₆ haloalkynyl, C ₂ -C ₆ alkoxyalkyl, C ₂ -C ₆ alkylthioalkyl, C ₁ -C ₆ nitroalkyl, C ₂ -C ₆ cyanoalkyl, C ₃ -C ₈ alkoxycarbonylalkyl, C ₃ -C ₆ cycloalkyl, C ₃ -C ₆
20	halocycloalkyl, halogen, CN, N ₃ , SCN, NO ₂ , OR ¹⁵ , SR ¹⁵ , S(O)R ¹⁵ , S(O) ₂ R ¹⁵ , OC(O)R ¹⁵ , OS(O) ₂ R ¹⁵ , CO ₂ R ¹⁵ , C(O)R ¹⁵ , C(O)N(R ¹⁵)R ¹⁶ , SO ₂ N(R ¹⁵)R ¹⁶ , N(R ¹⁵)R ¹⁶ , N(R ¹⁶)C(O)R ¹⁵ , OC(O)NHR ¹⁵ , N(R ¹⁶)C(O)NHR ¹⁵ , N(R ¹⁶)SO ₂ R ¹⁵ , phenyl
	optionally substituted with 1 to 3 substituents independently selected from W and benzyl optionally substituted with 1 to 3 substituents independently selected from W; or when m, n or p is 2, $(R^1)_2$ can be taken together, or $(R^2)_2$ can be taken together
25	or (R ¹⁴) ₂ can be taken together as -OCH ₂ O, -OCF ₂ O, OCH ₂ CH ₂ O, CF ₂ CF ₂ O, -CH ₂ C(CH ₃) ₂ O- or -OCF ₂ CF ₂ O- to
	form a cyclic bridge between adjacent atoms on the same ring;
	R^3 is selected from the group R^a , R^b and J ;
	R^a is selected from the group H, C_1 - C_6 alkyl, C_1 - C_6 haloalkyl,
30	C ₂ -C ₆ alkenyl, C ₂ -C ₆ haloalkenyl, C ₂ -C ₆ alkynyl, OR ³⁵ , C ₂ -C ₆ alkoxyalkyl, C ₂ -C ₆ cyanoalkyl, C ₃ -C ₈ alkoxycarbonylalkyl, CO ₂ R ¹⁵ , C(O)R ¹⁵ , C(O)N(R ¹⁵)R ¹⁶ , C(S)N(R ¹⁵)R ¹⁶ , C(S)R ¹⁵ , C(S)SR ¹⁵ , phenyl, phenyl substituted with $(R^{14})_p$, benzyl and

5	benzyl substituted with 1 to 3 substituents independently selected from W;
	R^b is selected from the group CN, N_3 , NO_2 , halogen, $N(R^{22})R^{23}$,
10	C(R ³³)=N-O-R ³⁴ , C ₂ -C ₆ epoxyalkyl optionally substituted with a group selected from C ₁ -C ₃ alkyl, phenyl optionally substituted with W, C(O)R ²⁴ , Si(R ²⁸)(R ²⁹) R ³⁰ , SR ²⁷ , S(O)R ²⁷ , SO ₂ R ²⁷ and -P(O)(OR ²⁷) ₂ , CO ₂ R ²⁴ ; or R ^b is C ₁ -C ₆ alkyl substituted with a group selected from C(O)N(R ²⁵)R ²⁶ , C(O)R ²⁵ , SR ²⁷ , S(O)R ²⁷ , SO ₂ R ²⁷ , SCN, C ₁ -C ₂ haloalkoxy, Si(R ²⁸)(R ²⁹)R ³⁰ , N(R ²²)R ²³ , NO ₂ , OC(O)R ²⁵ , Si(R ²⁸)(R ²⁹)R ³⁰ , -P(O)(OR ²⁷) ₂
15	and J;
	J is selected from the group consisting of saturated, partially
20	saturated and aromatic 5- or 6-membered substituted or unsubstituted heterocyclic rings, bonded through carbon or nitrogen, containing 1-4 heteroatoms independently selected from the group consisting of 0-2 oxygen, 0-2 sulfur and 0-4 nitrogen and optionally containing one carbonyl wherein the substituent groups are selected from W;
	R^4 is selected from the group H, C_1 - C_6 alkyl, C_1 - C_6 haloalkyl,
2 5	C ₂ -C ₆ alkenyl, C ₂ -C ₆ haloalkenyl, C ₂ -C ₆ alkynyl, C ₃ -C ₆ haloalkynyl, C ₂ -C ₆ alkoxyalkyl, C ₂ -C ₆ cyanoalkyl, phenyl optionally substituted with (R ¹⁴) _p and benzyl optionally substituted with 1 to 3 substituents independently selected from W;
	R^5 is selected from the group H, C_1 - C_4 alkyl, C_1 - C_4 haloalkyl,
30	C_2 - C_4 alkenyl, C_2 - C_4 haloalkenyl, SR^{15} , $S(O)R^{15}$, $S(O)_2R^{15}$, $C(O)R^{15}$, CO_2R^{15} , $C(O)N(R^{15})R^{17}$, $C(S)N(R^{15})R^{17}$, $C(S)R^{15}$, $C(S)OR^{15}$, $P(O)(OR^{15})_2$, $P(S)(OR^{15})_2$, $P(O)(R^{17})OR^{15}$, $P(O)(R^{15})(SR^{17})$, and optionally substituted phenyl and
	runk-yok-'i. and optionally substituted dienvi and

5	optionally substituted benzyl wherein the substituent(s) are selected from W;
	R6 is selected from the group C1-C3 alkyl, benzyl optionally
10	substituted with R ⁹ , C ₂ -C ₄ alkenyl, C ₂ -C ₄ alkynyl, C ₁ -C ₃ haloalkyl, C ₂ -C ₄ haloalkenyl, C ₃ -C ₆ cycloalkyl and C ₁ -C ₃ alkyl substituted with OCH ₃ , OCH ₂ CH ₃ , NO ₂ , CN, CO ₂ CH ₃ , CO ₂ CH ₂ CH ₃ , SCH ₃ or SCH ₂ CH ₃ ;
	$ m R^7$ is selected from the group H, C ₁ -C ₄ alkyl, C ₁ -C ₄ haloalkyl,
	$\mathrm{C}_2 ext{-}\mathrm{C}_4$ alkoxycarbonyl, optionally substituted phenyl and
15	optionally substituted pyridinyl wherein the substituent(s) are selected from \mathbb{R}^9 ; or
	$ m R^6$ and $ m R^7$ when attached to the same atom can be taken together
	to form $-(CH_2)_4$ -, $-(CH_2)_5$ - or $-CH_2CH_2OCH_2CH_2$ - each of which is optionally substituted with 1 or 2 CH ₃ groups;
	${ m R}^{8}$ is selected from the group H and C ₁ -C ₄ alkyl; or
20	${ m R}^7$ and ${ m R}^8$ can be taken together to form -(CH ₂) ₄ -, -(CH ₂) ₅ - or
	-CH ₂ CH ₂ OCH ₂ CH ₂ - each of which is optionally substituted with 1 or 2 CH ₃ groups;
	$ m R^{9}$ is selected from the group halogen, CN, C1-C3 haloalkyl and
	C ₁ -C ₃ haloalkoxy;
25	R^{10} is C_1 - C_6 alkyl;
	R^{11} is C_1 - C_4 alkyl;
	$ m R^{12}$ and $ m R^{13}$ are independently $ m C_1\text{-}C_4$ alkyl; or
	$ m R^{12}$ and $ m R^{13}$ when attached to the same atom, can be taken together as -CH2CH2CH2CH2CH2- or -CH2CH2OCH2CH2-;
30	$ m R^{15}$ is selected from the group H, C1-C6 alkyl, C1-C6 haloalkyl,

5	C ₂ -C ₆ alkenyl, C ₂ -C ₆ haloalkenyl, C ₂ -C ₆ alkynyl, C ₃ -C ₆ haloalkynyl, C ₂ -C ₆ alkoxyalkyl, C ₂ -C ₆ alkylthioalkyl, C ₁ -C ₆ nitroalkyl, C ₂ -C ₆ cyanoalkyl, C ₃ -C ₈ alkoxycarbonylalkyl, C ₃ -C ₆ cycloalkyl, C ₃ -C ₆ halocycloalkyl, optionally substituted phenyl
10	and optionally substituted benzyl wherein the substituent(s) are 1 to 3 substituents independently selected from W; wherein R^{15} is other than H when: (i) R^1 , R^2 , or R^{14} is $S(O)R^{15}$, SO_2R^{15} , $OC(O)R^{15}$ or $OS(O)_2R^{15}$, (ii) R^a is $C(S)R^{15}$ or $C(S)SR^{15}$, or (iii) R^5 is not $C(O)R^{15}$, $C(O)N(R^{15})R^{17}$ or $C(S)N(R^{15})R^{17}$;
	$ m R^{16}$ is selected from the group H and $ m C_1\text{-}C_4$ alkyl; or
15	R^{15} and R^{16} , when attached to the same atom, can be taken together as -(CH ₂) ₄ -, -(CH ₂) ₅ -, or -CH ₂ CH ₂ OCH ₂ CH ₂ -;
	$ m R^{17}$ is selected from the group H, $ m C_1\text{-}C_4$ alkyl, $ m C_2\text{-}C_4$ alkenyl and
	C ₂ -C ₄ alkynyl;
	R^{18} is selected from the group H, C_1 - C_4 alkyl, C_1 - C_4 haloalkyl,
20 25	C2-C4 alkylcarbonyl, C2-C4 haloalkylcarbonyl, C2-C4 alkoxycarbonyl, C2-C4 haloalkoxycarbonyl, C2-C5 alkylaminocarbonyl, C3-C6 cycloalkyl, C3-C6 halocycloalkyl, C4-C7 alkylcycloalkyl, C4-C7 haloalkylcycloalkyl, C1-C4 alkylsulfonyl, C1-C4 haloalkylsulfonyl and SO2Ph optionally substituted with Cl, Br or CH3;
	R^{19} , R^{20} and R^{21} are independently selected from the group H,
	$\mathrm{C}_1\text{-}\mathrm{C}_3$ alkyl, $\mathrm{CO}_2\mathrm{R}^{15}$, phenyl and phenyl substituted by W;
	$ m R^{22}$ is selected from the group H, C((O)C1-C6 alkyl, CO2C1-C6
30	alkyl, optionally substituted C_1 - C_4 alkyl, optionally substituted C_2 - C_4 alkenyl, and optionally substituted C_2 - C_4 alkynyl, the substituents selected from C_1 - C_2 alkoxy, CN, $C(O)R^{31}$ and
	$C(O)_2R^{27}$;

5	\mathbb{R}^{23} is selected from the group H, \mathbb{C}_1 - \mathbb{C}_3 alkyl, phenyl
	substituted with W, benzyl and benzyl substituted with W;
	$ m R^{24}$ is selected from the group H, $ m C_1\text{-}C_4$ alkyl $ m C_2\text{-}C_4$ alkenyl and
	C ₂ -C ₄ alkynyl;
	R^{25} and R^{26} are independently selected from the group H and
10	C ₁ -C ₂ alkyl;
	R^{27} is selected from the group $C_1\text{-}C_3$ alkyl, phenyl and phenyl
	substituted with W;
	R^{28} is C_1 - C_3 alkyl;
	\mathbb{R}^{29} is \mathbb{C}_1 - \mathbb{C}_3 alkyl;
15	R^{30} is selected from the group C ₁ -C ₃ alkyl, phenyl and phenyl
	substituted with W;
	R^{31} is selected from the group H, C ₁ -C ₃ alkyl, phenyl and phenyl
	substituted by W;
	$ m R^{32}$ is selected from the group H, CH3 and CO2CH3
20	$ m R^{33}$ is selected from the group H, Cl, C ₁ -C ₄ alkyl, C ₁ -C ₂ alkoxy,
	C ₁ -C ₂ thioalkyl and CN;
	$ m R^{34}$ is selected from the group H, $ m C_1\text{-}C_4$ alkyl, $ m C_2\text{-}C_3$
	alkylcarbonyl and C2-C3 alkoxycarbonyl;
	$ m R^{35}$ is selected from the group H, $ m C_1\text{-}C_4$ alkyl, $ m C_2\text{-}C_4$ alkenyl,
2 5	C_2 - C_4 alkynyl, C_2 - C_4 alkylcarbonyl, C_2 - C_4 alkoxycarbonyl, C_2 - C_4 alkylaminocarbonyl, C_2 - C_4 dialkylaminocarbonyl, and C_1 - C_4

alkylsulfonyl;

V is selected from the group O, S, N(R⁵), CH₂ and CH₂ substituted with 1-2 CH₃;

W is selected from the group halogen, CN, NO₂, C₁-C₂ alkyl, C₁-C₂ haloalkyl, C₁-C₂ alkoxy, C₁-C₂ haloalkoxy, C₁-C₂ alkylthio, C₁-C₂ haloalkylthio, C₁-C₂ alkylsulfonyl and C₁-C₂ haloalkylsulfonyl;

Z is selected from the group (CH₂)_q, -CH₂OCH₂-, -CH₂SCH₂- and -CH₂NHCH₂-, where said groups can be optionally substituted with 1-2 CH₃;

m is 1 to 3;

15 n is 1 to 3;

p is 1 to 3;

q is 2 to 4;

u is 1 or 2;

provided that (i) when Q is Q-3, R³ is other than CN, N(R²²)R²³, SR²⁷, S(O)R²⁷ or SO₂R²⁷, (ii) when Q is Q-4, R³ is other than CN or J, and (iii) when G is G-9, R³ is other than R^a.

Exemplary values of J include



J-1



J-2



J-3

J-4

J-5



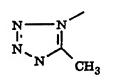
J-6



J-7



J-8



J-9



J-10



J-11



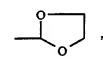
J-12



J-13



J-14



J-15

J-16

J-17



J-18

10

15

20

25

Preferred Compounds A are compounds of Formulae I and II wherein:

R¹ is selected from the group H, halogen, CN, SCN, NO₂, OR¹⁵, SR¹⁵, SO₂R¹⁵, CO₂R¹⁵, C(O)R¹⁵, C₁-C₆ alkyl, C₁-C₆ haloalkyl, C₂-C₆ alkenyl, C₂-C₆ haloalkenyl, C₂-C₆ alkynyl, C₃-C₆ haloalkynyl, C₂-C₆ alkoxyalkyl, C₂-C₆ alkylthioalkyl, C₁-C₆ nitroalkyl, C₂-C₆ cyanoalkyl, C₃-C₈ alkoxycarbonylalkyl, C₃-C₆ cycloalkyl, C₃-C₆ halocycloalkyl, phenyl optionally substituted with 1 to 3 substituents independently selected from W, and benzyl optionally substituted with 1 to 3 substituents independently selected from W:

R² and R¹⁴ are independently selected form the group H, halogen, CN, SCN, NO₂, OR¹⁵, SR¹⁵, S(O)₂R¹⁵, OC(O)R¹⁵, OS(O)₂R¹⁵, CO₂R¹⁵, C(O)R¹⁵, C(O)N(R¹⁵)R¹⁶, SO₂N(R¹⁵)R¹⁶, N(R¹⁵)R¹⁶, C₁-C₆ alkyl, C₁-C₆ haloalkyl, C₂-C₆ alkenyl, C₂-C₆ haloalkenyl, C₂-C₆ alkylthioalkyl, C₁-C₆ nitroalkyl, C₂-C₆ cyanoalkyl, C₂-C₆ alkoxycarbonylalkyl, C₃-C₆ cycloalkyl, C₃-C₆ halocycloalkyl, phenyl optionally substituted with 1 to 3 substituents independently selected from W, and benzyl optionally substituted with 1 to 3 substituents independently selected from W;

R³ is selected from the group R^a, R^b and J;

Ra is selected form the group H, C1-C4 alkyl, C3-C4 5 alkoxycarbonylalkyl, CO₂R¹⁵, C(O)R¹⁵, phenyl and phenyl substituted by (R¹⁴)_p; R^b is selected from the group CN, $C(R^{33})$ =NOR³⁴, C_2 - C_6 epoxyalkyl and SR²⁷; or R^b is C₁-C₄ alkyl substituted with a group selected from $C(O)R^{25}$, SR^{27} , $S(O)R^{27}$, SO_2R^{27} , C_1 - C_2 10 haloalkoxy and $Si(R^{28})(R^{29})R^{30}$; R4 is selected from the group H, C1-C3 alkyl, C2-C4 alkenyl and C2-C4 alkynyl; R^{18} is H; X is O; 15 X¹ is selected from the group Cl, OR⁶, SR⁶ and N(CH₃)₂; X² is selected from the group R⁶, OR⁶ and N(CH₃)₂; A is selected from the group O, CH2, OCH2, S, CH2CH2, SCH2, $N(R^5)CH_2$ and $N(R^5)$; $m R^{5}$ is selected from the group H, $m C_{1}\text{-}C_{4}$ alkyl, $m S(O)_{2}R^{15}$, $m CO_{2}R^{15}$ 20 and CONHR¹⁵: R15 is selected from the group C1-C4 alkyl, C1-C2 haloalkyl, C3-C4 alkenyl and propargyl; R¹⁶ is selected from H and CH3;

25 \mathbb{R}^{27} is \mathbb{C}_1 - \mathbb{C}_3 alkyl;

30

R33 is selected from the group H and CH3;

R³⁴ is selected from the group H and C₁-C₃ alkyl;

V is selected from the group O, $N(R^5)$ and CH_2 ;

W is selected from the group Cl, F, Br, CN, CF₃, C₁-C₂ alkyl, C₁-C₂ alkoxy, OCF₂H, OCF₃ and NO₂;

Y is selected from the group H, CH₃, C(O)CH₃ and CO₂CH₃;
Z is -CH₂CH₂- or -CH₂CH₂-;

m is 1 or 2;

n is 1 or 2; and

p is 1.

10

Preferred Compounds B are Compounds A wherein G is selected from the group G-2, G-3, G-7 and G-9.

Preferred Compounds C are Compounds B wherein J is selected from the group J-1, J-2, J-8, J-9 and J-16.

Preferred Compounds D are Compounds C of Formula I wherein Q is Q-1 and A is selected from the group CH_2 , OCH_2 , CH_2CH_2 , SCH_2 and $N(R^5)CH_2$.

20

30

Preferred Compounds E are Compounds C of Formula I wherein Q is Q-2 and A is selected from the group CH_2 , OCH_2 , CH_2CH_2 , SCH_2 and $N(R^5)CH_2$.

Preferred Compounds F are Compounds of Formula I wherein Q is Q-3 and A is selected from the group CH₂, OCH₂, CH₂CH₂, SCH₂ and N(R⁵)CH₂.

Preferred Compounds G are Compounds C of Formula I wherein Q is Q-4 and A is selected from the group CH_2 , OCH_2 , CH_2CH_2 , SCH_2 and $N(R^5)CH_2$.

In the above recitations, the term "alkyl", used either alone or in compound words such as "alkylthio" or "haloalkyl", denotes straight chain or branched alkyl, e.g., methyl, ethyl, n-propyl, isopropyl or the different butyl, pentyl, hexyl isomers.

10

15

Alkoxy denotes methoxy, ethoxy, n-propyloxy, isopropyloxy and the different butoxy, pentoxy or hexyloxy isomers.

Alkenyl denotes straight chain or branched alkenes, e.g., vinyl, 1-propenyl, 2-propenyl, 3-propenyl and the different butenyl, pentenyl and hexenyl isomers.

Alkynyl denotes straight chain or branched alkynes, e.g., ethynyl, 1-propynyl, 3-propynyl and the different butynyl, pentynyl and hexynyl isomers.

Alkylthio denotes methylthio, ethylthio and the different propylthio, butylthio, pentylthio and hexylthio isomers.

Alkylsulfinyl, alkylsulfonyl, alkylamino, etc., are defined analogously to the above examples.

Cycloalkyl denotes cyclopropyl, cyclobutyl, cyclopentyl and cyclohexyl.

The term "halogen", either alone or in compound words such as

"haloalkyl", denotes fluorine, chlorine, bromine or iodine. Further, when
used in compound words such as "haloalkyl" said alkyl may be partially
or fully substituted with halogen atoms, which may be the same or
different. Examples of haloalkyl include CH2CH2F, CF2CF3 and
CH2CHFCl. The terms "halocycloalkyl", "haloalkenyl" and "haloalkynyl"
are defined analogously to the term "haloalkyl".

The total number of carbon atoms in a substituent group is indicated by the "C_i-C_j" prefix where i and j are numbers from 1 to 8. For example, C₁-C₃ alkylsulfonyl designates methylsulfonyl through propylsulfonyl; C₂ alkoxyalkoxy designates OCH₂OCH₃; C₄

alkoxyalkoxy designates the various isomers of an alkoxy group substituted with a second alkoxy group containing a total of 4 carbon atoms, examples including OCH2OCH2CH2CH3 and OCH2CH2OCH2CH3; C2 cyanoalkyl designates CH2CN and C3 cyanoalkyl designates CH2CH2CN and CH(CN)CH3; C2 alkylcarbonyl would designates C(O)CH3 and C4 alkylcarbonyl includes C(O)CH2CH2CH3 and C(O)CH(CH3)2; and as a final example, C3 alkoxycarbonylalkyl designates CH2CO2CH3 and C4

5 alkoxycarbonylalkyl includes CH₂CH₂CO₂CH₃, CH₂CO₂CH₂CH₃ and CH(CH₃)CO₂CH₃.

DETAILS OF THE INVENTION

Compounds of Formula I where Q is Q-1

Preparation of certain pyrazoline compounds where Q is Q-1 and A is an alkylene bridge is reported in WO-88/07,994. Deprotonation of compounds of Formula I where R³ is equal to H, with base and reaction with suitable electrophilic reagents yields compounds of Formula I, where R³ is equal to E (Scheme 1).

15 Scheme 1

20

25

30

$$(R^{2})_{n}$$

$$N-N$$

$$O$$

$$Y$$

$$\frac{1. \text{ base}}{2. \text{ E}}$$

$$0$$

$$Y$$

$$I(Q-1)$$

$$R^{3} = H$$

$$I(Q-1), R^{3} = E$$

The electrophilic reagent can be chosen to give the preferred R³ groups of this invention. For example, p-toluenesulfonyl cyanide can be used to prepare compounds of Formula I where R³ is equal to cyano; see Collum et al. (<u>Tetrahedron Lett 1981, 22, 5011</u>). Methyl chloroformate can be used to prepare compounds of Formula I where R³ is equal to carbomethoxy. Dimethylformamide can be used to prepare compounds of Formula I where R³ is formyl (HCO). Acid chlorides can be used to prepare compounds of Formula I where R³ is acyl (C(O)alkyl).

The formyl and acyl derivatives can be further reacted with amine derivatives such as hydroxy- or alkoxy-amines to yield compounds of this invention as depicted in Scheme 2 (reaction a). The methyl ester derivative can be further reacted by hydrolysis to the acid, conversion of

the acid to an acid chloride, reaction of the acid chloride with trimethylsilylazide, Curtius rearrangement of the acyl azide to an isocyanate, and finally reaction of the isocyanate with methanol to yield the methyl carbamate (Scheme 2, reaction b). The intermediate acid chloride can alternatively be reacted to provide various other ester derivatives (Scheme 2, reaction c).

Scheme 2

a)
$$(R^2)_n$$
 $N-N$
 $N-G$
 $N-G$

 R^{33} is H or $C_1^-C_4$ alkyl

 R^{33} and R^{34} are independently H or C₁-C₄ alkyl

b)
$$(R^{2})_{n}$$

$$N-N$$

$$N-G$$

$$1. NaOH, MeOH, (R^{2})_{n}$$

$$2.SOCl_{2}$$

$$3. TMSN_{3}, PhCH_{3}$$

$$4. MeOH$$

$$I(Q-1)$$

$$R^{3}=CO_{2}Me$$

$$1. NaOH, MeOH, (R^{2})_{n}$$

$$2.SOCl_{2}$$

$$3. TMSN_{3}, PhCH_{3}$$

$$4. MeOH$$

$$I(Q-1)$$

$$R^{3}=NHCO_{2}Me$$

10

 $R^3 = CO_2Me$

10

15

20

$$\begin{array}{c} \text{C } \\ \text{C } \\$$

Methods for performing the ester-to-methyl carbamate transformation are well known in the art; see Washburne et al. (Syn. Commun. 1972, 2, 227 and J. Org. Chem. 1973, 38, 2982). Related electrophiles for preparation of other preferred compounds are known to those skilled in the art. The procedures of Scheme 2 (reactions a and c) and modifications thereof are known in the art; see, for example, J. Org. Chem., 1971, 36, 284 and WO 89/00991.

Compounds of Formula I where Q is Q-1 and G is G-9 can be prepared by the reaction of aryl isocyanates of Formula III and tetrahydropyrazoles of Formula IV as shown in Scheme 3. Typical reactions involve the combination of equimolar amounts of III and IV in conventional organic solvents such as ether, tetrahydrofuran, dichloromethane, and toluene. The reaction can be run at temperatures between -20°C and 80°C, with temperatures in the range of about 10°C to 30°C generally preferred.

Scheme 3

Alternative methods for preparing compounds of Formula I where
Q is Q-1 and G is G-1 through G-10 are summarized in Schemes 4 and 5.

Scheme 4

$$(R^{2})_{n}$$

$$IV$$

$$1. X^{4}CX^{3} (XXX)$$

$$base$$

$$2. GNHY$$

$$(XXXI)$$

$$(R^{2})_{n}$$

$$I$$

$$(Q=Q-1)$$

15 wherein:

5

 X^3 and X^4 are groups such as halogen, imidazole, toluenesulfonate, phenoxide or alkoxide; A, G, R^2 , R^3 , X and Y are as previously defined.

WO 92/03421 23 PCT/US91/05334

5 SCHEME 5

$$\frac{\text{Step (i)}}{\text{IV}} \qquad \frac{\text{XXX}}{\text{base}} \qquad \frac{\text{XXXII}}{\text{XXXII}}$$

wherein:

10

15

20

25

A, \mathbb{R}^2 , \mathbb{R}^3 , X and \mathbb{X}^4 are as previously defined.

The reaction described in Scheme 4 is typically carried out in an inert solvent such as, but not limited to, dichloromethane, chloroform, or tetrahydrofuran. Approximately 200 to 500 mol percent of an organic base such as pyridine or triethylamine or an inorganic base such as potassium carbonate is mixed with the solution of equimolar amounts of IV and XXX in the solvent at a temperature between about 0 to 30°C for a period of about one to twelve hours. An equimolar amount of XXXI is then added to the reaction mixture at a temperature of about 0 to 60°C and the resulting mixture is stirred at that temperature for a period of about 2 and 20 hours. The product, I, can be isolated by removal of the solvent and silica gel chromatography of the residue. Suitable solvent mixtures for silica gel chromatography include, but are not limited to, ether-hexanes, ethyl acetate-hexanes and methylene chloride-ethanol-ammonium hydroxide.

The reaction described in Step (i) of Scheme 5 is typically carried out in an inert solvent such as dichloromethane, chloroform or tetrahydrofuran. Approximately 0 to 200 mol percent of a base such as, but not limited to, triethylamine or pyridine is mixed with the solution of equimolar amounts of IV and XXX in the solvent at a temperature

between about 0 to 30°C for a period of about one to twelve hours.

Compound XXXII is isolated from the reaction mixture in crude form by removing the solvent, triturating the residue with a solvent such as ether, filtration and solvent removal of the filtrate.

The reaction described in Step (ii) of Scheme 5 is typically carried out in an inert solvent such as dichloromethane, chloroform or tetrahydrofuran. Approximately 100 to 300 mol percent of a base such as, but not limited to, triethylamine or pyridine is mixed with the solution of approximately equimolar amounts of XXXI and XXXII at a temperature between 0 to 60°C for a period of about 2 to 12 hours. The product, I, can be isolated by removal of the solvent and silica gel chromatography of the residue. Suitable solvent mixtures for silica gel chromatography include, but are not limited to, ether hexanes, ethyl acetate-hexanes and methylene chloride-ethanol-ammonium hydroxide.

Alternatively, 50 mol percent of trichloromethyl chloroformate (diphosgene) or 33 mol percent of bis-(trichloromethyl) carbonate (triphosgene) can be used in place of XXX using procedures analogous to those described in Schemes 4 and 5.

Alternatively, salts of XXXI (such as hydrohalides and the like) can be used in place of XXXI using procedures analogous to those described for Schemes 4 and 5.

Compounds of Formula IV can be prepared by acid hydrolysis of their phosphorylated derivatives of Formula V, as shown in Scheme 6.

SCHEME 6

30

10

15

20

25

V

The reaction of Scheme 6 is generally carried out by contacting the substrate with two to three equivalents of a strong acid such as HCl or ptoluene sulfonic acid, optionally containing some water, in a solvent such as ethanol or dioxane for 1 to 72 hours at 50°C to 100°C. The initial product of the reaction is a salt of IV and can either be neutralized and extracted into an organic solvent for use in the reaction of Schemes 3, 4 and 5 or used directly in the presence of an additional two equivalents of a suitable organic or inorganic base.

Alternative methods for preparing Formula IV compounds can be found in U.S. 4,960,784, which methods are incorporated herein by reference.

The compounds of Formula V are produced by the cyclization reaction of an olefinic phosphorylhydrazone of Formula VI induced by the combination of mild base such as triethylamine and a mild halogenating agent such as N-bromosuccinimide, as shown in Scheme 7.

20

15

5

10

SCHEME 7

$$(R^2)_n$$
 N
 N
 N
 Et_3N
 V
 $O=P(OEt)$
 V
 V

In the reaction of Scheme 7, the hydrazone of Formula VI is typically dissolved in an inert, anhydrous solvent in the presence of one of the base/haloimide components at -20°C to 50°C and the other component is added gradually. The product can be isolated by washing the reaction mixture with water, evaporating the solvent, and either recrystallizing or chromatographing to remove by-products.

The synthesis of compounds of Formula VI can be accomplished by condensing an aldehyde of Formula VII with diethyl phosphorohydrazidate VIII, as shown in Scheme 8.

SCHEME 8

10

15

20

25

$$(R^{2})_{n} \xrightarrow{A \qquad R^{3}} + H_{2}NNHP(OEt)_{2} \xrightarrow{-H_{2}O} VI$$

$$VII \qquad VIII$$

The reaction of Scheme 8 can be conducted by contacting aldehyde VII with slightly more than one equivalent of VIII in an inert solvent such as dichloromethane for about 0.1 to 16 hours, optionally in the presence of a catalytic (0.1 to 10%) amount of an acid such as acetic or ptoluenesulfonic, and optionally in the presence of an inorganic drying agent such as anhydrous magnesium sulfate. The product can be isolated by washing out excess VIII with water and concentrating the dried solution. It can be purified, if desired, by recrystallization or chromatography.

Compounds of Formula VII can be prepared by reacting an aniline, thiophenol or phenol of Formula IX with alkylating agents of Formula X, wherein L is a leaving-group, such as Cl, Br, I, OSO₂CH₃, or OSO₂Ar, as shown in Scheme 9.

SCHEME 9

10

15

20

25

30

The reaction of Scheme 9 is typically carried out by combining equimolar quantities of IX and X in a suitable solvent for alkylation reactions, such as dimethylformamide (DMF) in the presence of slightly more than one equivalent of a suitable base, such as anhydrous potassium carbonate or sodium hydride and stirring vigorously at a temperature between 0°C and 100°C. The product is generally isolated by diluting the mixture with water and filtering the precipitated product or extracting into a solvent such as ether. Recrystallization or chromatography affords the pure product.

Compounds of Formula IX will be recognized by those skilled in the art as being obtainable from substituted anilines, thiophenols or phenols by well-known methods. Compounds of Formula X are either known compounds or are readily available from olefins or allylic alcohols by know methods.

Compounds of Formula I where the preferred R^3 is a relatively stable group, for example substituted heterocycles, can be prepared from a compound of Formula X containing the desired R^3 group. Compounds of Formula I where the preferred R^3 is a reactive group, for example epoxide, must be prepared from a compound of Formula X containing a R^3 group that can be transformed into the desired R^3 group at the end of the synthesis. This strategy is depicted in Scheme 10.

SCHEME 10

$$(R^2) \xrightarrow{n} \xrightarrow{AH} \xrightarrow{L} (R^2) \xrightarrow{n} \xrightarrow{CHO} \xrightarrow{CHO} \xrightarrow{CHO} \xrightarrow{CHO} \xrightarrow{CHO} \xrightarrow{CHO} \xrightarrow{X} (R^2) \xrightarrow{n} \xrightarrow{CHO} \xrightarrow{CHO} \xrightarrow{X} (R^2) \xrightarrow{n} \xrightarrow{CHO} \xrightarrow{CHO} \xrightarrow{R^3 = CH = CH_2} \xrightarrow{R^3 = CH = CH_2} \xrightarrow{R^3 = CH = CH_2} \xrightarrow{A = OCH_2, N(R^5)CH_2, SCH_2}$$

WO 92/03421 28 PCT/US91/05334

$$(R^{2})_{n}$$

$$N^{-N}$$

$$N^{-N$$

Compounds of Formula I where Q is Q-2

5

10

15

20

25

Compounds of Formula I, where Q is Q-2, can be obtained by the reaction of activated carbonyl or thiocarbonyl compounds of Formula XI with Formula XXXI compounds in the presence or absence of an acid acceptor or suitable condensing agent. Methods for performing this transformation are well known in the art; see, Zabicky, "The Chemistry of the Amides", Interscience, 1970.

One particularly useful method involves the chlorination of an acid derivative (XI, L=OH) with thionyl chloride or another chlorinating agent followed by treatment with XXXI in the presence of an acid acceptor such as an amine base, preferably triethylamine (Scheme 11). Suitable solvents for the chlorination reaction are inert to hydrogen chloride and inleude benzene, toluene, and dichloromethane. Preferred temperatures for this process are from 20° to 100°C with temperatures between 20° and 80°C being particularly preferred. The latter reaction can be carried out in many different inert solvents such as dialkylethers, chlorinated hydrocarbons, and aromatic hydrocarbons. These reactions are normally run at atmospheric pressure, but can also be carried out at elevated pressures.

10

15

20

25

30

SCHEME 11

L=Cl, Br, OH, OR

Esters of Formula XI (L=C₁-C₆ alkoxy) can be converted directly to compounds of Formula I (Q=Q-2) in several ways. In the presence of Lewis acids such as AlMe₃, anilines react readily with esters of Formula XI. The reaction is best carried out at room temperature to 120°C. Suitable solvents include dichloromethane, 1,2-dichloroethane, and toluene. The method described by Weinreb et al., Organic Synthesis, 59, 49, (1982), proceeds best with esters of lower alcohols such as methanol or ethanol.

Acids of Formula XI (L=OH) can be converted directly to compounds of Formula I by use of coupling agents known in the peptide art in conjuction with anilines. Coupling agents include dicyclohexylcarbodiimide (DCC), N-hydroxysuccinimide, 2-chloro-N-methylpyridinium iodide, carbonyl diimidazole, or other agents capable of activating an acid function or acting as a dehydrating agent. These and other methods are described in Gross et al., "The Peptides," 3 Vols., Academic Press, New York, 1979 to 1981.

Compounds of Formula I (Q=Q-2) and intermediates of Formula XI can be obtained by the intramolecular dipolar cycloaddition reaction of nitrile-imines, generated from substituted phenylhydrazones of Formula XIII (Scheme 12). The presence of an acid acceptor (generally an amine base, for example, triethylamine) is necessary for the formation of the nitrile-imine. Suitable solvents include but are not restricted to benzene, toluene, 1,2-dichloroethane, chloroform, and tetrahydrofuran. The reaction can be carried out at temperatures ranging from 20° to 120°C

with the relative reactivity of the alkene moiety governing the required temperature for a given example.

SCHEME 12

$$(R^{2})_{n} \xrightarrow{A \qquad R^{3}} \xrightarrow{acid} \qquad XI \text{ or } I(Q-2)$$

$$\downarrow N \qquad \qquad \downarrow N \qquad \qquad \downarrow 1$$

$$\downarrow L^{3} \qquad \qquad \downarrow L^{2}$$

$$XIII \qquad \qquad \downarrow L^{2} = halogen$$

$$\downarrow L^{3} = OR, \quad -N - G$$

$$\mid \qquad \qquad \downarrow$$

$$Y$$

10

15

20

The required hydrazones of Formula XIII can be synthesized by the Japp-Klingemann reaction (Scheme 13). The coupling of diazonium salts with active methylene compounds is known. The more specific coupling of chloroacetoacetic acid derivatives of Formula XIV with diazotized anilines of Formula XV containing alkenyl substituents is described by Padwa et al. in J. Org. Chem., 43, 1664 (1978) and J. Org. Chem., 46, 1402 (1981). A similar process for this type of aniline is described by Garanti et al. in J. Org. Chem. 42, 1389 (1977), and J. Chem. Soc. Perkin I, 2245 (1981).

SCHEME 13

$$(R^{2})_{n} \xrightarrow{A} A R^{3} + Cl COMe NaNO_{2} \xrightarrow{or} i-amONO XIII$$

$$L^{3}O$$

$$XV XIV$$

$$L^{3}=OR, -N-G$$

$$Y$$

The anilines of Formula XV can be obtained by the reduction of
aromatic nitro compounds of Formula XVI (Scheme 14). There are many
methods known for this transformation. See, March, "Advanced Organic
Chemistry", 1985, Wiley, page 1103-1104. A particularly suitable method
involves the treatment of the nitro compound with tin (II) chloride in
alcoholic solvents. Refer to Bellamy et al. <u>Tetrahedron Letters</u>, 1984,
839.

WO 92/03421 PCT/US91/05334

5 SCHEME 14

25

32

Nitro compounds of Formula XVI containing a heteroatom in the
alkenyl chain can be obtained by alkylation reactions (Scheme 15).
Treatment of a substituted phenol, thiophenol, or aniline of Formula
XVII with an acid acceptor and an allyl or homoallyl halide or sulfonate
of formula XVIII gives compounds of Formula XVI as products. Preferred
acid acceptors for the process are inorganic bases such as potassium
carbonate. Preferred solvents include dimethylformamide,
dimethylsulfoxide, methylethyl ketone, and ethanol. The reaction is
generally carried out at room temperature, but higher temperatures may
be necessary in some cases. Alternative methods for these nitro
compounds of Formula XVI have been described by Oae et al. in Bull.

Chem. Jap., 54, 2374 (1981).

SCHEME 15

$$(R^{2})_{n}$$

$$XVII$$

$$A=0$$

$$NR^{5}$$

$$L=Br, Cl,$$

$$OSO_{2}Me, etc.$$

$$XVII$$

$$SCH_{2}$$

$$SCH_{2}$$

10

15

25

30

Compounds of Formula XVII and compounds of Formula XVIII having the preferred R³ groups of this invention are either known compounds or can be prepared by conventional methods.

Compounds of Formula I where Q is Q-3

The compounds of Formula I, where Q is Q-3 and G is G-9, can be prepared by the reaction of hydrazones of Formula XIX with an aryl isocyanate of Formula III (Scheme 16). Typical reactions involve combination of equimolar amounts of XIX and III in a suitable solvent at temperatures generally in the range of -10 to 100°C. Although the reaction can be run neat, a solvent is generally preferred. Suitable solvents typically have sufficient polarity to effect solution of the Formula XIX hydrazone and include, but are not limited to, ethers such as diethyl ether, tetrahydrofuran and dioxane; halogenated hydrocarbons such as methylene chloride, chloroform and carbon tetrachloride;

20 <u>SCHEME 16</u>

$$(R^{2})_{n} \xrightarrow{A} R^{3} + R^{4}$$

$$R^{1} \xrightarrow{R^{2}} R^{2}$$

$$(G=G-9)$$

$$R^{1} \xrightarrow{R^{2}} R^{2}$$

$$R^{2} \xrightarrow{R^{2}} R^{2}$$

aromatic hydrocarbons such as benzene, toluene and xylene; esters such as ethyl acetate and polar aprotic solvents such as dimethylformamide and dimethylacetamide.

The hydrazones of Formula XIX can be obtained by processes known in the art involving condensation of a ketone of Formula XX with either hydrazine or a substituted hydrazine of Formula XXI (Scheme 17).

This reaction is typically conducted with equimolar amounts of XX and XXI although greater than stoichiometric amounts of hydrazine can be used. Suitable solvents include the alcohols such as methanol, ethanol, propanol, butanol and the like at temperatures in the range of 0 to 150°C, with the reflux temperature of the solvent generally being a convenient reaction temperature. Acid catalysis can also be useful, particularly for some of the more sterically hindered Formula XX compounds. Typical acid catalysts include sulfuric, hydrochloric and p-toluenesulfonic acid.

SCHEME 17

$$(R^2)_n \xrightarrow{A} R^3 + Y^1NHNH_2 \longrightarrow XIX$$

$$XX \qquad XXI$$

An alternate process useful for the preparation of compounds of Formula I involves condensation of a semicarbazide of Formula XXII with a ketone of Formula XX (Scheme 18). Preferred conditions for this reaction include an acid catalyst such as hydrochloric, sulfuric or ptoluene sulfonic acid. Reaction temperatures can range from 0 to 150°C with the reflux temperature of the solvent used generally preferred. Suitable solvents include, but are not limited to, ethers such as tetrahydrofuran and dioxane; aromatic hydrocarbons such as benzene and toluene; and especially preferred are alcohols such as methanol, ethanol and isopropanol.

WO 92/03421 PCT/US91/05334

SCHEME 18

5

10

15

20

25

30

XXII

Semicarbazides of Formula XXII can be prepared using procedures described by Scheme 19. The first step (Step (i) of Scheme 19) involves the reaction of equimolar amounts of an amine XXXI (Y is H) with XXX in a suitable solvent at a temperature between about 0 to 100°C for a period of about 1 to 36 hours. In some cases it is desirable to use 0 to 200 mol percent of a base such as, but not limited to, triethylamine or pyridine. Suitable solvents include, but are not limited to, dichloromethane, chloroform or tetrahydrofuran. The product, XXXIII, is typically isolated by evaporation of the solvents and used directly in step ii of Scheme 19.

In step ii of Scheme 19, XXXIII is mixed with 1 to 5 equivalents of a hydrazine XXI in the presence of 0 to 3 equivalents of a base in a suitable solvent at a temperature ranging from 0°C to the boiling point of the solvent. Suitable bases include both inorganic bases as well as organic amines such as pyridine and triethylamine. Suitable solvents include, but are not limited to, alcohols such as methanol, ethanol and isopropanol, acetonitrile, chloroform and tetrahydrofuran. Reactions of step ii of Scheme 19 are typically run for a period of 6 to 72 hours, and the product XXII is usually isolated by chromatography on silica gel using a mixture of methylene chloride, ethanol and aqueous ammonium hydroxide as the eluant.

5 SCHEME 19

10

15

20

25

Alternatively, semicarbazides of Formula XXII can be prepared by reactions of hydrazines XXI with isocyanates III using procedures that are well-known to those skilled in the art.

Compounds of Formula I where Y and Y¹ are other than hydrogen can generally be prepared from the corresponding compounds where Y and Y¹ are hydrogen by reaction with electrophilic reagents such as alkyl halides, acyl halides, alkyl chloroformates and sulfenyl halides. The use of a base is generally preferred in these reactions but is dependent upon the specific nature of the reactants. For example, when the electrophilic reagent is selected from an alkyl halide, acyl halide or alkyl chloroformate, then metal hydrides such as sodium hydride or potassium hydride in solvents such as tetrahydrofuran or dimethylformamide are preferred. When sulfenyl halides are used then mine bases such as triethylamine in solvents such as diethyl ether or tetrahydrofuran are generally preferred. Of course, many of the compounds where Y¹ is other than H can also be prepared by use of the appropriate hydrazine XXI in Scheme 17. For example, methyl hydrazine and methyl carbazate will produce compounds where Y¹ is methyl and carbomethoxy, respectively.

25

The starting ketones of Formula XX are known or can be obtained 5 by processes analogous to known ones. Those skilled in the art will recognize the Formula XX compounds to include substituted indanones, tetralones, chromanones, thiochromanones, benzofuran-3-ones, thiobenzofuran-3-ones, isochromanones, isothiochromanones and others. 10

Compounds of Formula I where Q is Q-4

Compounds of Formula I where Q is Q-4 can be prepared by the methods reported in WO-88/900,910. Scheme 20 depicts a particularly useful method discussed in WO 88/900,910.

SCHEME 20 15

XXXV

Hydrazones of Formula XXXV can be prepared analogously to Formula XIII compounds as described in Scheme 13. Alkene XXIII can be prepared with R3 being equal to one of the preferred groups of this invention. Compounds of Formula XXIII are either known compounds or readily available by methods known to those skilled in the art. Compounds of Formula I where Q is Q-5 through Q-9

Compounds of Formula I (Q-5) wherein R^{18} is equal to H and G is G-9 can be prepared by the reaction of aryl isocyanates of Formula ${
m III}$ with substituted pyrazolidines of Formula XXIV as illustrated in Scheme 21. Typical reactions involve the combination of equimolar amounts of ${
m III}$ and XXIV in conventional organic solvents including ethyl acetate, ether,

tetrahydrofuran, methylene chloride, chloroform, benzene and toluene but not restricted to these. The reaction can be conducted at temperatures ranging from -20°C to 100°C with temperatures in the range of -10°C to 30°C generally preferred.

15

20

25

30

$$R^{1}$$
 $(R^{2})_{n}$
 $(R^{2})_{n}$
 R^{18}
 H

III

XXIV

Alternatively, compounds of Formula I (Q-5) where R¹⁸ is H and G is G-1 through G-10 can be prepared from compounds of Formula XXIV using procedures analogous to those already described for the syntheses of compounds of Formula I (Q-1) in Schemes 4 and 5 and, for the sake of brevity, will not be described further.

Compound of Formula XXIV can be prepared by the reaction of hydrazine dihydrochloride or hydrazine sulfate with compounds of Formula XXV as illustrated in Scheme 22. Modification of the procedure described in Bull. Chem. Soc. Jap., 1982, 55, 2450, and J. Org. Chem., 1987, 52, 2277, can be applied to the synthesis of compounds of the Formula XXIV. Typical reactions involve the addition of an excess of equivalents of hydrazine dihydrochloride or hydrazine sulfate, ranging from 1.1 to 20 equivalents with 5 to 15 equivalents being preferred, to one equivalent of a compound of Formula XXV in an alcohol/water solvent mixture ranging from 1 to 99% alcohol with 90% to 95% alcohol being preferred. Typical alcoholic solvents include methanol, ethanol, n-propanol, n-butanol, tert-butanol and the like. The reaction can be conducted at temperatures ranging from -20°C to 140°C with

15

20

25

temperatures in the range of 20°C to 80°C being preferred. The reaction is usually complete within 24 hours.

SCHEME 22

Substituted benzaldehydes of Formula XXV can be prepared by known methods or obvious modifications thereof by one skilled in the art.

Compounds of Formula I (Q-5) where R¹⁸ is not equal to H can be prepared by the reaction of Formula I (Q-5) where R¹⁸ is equal to H with a variety of electrophiles. For example, these electrophiles include, but are not limited to, alkyl halides, alkyl and aryl isocyanates, acyl halides, sulfonyl halides and alkyl chlorocarbonates. Reactions to prepare Formula I (Q-5) compounds where R¹⁸ is not equal to H can be conducted through standard procedures known to those skilled in the art. For example, the reaction of Formula I (Q-5) compounds where R¹⁸ is equal to H with methyl isocyanate (Scheme 23) can be conducted by reaction of equal molar amounts of the reactants in an inert solvent such as, but not limited to, ether, tetrahydrofuran, dimethoxyethane, ethyl acetate, methylene chloride and chloroform, in the presence of an acid scavenger such as tertiary alkylamines, substituted pyridines, alkali metals, and the like.

10

15

20

25

5 SCHEME 23

$$(R^2)_n$$
 R^3
 $MeNCO$
 R^3
 $MeNCO$
 $N-N$
 $N-N$

Compounds of Formula I (Q-6) can be prepared by the reaction of tri- and tetravalent metal species such as titanium, silicon, tin and the like in combination with a reducing agent such as sodium, lithium, or zinc borohydride, lithium aluminum hydride and the like with compounds of Formula I where Q is equal to Q-2 as illustrated in Scheme 24. Literature precedent for analogous reactions can be found in <u>J. Org. Chem., 1987, 54, 3750</u>, and <u>Synthesis, 1980</u>, 695. Typical reactions involve the addition of 1 equivalent of a compound of Formula I (Q-2) to a solution of 1.1 to 4 equivalents of titanium tetrachloride, with 1.5 to 2.5 equivalents being preferred, and 2.1 to 6 equivalents of sodium borohydride with 3.5-4.5 equivalents being preferred.

Conventional organic solvents such as ether, tetrahydrofuran, dimethoxyethane, methylene chloride and chloroform can be used with 1,2-dimethoxyethane being preferred. The reaction can be conducted at temperatures ranging from -70°C to 50°C with -10°C to 30°C being preferred. The reaction time can be 0.1 hour to 48 hours with 2 to 4 hours being preferred.

15

20

SCHEME 24

$$(R^{2})_{n} \xrightarrow{V} R^{3} \xrightarrow{\text{Ti Cl}_{4}} (R^{2})_{n} \xrightarrow{V} R^{3}$$

$$N = N - G$$

$$Y$$

$$I (Q-2)$$

$$A = VCH_{2}$$

Formula I (Q-6) compounds where R^{18} is not equal to H can be prepared in the same fashion as described for Formula I (Q-5) compounds where R^{18} is not equal to H.

Compounds of Formula I (Q-7) can be prepared by a titanium tetrachloride/sodium borohydride reduction of Formula I (Q-3) analogs in a similar fashion as described for Formula I (Q-6) compounds. Scheme 25 illustrates the methodology utilized.

SCHEME 25

$$(R^{2})_{n} \xrightarrow{A} R^{3} \qquad TiCl_{4} \qquad (R^{2})_{n} \xrightarrow{A} R^{3} \qquad R^{4} \qquad HN$$

$$NH - C \qquad N-G \qquad N-G \qquad V$$

$$I(Q-3) \qquad I(Q-7) \qquad R^{18} = H$$

The compounds represented by Formula I (Q-8) can be prepared by a titanium tetrachloride/sodium borohydride reduction of Formula XXVI analogs in a similar fashion as described for Formula I (Q-6) compounds. Scheme 26 illustrates this methodology.

10

15

SCHEME 26

$$(R^{2})_{n}$$

$$R^{3}$$

$$N-N$$

$$N-G$$

$$Y$$

$$XXVI$$

$$(R^{2})_{n}$$

$$R^{3}$$

$$R^{4}$$

$$N-N$$

$$N-N$$

$$Y$$

$$Y$$

$$Y$$

$$XXVI$$

$$R^{1}(Q-8)$$

$$R^{18} = H$$

The preparation of the compounds represented by Formula XXVI is described in U.S. 4,070,365, and 4,633,341.

For example, the 3-phenyl-dihydropyrazoline intermediate is prepared using conventional procedures and further reacted with an appropriately substituted phenyl isocyanate.

The compounds represented by Formula I (Q-9) can be prepared by a titanium tetrachloride/sodium borohydride reduction of Formula I (Q-4) analogs in a similar fashion as described for Formula I (Q-6) compounds.

Scheme 27 illustrates this methodology.

SCHEME 27

$$(R^{2})_{n}$$

$$R^{3}$$

$$R^{4}$$

$$N = R^{4}$$

Compounds of Formula I where Q is Q-10

The preparation of compounds of Formula I where Q is Q-10 is similar to that for compounds where Q is equal to Q-1. Cyclic alkylating agents of Formula XXVII are used in place of standard allylic halides for the alkylation of phenols or anilines as depicted in Scheme 28 (see Scheme 9 for reference).

15

10

SCHEME 28

XXVII IX L=halo, OSO_2Me , etc. V=O, NR^5

XXVIII

5 Compounds of Formula XXVIII are carried through the identical reaction sequence in Schemes 6 through 8 to give the desired tetracyclic pyrazoles XXXIV (Scheme 29).

10

15

20

25

30

(R²) n V Z
$$\frac{1}{2}$$
 1) H₂NNHP(OEt)₂ (R²) n V Z $\frac{1}{2}$ NES, Et₃N $\frac{1}{2}$ NeOH, HCl 4) NaHCO₃ H XXXIV

The tetracyclic pyrazoles XXXIV can then be converted to compounds of Formula I where Q is Q-10 and G is G-9 by treatment with acyl isocyanates using procedures completely analogous to those described in Scheme 3.

Compounds of Formula I where Q is Q-10 and G is G-1 through G-10 can be prepared from Formula XXXIV using procedures analogous to those described for Schemes 4 and 5.

Compounds of Formula I where Q is Q-11

Compounds of Formula I where Q is Q-11 can be prepared using the same procedure described for compounds where Q is equal to Q-3. The tricyclic ketone starting material can be prepared according to literature procedures (see Marchant, <u>J. Chem. Soc. 1957</u>, 3325). Compounds of Formula II where Q is Q-1 through Q-11

Compounds of Formula II can be prepared by the reaction of a Formula I compound with an appropriate halogenating agent such as phosphorous trichloride, phosphorous pentachloride, phosphorous tribromide, phosphorous pentabromide, thionyl chloride, sulfuryl chloride, triphenyl phosphine and carbon tetrachloride (Wolkoff, Can. J.

Chem., 1975, 53, 1333) and the like (see Fieser and Fieser, Reagents for Organic Synthesis, Vol. I, 1967) as illustrated in Scheme 30. Typical reactions involve the combination of Formula I compounds with an excess of the halogenating agent ranging from 1.1 to 10 equivalents, with 2 to 4 equivalents being preferred. The reaction can be conducted in the absence of a solvent or in the presence of a conventional organic solvent such as benzene, toluene, xylene, chloroform, methylene chloride, hexane and the like. The reaction temperature can range from -10°C to 200°C with 35°C to 100°C being preferred. The reaction is generally complete after 24 hours.

15

20

25

10

5

SCHEME 30

$$\begin{array}{c|c} O & & & X^1 \\ & & & \\ || & & \\ Q\text{-C-NH-G} & & & \\ I & & & \\ I & & & \\ X^1\text{=Cl, Br} \end{array}$$

Alternatively, compounds of Formula II when X¹ is equal to R⁶-S, can be prepared by the reaction of compounds of the Formula I with an electrophile of the Formula XXIX in the presence of a suitable base, as illustrated in Scheme 31. Typical reactions involve the combination of equimolar amounts of Formula I compounds and the appropriate electrophile of Formula XXIX. A base such as an alkali metal, tertiary amine or metal hydride can be used.

5 SCHEME 31

$$\begin{array}{c} S \\ \parallel \\ Q\text{-C-NH-G} \\ \end{array} + R^6L \xrightarrow[K_2\text{CO}_3]{N_8H} \qquad SR^6 \\ \downarrow \\ Q\text{-C-N-G} \\ \\ I \qquad XXIX \\ X=S \qquad L=\text{halo, OSO}_2\text{Me, etc.} \\ Y=H \end{array}$$

The following Examples further illustrate the invention.

10

15

20

25

30

EXAMPLE 1

Step A: ((2-(3-chlorophenyl)ethyl)methanesulfonate

To a 0°C solution of 30.0 g of 3-chlorophenethyl alcohol and 15.3 mL of methane sulfonyl chloride in 150 mL of THF was added, dropwise, a solution of 28.0 mL of triethylamine in 50 mL of THF. The reaction was warmed to room temperature, stirred overnight, and then filtered. The filtrate was partitioned between aqueous sodium bicarbonate and ether. The organic extracts were then dried over magnesium sulfate, filtered and concentrated under reduced pressure to afford 45.93 g of a clear, colorless oil. ¹H NMR was consistent with the structure for the named compound.

Step B: 3-chlorobenzenebutanoic acid

To a mixture of 8.0 g of 60% sodium hydride in 300 mL of THF, under N₂, was added dropwise a solution of 31.0 mL of diethyl malonate in 50 mL of THF. Upon complete addition of the diethyl malonate, a pale yellow homogeneous solution was obtained. To this was added a solution of 45.93 g of the sulfonate from Step A and the mixture was then heated at reflux overnight. The reaction was then cooled to room temperature, poured into 400 mL of 1N HCl, and extracted with ether. The ether extracts were dried over magnesium sulfate, filtered and concentrated under reduced pressure to afford 68.9 g of a yellow oil. The crude oil was dissolved in 400 mL of methanol, 100 mL of H₂O and 40 mL of 50%

.

aqueous NaOH. The reaction was stirred overnight and the methanol was then removed at reduced pressure. The crude residue was partitioned between H₂O and ether, the aqueous extracts were acidified with concentrated HCl and then extracted several times with ether. The ether extracts were dried over magnesium sulfate, filtered and concentrated to afford 51.7 g of a yellow oil. The crude residue was dissolved in 200 mL of toluene and heated at reflux for 4 days under N₂ to effect decarboxylation. After this time, toluene was removed by concentration at reduced pressure to afford 35.72 g of a yellow oil. ¹H NMR analysis of the crude product was consistent with 3-chlorobenzenebutanoic acid of purity estimated to be 80%. The crude product was used without further purification directly in the next step.

Step C: 6-chloro-3.4-dihydro-1(2H)-naphthalenone

20

25

30

35

A mixture of 35.72 g of the product from Step B and 50 mL of thionyl chloride was heated at reflux for 2 hours and then stirred at room temperature for 18 h. After this time, thionyl chloride was removed at reduced pressure and the product was dissolved in carbon tetrachloride and concentrated at reduced pressure. The residue was dissolved in 150 mL of dichloroethane cooled to 0°C and 28 g of aluminum chloride was added portionwise over about 1 hr in approximately 3 g portions. After stirring for 3 h, the reaction was poured over a mixture of ice/1N HCl and extracted three times with methylene chloride. The organic extracts were dried over magnesium sulfate and concentrated to approximately 30 g of a brown oil. Chromatography on silica gel with 10% ethyl acetate/hexane afforded 17.83 g of 6-chloro-3,4-dihydro-1(2H)-naphthalenone as a brown oil. ¹H NMR was consistent with the structure.

Step D: 7-chloro-3.3a.4.5-tetrahydro-N-[4-(trifluoromethyl)-phenyl]-2H-benz[g]-indazole-2-carboxamide

A mixture of 3.4 g (0.03 mole) of 6-chloro-3,4-dihydro-1(2H)-naphthalenone (Step C), 2.5 g of dimethylamine hydrochloride, 1.0 g of paraformaldehyde, 0.7 mL of concentrated HCl and 15 mL of ethanol was combined and heated at reflux for 18 h. The reaction was then

10

15

20

25

30

35

concentrated at reduced pressure and partitioned between H2O and ether. The aqueous extracts were made basic with 1N NaOH and then extracted three times with ether. The ether extracts were dried over magnesium sulfate and concentrated to 4.64 g of a yellow oil. This compound was dissolved in 25 mL of ethanol and 1.5 mL of hydrazine hydrate was added followed by 5 to 6 drops of 50% sodium hydroxide. The reaction was then heated at reflux, under N2, for 2 to 3 h after which time it was cooled and most of the ethanol was removed by concentration at reduced pressure. The crude residue was partitioned between saturated aqueous sodium bicarbonate and methylene chloride. The methylene chloride extracts were dried over magnesium sulfate and filtered. The methylene chloride extracts were then combined with 3.5 g of 4-trifluoromethylphenyl isocyanate and stirred under N2 overnight. The reaction was then concentrated and the crude residue triturated with ether to afford 3.35 g of the product as a white powder, mp 196-199°C. ¹H NMR (CDCl₃) δ 1.9 (m, 1H), 2.2 (m, 1H), 3.0 (m, 2H), 3.5 (m, 2H), 4.43 (m, 1H), 7.24 (m, 2H), 7.55 (d, 2H), 7.67 (d, 2H), 7.92 (d, 1H), 8.20 (s, 1H).

કુ

£

Step E: 7-chloro-3a-cyano-3a,4-dihydro-N-[4-(trifluoromethyl)-phenyl][1]benzopyrano[4,3-c]pyrazole-2(3H)-carboxamide

To a solution of diisopropylamine (0.42 mL, 3.0 mmoles) in 10 mL of THF was added, n-butyllithium (1.9 mL of 1.6 M, 3.0 mmole) at -78°C under N₂. After 15 min, a solution of the product of Step D (0.5 g, 1.3 mmole) in 3 mL of THF was added dropwise and the dark red solution that formed was stirred at -78°C for an additional 15 min. Then, p-toluene-sulfonyl cyanide (0.31 g, 1.7 mmole) in 2 mL of THF was added dropwise to the reaction mixture which decolorized near the end of the addition. Saturated aqueous NH₄Cl was added and the reaction was allowed to warm to room temperature. The aqueous phase was extracted with ethyl acetate and the combined organic phases were dried (MgSO₄) and concentrated (vacuum). The resulting solid was triturated with ether to afford the title compound (0.36 g, 66% yield) as a white solid, mp 227-229°C. ¹H NMR (CDCl₃): δ 8.10 (s, 1H), 7.91 (d, 1H), 7.62 (ABq, 4H),

30

35

5 7.30 (m, 2H), 4.79 (d, 1H), 3.81 (d, 1H), 3.41 (m, 1H), 3.04 (m, 1H), 2.68 (m, 1H), 2.15 (td, 1H).

EXAMPLE 2

Step A: 2-bromoethyl-1.3-butadiene

Phosphorous tribromide (5.13 mL, 0.054 mol) was added over 10
min to a solution of 11.3 g (0.13 mol) of 2-hydroxymethyl-1,3-butadiene
(R. Silverstein, et al., J. Org. chem., 1974, 39, 1957) and 270 mL of
diethyl ether at 0°C. The resulting solution was warmed to 24°C and
stirred at that temperature for 10 h and then cooled to 0°C and poured
into 500 mL of ice-water. The aqueous layer was separated and extracted
with two 100 mL portions of ether and the combined organic layers were
washed with 200 mL of saturated sodium bicarbonate solution, dried
(MgSO₄) and concentrated in vacuo at 24°C to give 10.5 g of a yellow oil
that is used immediately without further purification. 90 MHz ¹H NMR
(CDCl₃) δ 4.15 (s, 2H), 5.2-5.6 (m, 4H), 6.40 (dd, 1H).

20 Step B: 4-Chloro-2-hydroxybenzaldehyde

Modification of a procedure described by G. Casiraghi et al., (J. Chem. Soc., Perkin Transac. I., 1978, 318), was utilized in the preparation of substituted salicylaldehydes. To a mixture of 26 g of 3chlorophenol dissolved in 200 mL of toluene at 10°C was added dropwise 100 mL of a $2\underline{M}$ solution of ethylmagnesium chloride in diethyl ether. The ether was distilled off and the reaction mixture was cooled to room temperature and 26.0 g of 1,3-dimethyl-3,4,5,6-tetrahydro-2(1H)pyrimidinone (DMPU) was added followed by 15.0 g of paraformaldehyde. The reaction mixture was heated at 70°C for 18 h, cooled to 10°C and a solution of 20 mL of concentrated HCl in 100 g of ice was added followed by 200 mL of hexane. The organic phase was separated and washed twice with 100 mL of water, once with 100 mL of brine, dried over anhydrous MgSO4, filtered and concentrated under vacuum to afford a viscous oil. This was taken up in 150 mL of ethanol and a solution of 15.0 g of copper (II) acetate dissolved in 150 mL of ${
m H}_2{
m O}$ was added dropwise. The resulting precipitate was filtered, washed with water and then with ether and added to a solution of 200 mL of ether

and 20 mL of concentrated H₂SO₄ in 100 g of ice. The mixture was stirred until all solids were dissolved, 100 mL of hexane was added, and the organics were washed with 100 mL of H₂O, then 100 mL of brine, dried over anhydrous MgSO₄, filtered and concentrated under vacuum to afford 15.0 g of a yellow solid, mp 47-48°C. ¹H NMR (CDCl₃) δ 6.98 (d,

10 1H), 7.01 (s, 1H), 7.50 (d, 1H), 9.85 (s, 1H), 11.18 (s, 1H). IR (Nujol) 3300-2800, 1670 cm⁻¹.

15

30

35

Step C: 4-chloro-2-[(2-methylene-3-butenyl)oxylbenzaldehyde

To a solution of 4.1 g (0.026 mol) of 4-chloro-2-hydroxybenzaldehyde and 27 mL of DMF was added 5 g (0.034 mol) of 2-

bromomethyl-1,3-butadiene and 7.2 g (0.052 mol) of potassium carbonate. The resulting mixture was heated at 70°C for 4 h and then cooled to room temperature and poured into a mixture of 100 mL of water and 100 mL of methylene chloride. The aqueous layer was extracted with two 100 mL portions of methylene chloride and the combined organic layers were washed twice with 100 mL of water, twice with 100 mL of saturated

washed twice with 100 mL of water, twice with 100 mL of saturated sodium bicarbonate solution, dried over magnesium sulfate and concentrated to give 4.9 g of an oil. Purification by flash chromatography using 250 g of silica gel and eluting with 10:1 hexanes - EtOAc gave 2.6 g of the desired product. 200 MHz ¹H NMR (CDCl₃): δ 4.82 (s, 2H), 5.18-

25 5.42 (m, 4H), 6.45 (dd, 1H), 7.01 (s, 1H), 7.05 (d, 1H), 7.79 (d, 1H), 10.48 (s, 1H).

Step D: Diethyl[[4-chloro-2[(2-methylene-3-butenyl)oxylphenyl]methylenelphosphorohydrazidate

Diethyl phosphorohydrazidate (2.6 g, 0.015 mol) was added to a solution of 2.6 (0.012 mol) of the product of Step C, 0.8 mL of glacial acetic acid and 24 mL of ether at 0°C. The resulting solution was stirred at room temperature for 15 h and then was partitioned between methylene chloride and 1 M HCl. The aqueous layer was extracted with two portions of methylene chloride and the combined organics were washed twice with 1 M HCl, twice with saturated sodium bicarbonate solution, dried over anhydrous magnesium sulfate, filtered and concentrated to give 2.9 g of a yellow solid melting at 112-118°C. 200

÷

٩

5 MHz ¹H NMR (CDCl₃): δ 1.37 (t, 6H), 4.1-4.3 (m, 4H), 4.72 (s, 2H), 5.2-5.4 (m, 4H), 6.45 (dd, 1H), 6.75-7.0 (m, 3H), 7.81 (d, 1H), 8.05 (s, 1H).

Step E: Diethyl (7-chloro-3a-ethenyl-2,3,3a,4-tetrahydro[1]-benzopyrano[4,3-c]-pyrazol-2-yl) phosphonate

N-Chlorosuccinimide (1.0 g, 0.007 mol) was added to a mixture of 10 2.7 g (0.007 mol) of the product from Step D, 0.75 g of 3A molecular sieves and 120 mL of methylene chloride. The resulting mixture was cooled to 0°C and was treated with a solution of 1.3 mL of triethylamine and 60 mL of methylene chloride. The resulting mixture was stirred at room temperature for 4 h and then decanted into 100 mL of water. The 15 aqueous layer was extracted with two 50 mL portions of methylene chloride and the combined organic layers were washed with water and saturated sodium bicarbonate solution, dried over anhydrous magnesium sulfate and concentrated to give 2.9 g of a brown oil. Flash chromatography on silica gel using 1:3 hexanes- EtOAc afforded 1.0 g of a 20 light yellow oil. 200 MHz ¹H NMR (CDCl₃): δ 1.36 (t, 6H), 3.55 (dd, 1H), 3.9 (d, 1H), 4.1-4.3 (m, 5H), 4.60 (d, 1H), 5.15-5.3 (m, 2H), 5.90-6.05 (m, 1H), 6.9-7.0 (m, 2H), 7.78 (d, 1H).

Step F: 7-Chloro-3a-ethenyl-2.3.3a.4-tetrahydro-N-[4-(trifluoro-methyl)phenyl][1]benzopyrano[4, 3-c] pyrazole-2-carboxamide

25

30

35

A solution of 1.0 g (0.003 mol) of the product from Step E, 16 mL of methanol and 0.5 mL of concentrated hydrochloric acid was heated at reflux for 12 h and then was cooled to room temperature and concentrated. Trituration with ether gave 0.8 g of a tacky orange solid that was dissolved in 2 mL of water and treated with 0.6 g of sodium carbonate, 15 mL of methylene chloride, and 0.4 mL (0.003 mol) of ptrifluorotolylisocyanate. After stirring 12 h at room temperature, the reaction mixture was poured into water and the aqueous layer was extracted twice with methylene chloride. The combined organic layers were washed with water, dried over anhydrous magnesium sulfate, filtered and concentrated to give 0.9 g of a yellow solid melting at 157-

5 161°C. 200 MHz ¹H NMR (CDCl₃): δ 3.68 (d, 1H), 4.2-4.3 (overlapping doublets, 2H), 4.65 (d, 1H), 5.2-5.35 (m, 3 lines, 2H), 6.01 (dd, 1H), 6.95-7.05 (m, 2H), 7.62 (apparent q, 4H), 7.75 (d, 1H), 8.18 (br s, 1H).

Step G: 7-Chloro-2,3,3a,4-tetrahydro-3a-oxiranyl-N-[4-(trifluoro-methyl)phenyl][1]benzopyrano[4,3-c] pyrazole-2-carboxamide

10

15

20

25

30

35

The compound, m-chloroperbenzoic acid (0.7 g of 50% by weight material, 0.002 mol), was added to a solution of 0.3 g (0.0007 mol) of the product from Step F and 0.7 mL of chloroform. The resulting mixture was sittred at room temperature for 24 h and then was partitioned between 25 mL of methylene chloride and 25 mL of saturated sodium bicarbonate solution. The aqueous layer was extracted with two 15 mL portions of methylene chloride and the combined organic layers were washed with saturated sodium bicarbonate solution and then twice with saturated sodium thiosulfate solution, dried over anhydrous magnesium sulfate and concentrated. Flash chromatography on silica gel using 5:2 hexane- EtOAc gave 0.06 g of a solid melting at 187-190°C. 200 MHz ¹H NMR (CDCl₃): δ 2.65 (m, 1H), 2.75 (m, 1H), 3.35 (m, 1H), 3.50 (d, 1H), 4.06 (d, 1H) 4.26 (d, 1H), 4.75 (d, 1H), 7.0-7.1 (m, 2H), 7.60 (apparent q, 4H), 7.75 (d, 1H), 8.1 (brs, 1H).

Further elution with 1:1 hexane- EtOAc gave 0.02 g of the diastereomer as a solid melting at 79-81°C. 200 MHz 1 H NMR (CDCl₃): δ 2.78 (m, 2H), 3.22 (m, 1H), 3.61 (d, 1H), 4.20-4.28 (overlapping doublets, 2H) 4.63 (d, 1H), 7.0-7.1 (m, 2H), 7.62 (apparent q, 4H), 7.75 (d, 1H), 8.1 (brs, 1H).

EXAMPLE 3

Step A: 2-chloro-5-(1-methylethenyl)thiophene

A solution of 80 mL of 3<u>M</u> methyl magnesium bromide in ether was added to a solution of 32 g of 2-acetyl-5-chlorothiophene in 400 mL of ether at 0 to 10°C. After being stirred for 1 hour at 25°C, the reaction mixture was poured onto saturated aq NH₄Cl, washed with brine, dried (MgSO₄), and concentrated. The crude carbinol was dissolved in 400 mL of 1-chlorobutane, 0.5 g of potassium hydrogen sulfate was added, and

3

Ą

30

35

the mixture was heated at reflux with a water separator until the distillate was clear. The reaction mixture was washed with aq. NaHCO3, dried (MgSO4), and concentrated after first adding 0.1 g of 4-t-butyl catechol. Distillation at 90° at aspirator pressure afforded the title compound as a yellow oil. ¹H NMR (CDCl₃): δ 2.08 (s, 3H), 4.95 (s, 1H),
 5.25 (s, 1H), 6.77 (s, 2H).

Step B: 2-[[2-(5-chloro-2-thienyl)-2-propenyll-oxy-[-4-(trifluoromethyl)benzaldehyde

A solution containing 10 g of the product from Step A, 10 g of N-chlorosuccinimide, 0.8 g of diphenyl diselenide, and 0.5 mL of pyridine in 100 mL of dichloromethane was allowed to stir at 25°C for 20 hours. The mixture was washed with water, brine, dried (MgSO₄), concentrated, and chromatographed on silica gel with hexanes to provide 6.9 g of an oil which consisted of an approximately 1:1 mixture of vinylic and allylic chlorides by analysis by NMR. The allylic chloride had signals at δ (CDCl₃) 4.34 (s, 2), 5.32 (s, 1), 5.46 (s, 1), and 6.78 (s, 2).

To a solution of 6.8 g of this mixture in 15 mL of DMF was added 3.0 g of 4-trifluoromethyl salicylaldehyde, 3.0 g of potassium carbonate (anh.), and 0.2 g of sodium iodide, and the mixture was stored for 20 hours at 25°C. After adding 100 mL of water, the product was extracted with ethyl acetate, washed with aq NaOH, brine, dried (MgSO₄), and concentrated. The residue was chromatographed (10% EtOAc in hexane/silica gel) to provide 1.9 g of the title compound as a yellow solid, 1H NMR (CDCl₃): δ 4.98 (s, 2H), 5.39 (s, 1H), 5.61 (s, 1H), 6.87 (ABq, 2H), 7.3 (m, 2H), 7.98 (d, 1H), 10.5 (s, 1H).

Step C: diethyl [3a-(5-chloro-2-thienyl)-2,3,3a,4-tetrahydro-7-(trifluoromethyl)-[1]benzopyrano[4,3-c]pyrazol-2-yll-phosphonate

A solution of 2.5 g of the product from Step B and 1.5 g of diethyl phosphorohydrazidate in 20 mL of ethanol containing 2 drops of glacial acetic acid was kept for 1 hour at 25°, concentrated and extracted with

ether, washing with water and brine, and drying with MgSO₄. The dried and concentrated product (3 g) was redissolved in 200 mL of dichloromethane, cooled to 0°C, then, 1.2 g of N-bromosuccinimide and 0.6 g of triethylamine was added and the mixture was stirred for 15 minutes at 0°. The mixture was allowed to warm to room temperature, washed with aq. NaHCO₃, 1N HCl and brine, dried (MgSO₄), concentrated and chromatographed (EtOAc/silica gel) to afford 0.9 g of the product as a gum. ¹H NMR (CDCl₃) δ 1.37 (m, 6H), 3.8 (d of d, 1H), 4.2 (m, 5H), 4.4 (d, 1H), 4.93 (d, 1H), 6.8 (ABq, 2H), 7.2 (m, 2H), 7.95 (d, 1H).

15

25

30

35

Step D: 3a-(5-chloro-2-thienyl)-3a. 4-dihydro-7-(trifluoromethyl)-N[4-(trifluoromethyl)phenyl][1]benzopyrano[4.3-c]-pyrazole-2(3H)-carboxamide

To a solution of 0.9 g of the product from Step C in 10 mL of ethanol was added 0.5 mL of chlorotrimethylsilane and the mixture was heated at reflux for 20 hours. After concentration and trituration there was obtained 0.4 g of the HCl salt of the dephosphonylated product as a solid. ¹H NMR (Me₂SO-d₆): δ 3.6 (ABq, 2H), 4.5 (d, 1H), 5.0 (d, 1H), 6.6 (broad s), 6.9 (ABq, 2H), 7.3 (m, 2H), 7.83 (d, 1H).

A mixture of 0.3 g of this pyrazole product, 20 mL of CH₂Cl₂, 1 mL of saturated aq NaHCO₃, and 0.15 g of 4-trifluoromethylphenyl isocyanate was stirred at 25°C for 10 minutes. The organic layer was dried (MgSO₄), concentrated, and triturated with Et₂O/hexane to provide 0.3 g of the title compound, mp 184-187°. ¹H NMR (CDCl₃): δ 3.96 (d, 1H), 4.45 (d, 2H), 5.00 (d, 1H), 6.8 (ABq, 2H), 7.3 (m, 2H), 7.65 (ABq, 4H), 8.0 (d, 1H), 8.2 (brs, 1H).

EXAMPLE 4

Step A: 2-[(2-cyclohexen-1-)oxyl-4-(trifluoromethyl)benzaldehyde

To a solution of 3.6 g (0.026 moles) of K2CO3 in 75 mL of DMF was

added 5.0 g (0.026 moles) of 4-trifluoromethyl-2-hydroxybenzaldehyde

and 3.5 g (0.032 moles) of 3-bromo cyclohex-1-ene. The reaction mixture

20

25

30

35

was heated at 70°C for 5h, cooled to room temperature, poured into ice cold 5% HCl (200 mL), extracted with ethyl acetate (3 X 75 mL), washed organic phase with 1 N NaOH (2 x 50 mL), and with brine (100 mL), dried over anhydrous magnesium sulfate, filtered and concentrated in vacuo to afford 4.3 g of an oil. ¹H NMR (CDCl₃): δ 1.4-2.2 (m, 6H), 4.98
(m, 1H), 5.7-6.1 (m, 2H), 7.21 (d, 1H), 7.25 (s, 1H), 7.92 (d, 1H), 10.5 (s, 1H).

Step B: Diethyl[[2-[(2-cyclohexen-1-yl)oxy]-4-(trifluoromethyl)-phenyllmethylenelphosphorohydrazidate

To 4.3 g (0.019 mol) of the product obtained from Step A was added 50 mL of ether and 3.94 g (0.023 moles) of diethyl phosphorylhydrazidate. The reaction mixture was stirred at room temperature for 1 h, added 100 mL of ethyl acetate and extracted with brine (100 mL), dried over anhydrous magnesium sulfate, filtered and concentrated under vacuum to afford 6.6 g of an oil. 1 H NMR (CDCl₃): δ 1.36 (t, 6H), 1.6-2.2 (m, 6H), 4.18 (m, 4H), 4.84 (m, 1H), 5.88-6.04 (m, 2H), 7.0-7.25 (m, 3H), 7.99 (d, 1H), 8.12 (s, 1H). IR (neat) 3100,1030 cm⁻¹.

Step C: 2a.3.4.5.5a.10c-Hexahydro-8-(trifluoromethyl)-N-[4-(trifluoromethyl)phenyl]-2H-[1]benzopyrano[4.3.2-cd]-indazole-2-carboxamide

To 6.6 g (0.018 moles) of the product obtained from Step B was added 2.8 g (0.021 moles) of N-chlorosuccinimide and 150 mL of methylene chloride. 2.7 g (0.027 moles) of triethylamine was dissolved in 50 mL methylene chloride and added dropwise over 1 h while maintaining the reaction temperature at room temperature. The reaction mixture was extracted with 5% HCl (150 mL) and water (150 mL), dried over anhydrous magnesium sulfate, filtered and concentrated under vacuum to afford 6.6 g of a viscous oil which was dissolved in 60 mL of methanol and 5.8 mL of concentrated HCl. The reaction mixture was heated at reflux for 2 h, cooled and concentrated under vacuum, triturated with ether and the solid was filtered to afford 3.6 g of a yellow

solid. Then, 1.8 g (0.0063 moles) of this solid was dissolved in a solution of 50 mL of ethyl acetate and 50 mL of water and 0.870 g (0.0063 moles) of K₂CO₃ and 1.2 g (0.0063 moles) of α,α,α-trifluoro-p-tolyl isocyanate was added. The reaction mixture was stirred for 0.5 h, extracted with ethyl acetate (3 X 75 mL), washed with brine (100 mL) dried over
anhydrous magnesium sulfate, filtered and concentrated under vacuum to afford a yellow solid. The solid was chromatographed on silica gel using 25% THF/75% hexane to afford 0.970 g of a white solid, m.p. 194°-195°C. ¹H NMR (CDCl₃): δ 1.05 (m, 1H), 1.35 (m, 2H), 1.65 (m, 1H), 2.15 (m, 1H), 2.41 (m, 1H), 3.65 (t, 1H), 4.78 (t, 1H), 4.80 (t, 1H), 7.23 (s, 1H), 7.24 (d, 1H), 7.62 (ABq, 4H), 7.99 (d, 1H), 8.19 (s, 1H), I.R. (nujol) 3390, 1691 cm⁻¹.

EXAMPLE 5

2-(2.3.3A.8A-Tetrahydrocyclopent-((A))inden-8(1H)-ylidene-N-((4-(trifluoromethyl)phenyl))hydrazinecarboxamide

20

25

30

The compound, 1:2:3:3a:8:8a-hexahydrocyclopent[a]inden-8-one, (J. Chem. Soc. 3325 (1957)) (3.0 g) was treated with hydrazine hydrate (1.5 mL) and ethanol (20 mL). The mixture was refluxed for 3.5 h and concentrated to dryness. The residue was dissolved in methylene chloride (30 mL) and washed with water. The dried organic layer was concentrated. One third of the residue was dissolved in Et₂O (10 mL) and treated with p-trifluoromethylphenyl isocyanate (0.8 mL). An exothermic reaction ensued and a white precipitate was formed. After 1 h, the mixture was filtered and washed with ethyl ether to give the product (1.3 g) mp 214-215°C. HNMR (CDCl₃): δ 8.5 (NH), 8.3 (NH), 7.7-7.2 (Ar, M), 3.8 and 3.4 (M, CH₂) 2.1-1.2 (CH₂X₃).

3

Ē

5

10

EXAMPLE 6

Step A

2-Nitro-4-(trifluoromethyl)benzonitrile

A mixture of 45.0 g (0.20 mol) of 1-chloro-2-nitro-4-(trifluoromethyl)benzene, 39.4 g (0.44 mol) cuprous cyanide, 6.4 g (0.040 mol) bromine, 3.2 g (0.040 mol) of pyridine, and 200 mL of dimethylformamide was heated at reflux for 4 hours. After cooling the mixture to 20°C, 200 mL of water, 75 g of ferric chloride and 60 mL of concentrated hydrochloric acid were added. The mixture was extracted with three 500 mL portions of toluene. The combined toluene extracts were washed with water, dried over anhydrous magnesium sulfate, and the toluene was evaporated. The residue (33 g) crystallized on standing.

Recrystallization from a mixture of chlorobutane and cyclohexane yielded a colorless solid melting at 43-45°C. The IR (neat) spectra displays a strong band at 4.5 m.

20

25

15

Step B

2-Methoxy-4-(trifluoromethyl)benzonitrile

Twenty-five percent sodium methoxide in methanol (24.3 g, 11 mol) was added dropwise to a stirred mixture of 19 g (0.091 mol) 2-nitro-4-(trifluoromethyl)benzonitrile and 100 mL of methanol at 20-30°C. After 1 hour at 25°C, 100 mL of water was added. The solids were collected on a filter and washed with water. The solids were dried to yield 12.5 g of product melting at 59-61°C. 1H NMR (CDCl₃) δ 4.00 (s, 3H), 7.2-7.7 (3H).

30

35

Step C

2-Methoxy-4-(trifluoromethyl)benzaldehyde

Diisobutylaluminum hydride (1.5 <u>M</u> in toluene, 35 mL, 0.05 mol) was added over a 1 hour period to a mixture of 10.4 g (0.05 mol) 2-methoxy-4-(trifluoromethyl)benzonitrile and 50 mL toluene at -10°C. The mixture was agitated for 3 hours at 20-25°C and then poured into a mixture of 15 mL of concentrated hydrochloric acid and 50 mL of water at

0-5°C. The mixture was extracted with three 50 mL portions of toluene. The combined extracts were washed with water, dried over anhydrous magnesium sulfate, filtered, and evaporated. An oil (10.3 g) was obtained which, according to glc analysis, consisted of about 90% of the aldehyde, 4% of unreacted nitrile and 6% of the benzylamine. The infrared
spectrum exhibited a strong band at 5.9 μ and only a very weak band at 4.5 μ. Recrystallization from cyclohexane yielded colorless crystals melting at 54-55°C.
1H NMR (CDCl₃) δ 4.00 (s, 3H), 7.22 (s, 1H), 7.30 (d, 1H), 7.92 (d, 1H), 10.49 (s, 1H).

¥.

· &.

ð.

15

20

30

35

Step D

2-Hydroxy-4-(trifluoromethyl)benzaldehyde

A mixture of 10.0 g (0.042 mol) 2-methoxy-4-(trifluoromethyl)benzaldehyde, 5.3 g (0.126 mol) lithium chloride, and 50 mL of dimethylformamide was heated at 155°C for 3 hours. The reaction mass was cooled, diluted with 100 mL of water, acidified with hydrochloric acid, and then extracted three times with 100 mL of ether. The ether extracts were washed with water, dried over anhydrous magnesium sulfate, and evaporated to provide 6.8 g of the product as an oil.

25 ¹H NMR (CDCl₃) δ 7.01-7.88 (m, 4H), 11.16 (s, 1H).

Step E

Methyl 2-[[2-formyl-5-(trifluoromethyl)-phenoxylmethyl]-2-propenoate

A mixture of 19.0 g (0.10 mol) 2-hydroxy-4-(trifluoromethyl)-benzaldehyde, 13.8 g potassium carbonate, 19.6 g (0.11 mol) methyl(2-bromomethyl)-2-propenoate and 75 mL of dimethylformamide was heated at 50°C for 1 hour. After cooling to 20°C, 100 mL of water was added. The solids were collected on a filter, washed with water, and dried to yield 22.4 g of a colorless solid. The product was recrystallized from cyclohexane to yield a colorless solid melting at 77-79°C.

15

20

25

30

5 1_{H NMR} (CDCl₃) δ 3.84 (s, 3H), 4.92 (t, 2H), 6.06 (q, 1H), 6.50 (d, 1H), 7.26 (s, 1H), 7.32 (d, 1H), 7.96 (d, 1H), 10.53 (s, 1H).

Step F

Methyl 2-[[2-[[(diethoxyphosphinyl)hydrazonolmethyl]-5-(trifluoromethyl)phenoxylmethyl]-2-propenoate

To a mixture of 14.4 g (0.050 mol) of the product of Step E and 50 mL of diethylether, 13.2 g (0.052 mol, assay 66%) of diethyl phosphorohydrazidate was added at 25°C. After agitating for 2 hours, the solids were collected on a filter and washed with 25 mL of ether, then with water. The solids were dried to yield 18.4 g of a colorless solid melting at 143-144°C.

Step G

Methyl 2-(diethoxyphosphinyl)-2.3.3a.4-tetrahydro-7-(trifluoromethyl)[1]benzopyrano[4.3-c]pyrazole-3a-carboxylate

To a mixture of 21.9 g (0.050 mol) of the product of Step F, 7.4 g (0.055 mol) of N-chlorosuccinimide and 750 mL of methylene chloride were added, and the mixture was heated to reflux. A mixture of 7.5 g (0.07 mol) triethylamine and 50 mL of methylene chloride was added over a 4 hour period. The reaction mass was refluxed for an additional 1 hour. After cooling to 25°C, the methylene chloride solution was washed three times with 150 mL of water. The methylene chloride solution was dried with magnesium sulfate, filtered, and the solvent was evaporated. An oil remained (24 g) which solidified upon standing. The solid was recrystallized from a mixture of cyclohexane and ethyl acetate to yield a colorless solid melting at 85-86°C.

5 Step H

10

15

20

25

30

Methyl 2.3.3a.4-tetrahydro-7-(trifluoromethyl)[1]benzopyrano[4.3-c]pyrazole-3a-carboxylate.hydrochloride

A mixture of the product of Step G (38 g, 0.087 mol), 200 mL of methanol, and 18 g (0.50 mol) hydrogen chloride was heated at reflux (66°C) for 2 1/2 hours. The solution was concentrated, toluene (50 mL) was added to the residue and, the mixture concentrated again. Another 50 mL toluene was added and evaporated a second time. After adding 100 mL of toluene, the slurry was chilled, and then filtered. The solid was washed with toluene, then hexanes to yield 26 g of a pale yellow product, m.p. 175-176°C, dec. Recrystallization from a mixture of tetrahydrofuran and methanol gave a colorless solid, m.p. 185-186°C, dec.

Step I

Methyl 2-[[(6-chloro-3-pyridinyl)aminolcarbonyl]-2.3-dihydro-7-(trifluoromethyl)[1]benzopyrano-[4.3-c]pyrazole-3a(4H)-carboxylate

To a solution of 0.15 g (0.0005 mol) of bis(trichloromethyl)-carbonate, 20 mL of methylene chloride and 0.39 mL (0.0048 mol) of pyridine was added 0.5 g (0.0015 mol) of the product from Step H at 0°C. The resulting mixture was stirred at 0°C for 2 hours and then 0.21 g (0.0016 mol) of 5-amino-2-chloro pyridine was added. The resulting mixture was stirred at room temperature for 12 hours and then 10 mL of ethanol followed by 3 g of silica gel were added and the solvent was removed. Flash chromatography of the residue on silica gel using 2:1 hexanes-EtOAc gave 0.28 g of an off-white solid that melted at 236-238°C. ¹H NMR (CDCl₃, 200 MHz) δ 3.77 (s, 3H), 3.85 (d, 1H), 4.24 (d, 1H), 4.53 (d, 1H), 5.07 (d, 1H), 7.20-7.35 (m, 3H), 7.98 (d, 1H), 8.02 (s, 1H), 8.12 (dd, 1H), 8.42 (d, 1H).

÷.

15

25

30

35

EXAMPLE 7

Methyl 2.3-dihydro-7-(trifluoromethyl)-2-[[[5-(trifluoromethyl)-2-pyridinyllaminolcarbonyl]-[1]benzopyrano[4.3-c]pyrazole-3a(4H)-carboxylate

Application of the procedure described in Step I of Example 6,
increasing the amounts of reagents by a factor of 1.3 and substitution of
2-amino-5-trifluoromethyl pyridine afforded 0.14 g of a white solid, mp
194-195°C.

1_{H NMR} (CDCl₃, 200 MHz) δ 3.77 (s, 3H), 3.83 (d, 1H), 4.25 (d, 1H), 4.55 (d, 1H), 5.08 (d, 1H), 7.22 (s, 1H), 7.33 (d, 1H), 7.91 (d, 1H), 7.99 (d, 1H), 8.28 (d, 1H), 8.55 (s, 1H), 8.85 (s, 1H).

EXAMPLE 8

Methyl 2-[[(5-chloro-2-pyridinyl)aminolcarbonyll-2.3-dihydro-7-(trifluoromethyl)[1]benzopyrano-

20 [4.3-clpyrazole-3a(4H)-carboxylate

Application of the procedure described in Step I of Example 6, increasing the amounts of reagents by a factor of 1.5 and substitution of 2-amino-5-chloropyridine gave 0.42 g of a white solid, mp 196-198°C. 1H NMR (CDCl₃, 200 MHz) δ 3.76 (s, 3H), 3.82 (d, 1H), 4.23 (d, 1H), 4.55 (d, 1H), 5.05 (d, 1H), 7.21 (s, 1H), 7.32 (d, 1H), 7.68 (dd, 1H), 7.98 (d, 1H), 8.13 (d, 1H), 8.25 (d, J<1Hz, 1H), 8.65 (s, 1H).

EXAMPLE 9

Methyl 2-[[(5-bromo-2-pyridinyl)aminolcarbonyll-2.3-dihydro-7-(trifluoromethyl)[1]benzopyrano-[4.3-cl-pyrazole-3a(4H)-carboxylate

Application of the procedure described in Step I of Example 6, increasing the amounts of reagents by a factor of 1.5 and substitution of 2-amino-5-bromopyridine gave 0.40 g of a white solid, mp 179-181°C. 1H NMR (CDCl₃, 200 MHz) δ 3.76 (s, 3H), 3.82 (d, 1H), 4.22 (d, 1H), 4.52 (d, 1H), 5.05 (d, 1H), 7.22 (s, 1H), 7.30 (d, 1H), 7.80 (dd, 1H), 7.98 (d, 1H), 8.05 (d, 1H), 8.25 (d, J<1Hz, 1H), 8.65 (brs, 1H).

10

20

25

30

35

EXAMPLE 10

1-(4-Chlorophenyl)-N-(6-chloro-3-pyridinyl)-5-(4-fluorophenyl)-4.5-dihydro-1H-pyrazole-3-carboxamide

A solution of 0.9 g of 1-(4-chlorophenyl)-5-(4-fluorophenyl)-4,5-dihydro-1H-pyrazole-3-carbonyl chloride in 10 mL of THF was added to a solution of 0.4 g of 5-amino-2-chloropyridine, 2.0 mL of triethylamine and 10 mL of THF. The resulting suspension was stirred at room temperature for 2 hours. The solvent was evaporated and the residue was triturated with 10 mL of methanol to afford 0.66g of the title compound, mp 191-193°C.

¹H NMR (CDCl₃, 200 MHz) δ 3.15 (dd, 1H), 3.8 (dd, 1H), 5.4 (dd, 1H), 8.7-7.0 (m, 12H).

EXAMPLE 11

Step A: <u>1-[1-(bromomethyl)ethenyl]-4-chlorobenzene</u>

A mixture of 31.0 g (0.20 mol) of 4-chloro-α-methylstyrene, 38.0 g (0.21 mol) of N-bromosuccinimide, 3.9 g (0.02 mol) of p-toluenesulfonic acid monohydrate and 610 mL of THF was heated at reflux for 3 h and then concentrated at room temperature. The residue was partitioned between 400 mL of hexanes and 200 mL of H₂0, the organic phase was washed twice with 100 mL H₂O and dried (MgSO₄). Removal of solvent gave 47.8 g of a yellow oil that was used without further purification.

1H NMR (CDCl₃, 200 MHz) δ 4.33 (s, 2H), 5.50 (d, 2H), 7.2-7.5 (m, 5H).

Step B: 4-chloro-2-[[2-(4-chlorophenyl)-2-propenylloxylbenzaldehyde

A mixture of 19.2 g (0.083 mol) of the product from Step A, 10.0 g (0.064 mol) of 4-chlorosalicylaldehyde, 17.6 g (0.13 mol) of potassium carbonate and 64 mL of DMF was stirred at RT for 72 h. Resulting mixture treated with 125 mL of $\rm H_2O$ and filtered. Solids were washed with several portions of $\rm H_2O$ and dried in a vacuum oven at 40°C to give 21.0 g of a brown solid. ¹H NMR (CDCl₃, 200 MHz) δ 4.99 (s, 2H), 5.51 (s, 1H), 5.65 (s, 1H), 7.04-7.07 (m, 2H), 7.25-7.45 (m, 4H), 7.77 (d, 1H), 10.28 (s, 1H).

10

15

20

Step C: diethyl 7-chloro-3a-(4-chlorophenyl)-3a.4-dihydro-[1]benzopyrano[4.3-c]pyrazol-2(3H)-yl-phosphonate

A mixture of 11.3 g (0.04 mol) of the product from Step B and 73 mL of ether was treated sucessively with 2.4 mL of glacial acetic acid and 8.0 g (0.05 mol) of diethyl phosphorohydrazidate at 0°C. Resulting mixture was stirred at room temperature for 2.5 h and then filtered to give 7.8 g of a white solid.

A solution of 11.8 g (0.026 mol) of the product from the above procedure and 645 mL of methylene chloride was treated successively with 3.5 g (0.026 mol) of N-chlorosuccinimide and 4.5 mL (0.032 mol) of triethylamine at 0°C. The resulting mixture was allowed to stir at room temperature for 72 h and was then treated with 300 mL H₂O. The aqueous layer was extracted with 100 mL methylene chloride and the combined organic layers were washed with 200 mL of 1N HCl, 200 mL of H₂O, dried (MgSO₄) and concentrated to give 11.3 g of a tacky yellow solid.

1H NMR (CDCl₃, 200 MHz) δ 1.34 (t, 6H), 3.77 (dd, 1H), 3.93 (d, 1H), 4.0-4.3 (m, 4H), 4.38 (d, 1H), 5.00 (d, 1H), 6.82 (d, 1H), 6.97 (dd, 1H), 7.25 (ABq, 4H), 7.83 (d, 1H).

25

30

35

Step D: 7-chloro-3a-(4-chlorophenyl)-N-(6-chloro-3-pyridinyl)-3a.4-dihydro[1]benzopyrano[4,3-clpyrazole-2(3H)-carboxamide

Concentrated hydrochloric acid (37%, 4.9 mL) was added to a solution of 11.2 g (0.025 mol) of the product from Step C and methanol (148 mL) at room temperature. The resulting solution was heated at reflux for 7 h, cooled to room temperature and concentrated. The resulting brown residue was triturated with Et₂O and filtered to give 7.9 g of a light brown solid.

The product from the above procedure (7.9 g) was dissolved in a mixture of 60 mL of chloroform and 20 mL of saturated sodium bicarbonate. The organic layer was dried over anhydrous K₂CO₃ and filtered. The filtrate was treated with 23 mL of a solution of phosgene in

toluene (1.93 M, 0.044 mol) at 0°C. The resulting mixture was stirred at room temperature for 1 h and then concentrated to give 8.4 g of a brown semisolid that was used without further purification.

A mixture of 1.5 g of the product from the above procedure and 15 mL of CH₂Cl₂ was treated successively with 1.4 mL of N,N-diisopropyl ethylamine and 0.50 g (0.004 mol) of 5-amino-2-chloropyridine at room temperature. The resulting mixture was stirred for 12 h and then 10 mL of acetone and 4 g of silica gel were added and the mixture was concentrated. The residue was chromatographed on 100 g of silica gel eluting with 2:1 hexanes- EtOAc to give 0.84 g of a pale yellow solid melting at >250°C.

¹H NMR (200 MHz, CDCl₃) δ 3.92 (d, 1H), 4.23 (d, 1H), 4.42 (d, 1H), 5.08 (d, 1H), 6.90 (d, 1H), 7.05 (d, 1H), 7.19-7.35 (m, 5H), 7.82 (d, 1H), 8.08-8.18 (m, 2H), 8.41 (d, 1H).

By the general procedures described herein, or obvious 20 modifications thereof, the compounds of Tables 1 through 14 can be prepared.

In Tables 1-14 the following notations have been used.

25

Ŧ

$$\begin{array}{c} 2 \\ 3 \\ 4 \end{array}$$

$$-CH_{2} \xrightarrow{2} \xrightarrow{3} 4$$

$$-CH_2CH_2 \xrightarrow{1 \atop 6} \overset{2}{\overbrace{}^3} \overset{3}{\underbrace{}} \overset{4}{\underbrace{}}$$

$$Et =$$

$$-CH_2CH_3$$

$$n-Pr =$$

$$-CH_2CH_2CH_3$$

$$-CH(CH_3)_2$$

Key for Table 1

$$\begin{array}{c|c}
R^2 & \xrightarrow{5} & \xrightarrow{4} & \xrightarrow{A} & R^3 \\
\hline
7 & & & & & & \\
\hline
7 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & & & \\
\hline
0 & & & & \\
\hline
0 & & & & & \\
\hline$$

Group	\mathbb{R}^1
а	CF ₃
Ъ	OCF ₃
c	Cl
d	\mathbf{Br}

Compounds of Table 1 wherein A, R², and R³ are as set out therein can be prepared having the recited values of Groups a through d. That is, for each value of A, R², and R³ in Table 1, R¹ can be CF₃, OCF₃, Cl or Br. All of said compounds are specifically included within the scope of this invention.

15

20

TABLE 1

This Table contains a large number of compounds of Formula I in a format adopted to avoid mechanical reproduction of substituent values that do not vary. For example, the first table entry (where A is CH₂CH₂, R² is 4-F and R³ is CN) actually specifies 4 separate and distinct compounds because the key for Table 1 identifies additional substituent values of R¹, namely Group a through d for each Table 1 entry.

A is CH ₂ CH ₂ ;	A is CH_2CH_2 ;	A is CH_2CH_2 ;
\mathbb{R}^2 is 4-F	\mathbb{R}^2 is 5-F	\mathbb{R}^2 is 5-Cl
\mathbb{R}^3	\mathbb{R}^3	\mathbb{R}^3
CN	CN	CN
NO_2	$^{\circ}$ NO $_{f 2}$	NO_2
N_3	N_3	N_3
Br	Br	Br
NMe ₂	NMe ₂	NMe_2
NHCO ₂ Me	NHCO ₂ Me	${ m NHCO_2Me}$
epoxyethyl	epoxyethyl	epoxyethyl
epoxyisopropenyl SiMe3	epoxyisopropenyl SiMe3	epoxyisopropenyl SiMe3
SMe	SMe	SMe
S(O)Me	S(O)Me	S(O)Me
SO ₂ Me	SO_2Me	$\mathrm{SO}_{2}\mathrm{Me}$
$P(O)(OMe)_2$	$P(O)(OMe)_2$	$P(O)(OMe)_2$
CH ₂ C(O)NMe ₂	$\mathrm{CH_2C}(\mathrm{O})\mathrm{NMe_2}$	${ m CH_2C}(O){ m NMe_2}$
CH ₂ SMe	$\mathrm{CH_2SMe}$	$\mathrm{CH_2SMe}$
CH ₂ OCH ₂ CF ₃	$\mathrm{CH_2OCH_2CF_3}$	CH2OCH2CF3
CH ₂ SiMe ₃	$\mathrm{CH}_2\mathrm{SiMe}_3$	$\mathrm{CH_2SiMe_3}$
CH ₂ NMe ₂	CH_2NMe_2	$\mathrm{CH_2NMe_2}$
CH ₂ -2-pyridyl	CH ₂ -2-pyridyl	CH ₂ -2-pyridyl
2-pyridyl C(H)=NOCH3	2-pyridyl C(H)=NOCH3	2-pyridyl C(H)=NOCH3
3-pyridyl	3-pyridyl	3-pyridyl
2-thienyl	2-thienyl	2-thienyl
3-thienyl	3-thienyl	3-thienyl
2-oxazolinyl	2-oxazolinyl	2-oxazolinyl
2-(1,3-dioxolanyl)	2-(1,3-dioxolanyl)	2-(1,3-dioxolanyl)

A is CH_2CH_2 ;	A is OCH_2 ;	A is OCH_2 ;
\mathbb{R}^2 is 5-CF ₃ ;	\mathbb{R}^2 is 4-F;	\mathbb{R}^2 is 5-F
•	•	
<u>R</u> 3	\mathbb{R}^3	\mathbb{R}^3
CN	CN	CN
NO ₂	NO_2	NO_2
N ₃	N_3	N_3
Br	Br	Br
NMe ₂	NMe_2	NMe_2
NHCO ₂ Me	NHCO ₂ Me	NHCO ₂ Me
epoxyethyl	epoxyethyl	epoxyethyl
epoxyisopropenyl	epoxyisopropenyl	epoxyisopropenyl
SiMe ₃	SiMe3	SiMe ₃
SMe	SMe	SMe
S(O)Me	S(O)Me	S(O)Me
SO ₂ Me	SO_2Me	SO ₂ Me
$P(O)(OMe)_2$	$P(O)(OMe)_2$	$P(O)(OMe)_2$
$CH_2C(O)NMe_2$	$CH_2C(O)NMe_2$	$CH_2C(O)NMe_2$
CH ₂ SMe	$\mathrm{CH_2SMe}$	$\mathrm{CH_2SMe}$
CH ₂ OCH ₂ CF ₃	$CH_2OCH_2CF_3$	CH ₂ OCH ₂ CF ₃
CH ₂ SiMe ₃	${ m CH_2SiMe_3}$	$\mathrm{CH_2SiMe_3}$
CH_2NMe_2	$\mathrm{CH_2NMe_2}$	CH ₂ NMe ₂
CH ₂ -2-pyridyl	CH ₂ -2-pyridyl	CH ₂ -2-pyridyl
2-pyridyl	2-pyridyl	2-pyridyl
3-pyridyl	3-pyridyl	3-pyridyl
2-thienyl	2-thienyl	2-thienyl
3-thienyl	3-thienyl	3-thienyl
2-oxazolinyl	2-oxazolinyl	2-oxazolinyl
2-(1,3-dioxolanyl)	2-(1,3-dioxolanyl)	2-(1,3-dioxolanyl)

A is OCH₂; R² is 5-Cl;

R³
CN
NO₂
N₃
Br
NMe₂
NHCO₂Me
epoxyethyl
epoxyisopropenyl
SiMe₃

 $\begin{array}{l} \mathrm{SMe} \\ \mathrm{S(O)Me} \\ \mathrm{SO}_2\mathrm{Me} \\ \mathrm{P(O)(OMe)}_2 \\ \mathrm{CH}_2\mathrm{C(O)NMe}_2 \\ \mathrm{CH}_2\mathrm{SMe} \end{array}$

A is OCH_2 ; R^2 is 5-CF₃;

R³
CN
NO₂
N₃
Br
NMe₂
NHCO₂Me
epoxyethyl
epoxyisopropenyl
SiMe₃
SMe
S(O)Me
SO₂Me

 ${
m CH_2C(O)NMe_2}$ ${
m CH_2SMe}$

 $P(O)(OMe)_2$

A is NHCH₂;

 R^2 is 4-F

R³
CN
NO₂
NMe₃
NHCO₂Me
epoxyethyl
SiMe₃

 $\begin{array}{c} \mathrm{SMe} \\ \mathrm{P(O)(OMe)_2} \\ \mathrm{CH_2SMe} \end{array}$

CH₂OCH₂CF₃ CH₂SiMe₃ 2-pyridyl 2-thienyl A is OCH_2 ; R² is 5-Cl;

R³
CH₂OCH₂CF₃
CH₂SiMe₃
CH₂NMe₂
CH₂-2-pyridyl
2-pyridyl
3-pyridyl
2-thienyl
3-thienyl
2-oxazolinyl
2-(1,3-dioxolanyl)
C(H)=NOCH₃
C(CH₃)=NOCH₃
C(CN)=NOCH₃

C(OCH₃)=NOCH₃

A is OCH₂; R² is 5-CF₃;

R³
CH₂OCH₂CF₃
CH₂SiMe₃
CH₂NMe₂
CH₂-2-pyridyl
2-pyridyl
3-pyridyl
2-thienyl
3-thienyl
2-oxazolinyl
2-(1,3-dioxolanyl)
C(SCH₃)=N(OCH₃)
C(H)=NOCH₃
C(H)=NOCH₂CH₃
C(CH₃)=NOCH₃

C(H)=NOH

A is NHCH₂; R₂ is 5-F;

R³
CN
NO₂
NMe₂
NHCO₂Me
epoxyethyl
SiMe₃
SMe
P(O)(OMe)₂
CH₂SMe
CH₂OCH₂CF₃
CH₂SiMe₃
2-pyridyl

2-thienyl

A is NHCH ₂ ;	A is N(Me)CH ₂ ;	A is N(Me)CH ₂ ;
\mathbb{R}^2 is 5-Cl;	\mathbb{R}^2 is 4-F;	\mathbb{R}^2 is 5-Cl
R3	R ³	\mathbb{R}^3
CN	ČN	CN
NO ₂	NO_2	NO_2
NMe ₂	NMe ₂	NMe ₂
NHCO ₂ Me	NHCO ₂ Me	NHCO ₂ Me
epoxyethyl	epoxyethyl	epoxyethyl
SiMe3	SiMe3	SiMeg
SMe	SMe	SMe
$P(O)(OMe)_2$	$P(O)(OMe)_2$	$P(O)(OMe)_2$
CH ₂ SMe	CH ₂ SMe	CH ₂ SMe
CH ₂ OCH ₂ CF ₃	$\mathrm{CH_2OCH_2CF_3}$	$CH_2OCH_2CF_3$
CH ₂ SiMe ₃	CH ₂ SiMe ₃	${ m CH_2SiMe_3}$
2-pyridyl	2-pyridyl	2-pyridyl
2-thienyl	2-thienyl	2-thienyl

A is N(Me)CH₂; A is NHCH2; A is $N(Me)CH_2$; \mathbb{R}^2 is 5-CF₃; \mathbb{R}^2 is 5-F; \mathbb{R}^2 is 5-CF3; \mathbb{R}^3 \mathbb{R}^3 \mathbb{R}^3 CN CN CN NO_2 NO_2 NO_2 NMe₂ NMe₂ NMe₂ NHCO₂Me NHCO₂Me NHCO₂Me epoxyethyl epoxyethyl epoxyethyl SiMeg SiMe₃ SiMeg SMe SMe SMe $P(O)(OMe)_2$ $P(O)(OMe)_2$ $P(O)(OMe)_2$ CH₂SMe CH₂SMe CH_2SMe $CH_2OCH_2CF_3$ CH₂OCH₂CF₃ CH₂OCH₂CF₃ CH₂SiMe₃ CH₂SiMe₃ CH₂SiMe₃ 2-pyridyl 2-pyridyl 2-pyridyl 2-thienyl 2-thienyl 2-thienyl C(CH₃)=NOCH₃ C(OCH₃)=NOCH₃ $C(H)=NOCH_3$

KEY FOR TABLE 2

$$R^2 \xrightarrow{\frac{5}{6}} \xrightarrow{\frac{4}{7}} \xrightarrow{A} \xrightarrow{R^3} C-NH$$

Group R¹
a CF₃
b OCF₃

Compounds of Table 2 wherein A, R², and R³ are as set out therein can be prepared having the recited values of Groups a through b. That is, for each value of A, R² and R³, R¹ can be CF₃ or OCF₃. All of said compounds are specifically included within the scope of this invention.

15

20

10

TABLE 2

This Table contains a large number of compounds of Formula I in a format adopted to avoid mechanical reproduction of substituent values that do not vary. For example, the first table entry (where A is CH₂CH₂, R² is 5-F and R³ is CN) actually specifies 2 separate and distinct compounds because the key for Table 2 identifies additional substituent values of R¹, namely Groups a and b for each Table 2 entry.

A is CH_2CH_2 ;	A is CH_2CH_2 ;	A is OCH ₂ ;
\mathbb{R}^2 is 5-F;	\mathbb{R}^2 is 5-CF ₃ ;	\mathbb{R}^2 is 5-F;
_		_
<u>R</u> 3	\mathbb{R}^3	<u>R</u> 3
CN	CN	CN
NO_2	NO_2	NO_2
N_3	N_3	N_3
Br	Br	Br
NMe ₂	NMe ₂	NMe ₂
NHCO ₂ Me	NHCO ₂ Me	NHCO ₂ Me
epoxyethyl	epoxyethyl	epoxyethyl
epoxyisopropenyl	epoxyisopropenyl	epoxyisopropenyl
SiMe ₃	SiMe3	SiMe3
SMe	SMe	SMe
S(O)Me	S(O)Me	S(O)Me
SO ₂ Me	SO ₂ Me	SO_2Me
$P(O)(OMe)_2$	$P(O)(OMe)_2$	$P(O)(OMe)_2$
$CH_2C(O)NMe_2$	$CH_2C(O)NMe_2$	$CH_2C(O)NMe_2$
CH ₂ SMe	CH ₂ SMe	CH ₂ SMe
$CH_2OCH_2CF_3$	CH ₂ OCH ₂ CF ₃	$CH_2OCH_2CF_3$
$\mathrm{CH_2SiMe_3}$	${ m CH_2SiMe_3}$	CH ₂ SiMe ₃
$\mathrm{CH_2NMe_2}$	CH ₂ NMe ₂	${ m CH_2NMe_2}$
CH ₂ -2-pyridyl	CH ₂ -2-pyridyl	CH ₂ -2-pyridyl
2-pyridyl	2-pyridyl	2-pyridyl
3-pyridyl	3-pyridyl	3-pyridyl
2-thienyl	2-thienyl	2-thienyl
3-thienyl	3-thienyl	3-thienyl
2-oxazolinyl	2-oxazolinyl	2-oxazolinyl
2-(1,3-dioxolanyl)	2-(1,3-dioxolanyl)	2-(1,3-dioxolanyl)
	•	

A is OCH₂;

 \mathbb{R}^2 is 5-CF3;

 \mathbb{R}^3

CN NO₂

 N_3

Br

Br NMe₂

 $NHCO_2Me$

epoxyethyl

epoxyisopropenyl

SiMeg

SMe

S(O)Me

SO₂Me P(O)(OMe)₂

CH₂C(O)NMe₂

CH2C(U)MMe2

 ${
m CH_2SMe}$

CH₂OCH₂CF₃

 CH_2SiMe_3

CH₂NMe₂

CH₂-2-pyridyl

2-pyridyl

3-pyridyl

2-thienyl

3-thienyl

2-oxazolinyl

2-(1,3-dioxolanyl)

A is NHCH2;

 \mathbb{R}^2 is 5-F;

 \mathbb{R}^3

CN NO₂

NMe₂

NHCO₂Me

epoxyethyl

SiMeg

SMe

 $P(O)(OMe)_2$

CH₂SMe

CH₂OCH₂CF₃

CH2SiMe3

2-pyridyl

2-thienyl

A is N(Me)CH2;

 \mathbb{R}^2 is 5-F;

 \mathbb{R}^3

CN NO₂

NMe₂

NHCO₂Me

epoxyethyl

SiMe3

SMe

 $P(O)(OMe)_2$

CH₂SMe

CH₂OCH₂CF₃

CH₂SiMe₃

2-pyridyl

2-thienyl

٤

A is NHCH₂;

A is N(Me)CH2;

 R^2 is 5-CF₃

 \mathbb{R}^2 is 5-CF3;

 \mathbb{R}^3

 \mathbb{R}^3

CN NO CN

 NO_2

 NO_2

 NMe_2

 NMe_2

NHCO₂Me

 ${
m NHCO_2Me}$

epoxyethyl

epoxyethyl SiMe₃

SiMe₃

SMe

SMe P(O)(OMe)₂

 $P(O)(OMe)_2$

CH₂SMe

CH₂SMe

CH₂OCH₂CF₃

CH₂OCH₂CF₃

CH₂SiMe₃

CH₂SiMe₃

2-pyridyl

2-pyridyl

2-thienyl

2-thienyl

a

5

KEY FOR TABLE 3

CH₂ OCF₃ b

CF₃ 0 C

OCF₃ 0 d

10

15

20

Compounds of Table 3 wherein \mathbb{R}^2 , \mathbb{R}^3 and \mathbb{R}^4 are as set out therein can be prepared having the recited values of groups a through d. That is, for each value of \mathbb{R}^2 , \mathbb{R}^3 and \mathbb{R}^4 in the Table, \mathbb{R}^1 and A can have the value verified in each of the Groups a through d above. All of said compounds are specifically included within the scope of this invention.

TABLE 3

This Table contains a large number of compounds of Formula I in a format adopted to avoid mechanical reproduction of substituent values that do not vary. For example, the first table entry (where ${\bf R}^2$ is 5-F, ${\bf R}^4$ is H and \mathbb{R}^3 is NO_2) actually specifies 4 separate and distinct compounds

because the key for Table 3 identifies additional substituent values of \mathbb{R}^1 and A, namely Groups a and d for each Table 3 entry.

\mathbb{R}^2 is 5-F;	\mathbb{R}^2 is 5-Cl;	\mathbb{R}^2 is 5-F;
\mathbb{R}^4 is H;	\mathbb{R}^4 is H;	R ⁴ is Me;
R ³ NO ₂ N ₃	R ³ NO ₂ N ₃	$ m R^3$ $ m NO_2$ $ m NHCO_2Me$
Br NHCO-M-	Br NHCO-Mo	epoxyethyl
NHCO ₂ Me epoxyethyl	NHCO ₂ Me epoxyethyl	SiMe ₃ P(O)(OMe) ₂
epoxyisopropenyl SiMe ₃ P(O)(OMe) ₂	epoxyisopropenyl SiMe3 P(O)(OMe) ₂	CH ₂ SMe CH ₂ OCH ₂ CF ₃ CH ₂ SiMe ₃
CH ₂ C(O)NMe ₂ CH ₂ SMe CH ₂ OCH ₂ CF ₃	CH ₂ C(O)NMe ₂ CH ₂ SMe CH ₂ OCH ₂ CF ₃	2-pyridyl 2-thienyl
CH ₂ SiMe ₃ CH ₂ NMe ₂ CH ₂ -2-pyridyl	CH ₂ SiMe ₃ CH ₂ NMe ₂ CH ₂ -2-pyridyl	R^2 is 5-Cl; R^4 is Me;
2-pyridyl 3-pyridyl 2-thienyl	2-pyridyl 3-pyridyl 2-thienyl	R ³ NO ₂ NHCO ₂ Me
3-thienyl 2-oxazolinyl 2-(1,3-dioxolanyl) C(H)=NOCH ₃	3-thienyl 2-oxazolinyl 2-(1,3-dioxolanyl) C(H)=NOCH ₃	epoxyethyl SiMe3 P(O)(OMe) ₂ CH ₂ SMe

 \mathbb{R}^2 is 5-Cl; \mathbb{R}^2 is 5-F; \mathbb{R}^4 is H; \mathbb{R}^4 is H; \mathbb{R}^3 \mathbb{R}^3 CH₂OCH₂CF₃ CH₂OCH₂CF₃ CH₂SiMe₃ CH₂SiMe₃ 2-pyridyl 2-pyridyl 2-thienyl 2-thienyl \mathbb{R}^2 is 5-F: \mathbb{R}^2 is 5-F; \mathbb{R}^4 is Ph; \mathbb{R}^3 R^4 is 4-F-Ph; $\overline{N}0_2$ \mathbb{R}^3 NHCO₂Me $\overline{N}O_2$ epoxyethyl NHCO₂Me SiMe₃ epoxyethyl $P(O)(OMe)_2$ SiMeg CH₂SMe $P(O)(OMe)_2$ CH2OCH2CF3 CH_2SMe CH₂SiMe₃ CH2OCH2CF3 2-pyridyl CH₂SiMe₃ 2-thienyl 2-pyridyl 2-thienyl

R² is 5-F; R⁴ is Me; R³ CH₂OCH₂CF₃ CH₂SiMe₃ 2-pyridyl

2-thienyl

 R^2 is 5-Cl; R^2 is 5-Cl; R^4 is Ph; R^4 is 4-F-Ph;

R³ R³ NO₂ NO₂

NHCO₂Me
epoxyethyl
siMe₃
epoxyethyl
SiMe₃
P(O)(OMe)e

 $\begin{array}{ll} P(O)(OMe)_2 & \quad P(O)(OMe)_2 \\ CH_2SMe & \quad CH_2SMe \end{array}$

CH₂OCH₂CF₃ CH₂OCH₂CF₃ CH₂SiMe₃ CH₂SiMe₃ 2-pyridyl 2-pyridyl 2-thienyl 2-thienyl

3

KEY FOR TABLE 4

Group	<u>R</u> 1	\mathbb{R}^2	
a	CF3	4-Cl	
b	OCF3	4-Cl	
c	CF ₃	4-F	
d	OCF ₃	4-F	

10

Compounds of Table 4 wherein \mathbb{R}^3 and \mathbb{R}^4 are as set out therein can be prepared having the recited values of Groups a through d. That is, for each value of \mathbb{R}^3 and \mathbb{R}^4 in the Table, \mathbb{R}^1 and \mathbb{R}^2 can have the specific values recited in each of Groups a through d above. All of said compounds are compounds of this invention.

15

TABLE 4

This Table contains a large number of compounds of Formula I in a format adopted to avoid mechanical reproduction of substituent values that do not vary. For example, the first table entry (where R^4 is H and R^3 is NO_2) actually specifies 4 separate and distinct compounds because the key for Table 4 identifies additional substituent values of R^1 and R^2 , namely Groups a through d for each Table 4 entry.

 R^4 is H;

 \mathbb{R}^4 is Ph;

R³ NO₂ N₃ Br

Br NMe₂ NHCO₂Me

epoxyethyl epoxyisopropenyl SiMe3

 $SMe \\ S(O)Me \\ SO_2Me \\ P(O)(OMe)_2$

 $P(O)(OMe)_2$ $CH_2C(O)NMe_2$

CH₂SMe

 $CH_2OCH_2CF_3$

CH₂SiMe₃ CH₂NMe₂ $m R^3$ NO $_2$ NMe $_2$

NHCO₂Me epoxyethyl SiMe₃ SMe

 $P(O)(OMe)_2$ CH_2SMe

 $\begin{array}{l} \mathrm{CH_{2}OCH_{2}CF_{3}} \\ \mathrm{CH_{2}SiMe_{3}} \\ \mathrm{C(H)=NOCH_{3}} \end{array}$

4

•

\mathbb{R}^4 is 4-F-Ph;
\mathbb{R}^3
NO_2
NMe_2
NHCO ₂ Me
epoxyethyl
SiMe ₃
\mathbf{SMe}
$P(O)(OMe)_2$
$\mathrm{CH_2SMe}$
CH ₂ OCH ₂ CF ₃
CH ₂ SiMe ₃

KEY FOR TABLE 5

Groups	\mathbb{R}^1	Y
а	CF3	CH_2
b	OCF ₃	CH_2
c	CF ₃	0
đ	ocf3	О

10

15

Compounds of Table 5 wherein R^2 and R^3 are as set out therein can be prepared having the recited values of Groups a through d. That is, for each value of R^2 and R^3 in the Table, R^1 and V can have the values recited in each of the Groups a through d above. All of said compounds are of this invention.

TABLE 5

This Table contains a large number of compounds of Formula I in a format adopted to avoid mechanical reproduction of substituent values that do not vary. For example, the first table entry (where R² is 5-F and R³ is CN) actually specifies 4 separate and distinct compounds because the key for Table 5 identifies additional substituent values of R¹ and V, namely Groups a through d for each Table 5 entry.

3

\mathbb{R}^2 is 5-F;	\mathbb{R}^2 is $\mathbb{C}\mathbb{I}$;
<u>R</u> 3	<u>R</u> 3
CN	CN
NO_2	NO_2
NMe ₂	NMe ₂
NHCO ₂ Me	${ m NHCO_2Me}$
epoxyethyl	epoxyethyl
SiMe ₃	SiMe3
SMe	SMe
$P(O)(OMe)_2$	$P(O)(OMe)_2$
CH ₂ SMe	$\mathrm{CH_2SMe}$
CH ₂ OCH ₂ CF ₃	$CH_2OCH_2CF_3$
CH ₂ SiMe ₃	$\mathrm{CH_2SiMe_3}$
2-pyridyl	2-pyridyl
2-thienyl	2-thienyl

KEY FOR TABLE 6

Groups	<u>R</u> 1	<u>v</u>
а	CF ₃	CH ₂
b	OCF3	CH ₂
c	CF ₃	o
d	OCF3	O

Compounds of Table 6 wherein R^2 and R^3 are as set out therein can be prepared having the recited values of Groups a through d. That is, for each value of R^2 and R^3 in the table, R^1 and V can have the values recited in Groups a through d above. All of said compounds are of this invention.

15

20

10

TABLE 6

This Table contains a large number of compounds of Formula I in a format adopted to avoid mechanical reproduction of substituent values that do not vary. For example, the first table entry (where R^2 is 5-F and R^3 is CN) actually specifies 4 separate and distinct compounds because the key for Table 6 identifies additional substituent values of R^1 and V, namely Groups a through d for each Table 6 entry.

Ē

 R^2 is 5-F; R^2 is 5-Cl;

 \mathbb{R}^3 \mathbb{R}^3 CN CN NO_2 NO_2 NMe_2 NMe_2 NHCO₂Me $NHCO_2Me$ epoxyethyl epoxyethyl SiMe3 SiMeg SMe SMe $P(O)(OMe)_2$ $P(O)(OMe)_2$

 $\begin{array}{ll} \text{CH}_2\text{SMe} & \text{CH}_2\text{SMe} \\ \text{CH}_2\text{OCH}_2\text{CF}_3 & \text{CH}_2\text{OCH}_2\text{CF}_3 \\ \text{CH}_2\text{SiMe}_3 & \text{CH}_2\text{SiMe}_3 \end{array}$

2-pyridyl 2-pyridyl 2-thienyl 2-thienyl

KEY FOR TABLE 7

Group	\mathbb{R}^1	A .
а	CF ₃	CH ₂
b	OCF ₃	CH ₂
c ·	CF ₃	0
d	OCF3	0

10

15

20

Compounds of Table 7 wherein R^2 , R^3 and R^4 are as set out therein can be prepared having the recited values of Groups a through d. That is, for each value of R^2 , R^3 and R^4 in the Table, R^1 and A can have the values recited in each of the Groups a through d above. All of said compounds are within the scope of this invention.

TABLE 7

This Table contains a large number of compounds of Formula I in a format adopted to avoid mechanical reproduction of substituent values that do not vary. For example, the first table entry (where R² is 5-F, R⁴ is H and R³ is CN) actually specifies 4 separate and distinct compounds because the key for Table 7 identifies additional substituent values of R¹ and A, namely Groups a through d for each Table 7 entry.

\mathbb{R}^2 is 5-F;	\mathbb{R}^2 is 5-F;	\mathbb{R}^2 is 5-F;
\mathbb{R}^4 is H;	\mathbb{R}^4 is Me;	\mathbb{R}^4 is Ph;
<u>R</u> 3	\mathbb{R}^3	\mathbb{R}^3
CN	CN	CN
NO_2	NO_2	NO_2
NMe ₂	NMe_2	NMe_2
NHCO ₂ Me	NHCO ₂ Me	NHCO ₂ Me
epoxyethyl	epoxyethyl	epoxyethyl
SiMe3	SiMe3	SiMe3
SMe	SMe	SMe
$P(O)(OMe)_2$	$P(O)(OMe)_2$	$P(O)(OMe)_2$
CH ₂ SMe	CH ₂ SMe	CH ₂ SMe
CH ₂ OCH ₂ CF ₃	CH ₂ OCH ₂ CF ₃	CH ₂ OCH ₂ CF ₃
$\mathrm{CH_2SiMe_3}$	CH ₂ SiMe ₃	CH ₂ SiMe ₃
2-pyridyl	2-pyridyl	2-pyridyl
2-thienyl	2-thienyl	2-thienyl
\mathbb{R}^2 is 5-F;	\mathbb{R}^2 is 5-C1;	\mathbb{R}^2 is 5-Cl;
	R^4 is Me;	
R ⁴ is 4-F-Ph;	K* is ivie;	\mathbb{R}^4 is 4-F-Ph;
<u>R</u> 3	<u>R</u> 3	<u>R</u> 3
CN	CN	CN
NO_2	NO_2	NO_2
NMe ₂	NMe ₂	NMe_2
NHCO ₂ Me	NHCO ₂ Me	NHCO ₂ Me
epoxyethyl	epoxyethyl	epoxyethyl
- , .	- •	

 \mathbb{R}^2 is 5-F; \mathbb{R}^4 is 4-F-Ph;

 \mathbb{R}^2 is 5-Cl; R⁴ is Me;

 \mathbb{R}^2 is 5-Cl; R^4 is 4-F-Ph;

 \mathbb{R}^3 SiMe₃ SMe $P(O)(OMe)_2$ CH₂SMe CH2OCH2CF3 CH₂SiMe₃ 2-pyridyl 2-thienyl

 \mathbb{R}^3 SiMea SMe $P(O)(OMe)_2$ CH₂SMe CH2OCH2CF3

CH₂SiMe₃

2-pyridyl

2-thienyl

 \mathbb{R}^3 SiMeg SMe $P(O)(OMe)_2$ CH₂SMe CH₂OCH₂CF₃ CH₂SiMe₃

2-pyridyl

2-thienyl

 \mathbb{R}^2 is 5-Cl; \mathbb{R}^4 is H;

 \mathbb{R}^2 is 5-Cl; \mathbb{R}^4 is Ph;

 \mathbb{R}^3

CN

 \mathbb{R}^3 CN NO_2 NMe₂ NHCO₂Me epoxyethyl SiMe3 SMe $P(O)(OMe)_2$ CH₂SMe

CH2OCH2CF3 CH₂SiMe₃

2-pyridyl

2-thienyl

NMe₂ NHCO₂Me epoxyethyl SiMe₃ SMe $P(O)(OMe)_2$ CH₂SMe CH₂SiMe₃ 2-pyridyl 2-thienyl

 NO_2 $CH_2OCH_2CF_3$

5

10

KEY FOR TABLE 8

Group	<u>R</u> 1	\mathbb{R}^2
а	CF ₃	4-Cl
b	OCF3	4-Cl
c	CF ₃	4-F
d	OCF3	4-F

Compounds of Table 8 wherein \mathbb{R}^3 and \mathbb{R}^4 are as set out therein can be prepared having the recited values of Groups a through d. That is, for each value of \mathbb{R}^3 and \mathbb{R}^4 in the Table, \mathbb{R}^1 and \mathbb{R}^2 can have the values recited in Groups a through d above. All of said compounds are of this invention.

15 <u>TABLE 8</u>

This Table contains a large number of compounds of Formula I in a format adopted to avoid mechanical reproduction of substituent values that do not vary. For example, the first table entry (where \mathbb{R}^4 is 4-F and \mathbb{R}^3 is CN) actually specifies 4 separate and distinct compounds because

the key for Table 8 identifies additional substituent values of \mathbb{R}^1 and \mathbb{R}^2 , namely Groups a through d for each Table 8 entry.

CH₂SiMe₃ 2-pyridyl

2-thienyl

R ⁴ is H;	R ⁴ is Ph;
\mathbb{R}^3	\mathbb{R}^3
CN	CN
NO_2	NO_2
NMe ₂	NMe ₂
NHCO ₂ Me	NHCO ₂ Me
epoxyethyl	epoxyethyl
SiMe ₃	SiMe3
SMe	SMe
$P(O)(OMe)_2$	$P(O)(OMe)_2$
CH ₂ SMe	CH ₂ SMe
CH ₂ OCH ₂ CF ₃	CH ₂ OCH ₂ CF ₃
CH ₂ SiMe ₃	CH ₂ SiMe ₃
2-pyridyl	2-pyridyl
2-thienyl	2-thienyl
R ⁴ is Me;	R ⁴ is 4-F-Ph;
R 3	R ³
CN	CN
NO_2	NO_2
NMe_2	NMe ₂
NHCO ₂ Me	NHCO ₂ Me
epoxyethyl	epoxyethyl
SiMe ₃	$SiMe_3$
SMe	SMe
$P(O)(OMe)_2$	$P(O)(OMe)_2$
CH ₂ SMe	CH ₂ SMe
CH ₂ OCH ₂ CF ₃	$CH_2OCH_2CF_3$
CTT C:3.5	ATT A:> E

CH₂SiMe₃

2-pyridyl 2-thienyl

KEY FOR TABLE 9

Group	<u>R</u> 1	\mathbb{R}^2
a	CF ₃	4-C1
b	OCF3	4-Cl
c	CF3	4-F
ď	OCF3	4-F

Compounds of Table 9 wherein R^3 and R^4 are as set out therein can be prepared having the recited values of Groups a through d. That is, for each value of R^3 and R^4 in the Table, R^1 and R^2 can have the values recited in Groups a through d above. All of said compounds are of this invention.

15

20

10

TABLE 9

This Table contains a large number of compounds of Formula I in a format adopted to avoid mechanical reproduction of substituent values that do not vary. For example, the first table entry (where \mathbb{R}^4 is H and \mathbb{R}^3 is CN) actually specifies 4 separate and distinct compounds because

the key for Table 9 identifies additional substituent values of \mathbb{R}^1 and \mathbb{R}^2 , namely Groups a through d for each Table 9 entry.

 \mathbb{R}^4 is H;

 \mathbb{R}^4 is Ph;

<u>R</u>3

CN

 NO_2

NMe₂

R³
CN
NO₂
NMe₂
NHCO₂Me
epoxyethyl
SiMe₃
SMe
P(O)(OMe)₂

epoxyethyl SiMe3 SMe P(O)(OMe)2

 CH_2SMe

NHCO₂Me

CH₂SMe CH₂OCH₂CF₃ CH₂SiMe₃ 2-pyridyl 2-thienyl

CH₂OCH₂CF₃ CH₂SiMe₃ 2-pyridyl 2-thienyl

R4 is Me;

 \mathbb{R}^4 is 4-F-Ph;

R³
CN
NO₂
NMe₂
NHCO₂Me
epoxyethyl
SiMe₃
SMe
P(O)(OMe)₂

R3 CN NO2 NMe2 NHCO2Me epoxyethyl SiMe3 SMe P(O)(OMe)2

CH₂SMe CH₂OCH₂CF₃ CH₂SiMe₃ 2-pyridyl 2-thienyl

CH₂SMe CH₂OCH₂CF₃ CH₂SiMe₃ 2-pyridyl 2-thienyl

KEY FOR TABLE 10

Group	<u>R</u> 1	$\underline{\mathbf{v}}$
а	CF ₃	\mathtt{CH}_2
Ъ	OCF_3	CH_2
С	Cl	CH_2
d	Br	$\mathtt{CH_2}$
е	$\mathbf{CF_3}$	0
${f f}$	OCF ₃	0
g	Cl	O
h	Br	0
i	$\mathbf{CF_3}$	NH
j	OCF_3	NH
k	CF ₃	NMe
1	OCF ₃	NMe

10

15

Compounds of Table 10 wherein R^2 , R^{19} , R^{32} and q are as set out therein can be prepared having the recited values of Groups a through l. That is, for each value of R^2 , R^{19} , R^{32} and q in the Table, R^1 and V can have the values recited in Groups a through l above. Each compound having the recited values for , R^1 , V, R^2 , R^{19} , R^{32} and q is included within the scope of this invention by the specific designation of its structure.

TABLE 10

This Table contains a large number of compounds of Formula I in a format adopted to avoid mechanical reproduction of substituent values that do not vary. For example, the first table entry (where R^2 is 4-F, q is 2 and R^{19} and R^{32} are H) actually specifies 12 separate and distinct compounds because the key for Table 10 identifies additional substituent values of R^1 and V, namely Groups a through 1 for each Table 10 entry.

R ² is 4-F q is 2;		R^2 is 5-CF ₃ q is 2;	
R ¹⁹	<u>R</u> 32	<u>R</u> 19	\mathbb{R}^{32}
H	H	н	н
H	Me	H	Me
H	CO ₂ Me	H	CO_2Me
Me	H	Me	H
iPr	H	iPr	H
Ph	H	Ph	H
CO ₂ Me	н	CO ₂ Me	н

$ m R^2$ is 5-F q is 2;		$ m R^2$ is 4-F q is 3;	
4.	,	7	10 0,
R ¹⁹	<u>R</u> 32	R ¹⁹	R ³²
н	H	н	H
H ·	Me	H	Me
H	$\rm CO_2Me$	H	CO ₂ Me
Me	H	ME	H
iPr	H	i Pr	H
Ph	H	Ph	H
${ m CO_2Me}$	H	${ m CO_2Me}$	H

R ² is 5-Cl q is 2;		$ m R^2$ is 5-F q is 3;		
<u>R</u> 19	<u>R</u> 32	<u>R</u> 19	<u>R</u> 32	
н	H	н	н	
H	Me	H	Me	
H	${ m CO_2Me}$	H	CO_2Me	
Me	H	Me	H	
iPr	H	iPr	H	
Ph	H	Ph	H	
CO_2Me	H	$_{ m CO_2Me}$	H	

20 20 01	\mathbb{R}^2 is 5-F q is 4;		
R ¹⁹ R ³² R ¹⁹	<u>R</u> 32		
н н н	H		
H Me H	Me		
H CO ₂ Me H	CO ₂ Me		
Me H Me	H		
iPr H iPr	H		
Ph H Ph	H		
CO ₂ Me H CO ₂ Me	H		

$ m R^2$ is 5-CF3 q is 3;		R^2 is 5-Cl q is 4;	
R ¹⁹	<u>R</u> 32	R ¹⁹	<u>R</u> 32
н	н	H	н
H	Me	H	Me
H	${ m CO_2Me}$	H	CO ₂ Me
Me	H	Me	H
i Pr	H	iPr	H
Ph	H	\mathbf{Ph}	H
CO ₂ Me	H	$_{ m CO_2Me}$	H

$ m R^2$ is 4-Fl q is 4;		R^2 is CF_3 q is 4;		
<u>R</u> 19	<u>R</u> 32	<u>R</u> 19	R ³²	
H	H	н	н	
H	Me	H	Me	
H	${ m CO_2Me}$	H	CO_2Me	
Me	H	Me	H	
iPr	H	i Pr	H	
Ph	H	Ph	H	
CO_2Me	\mathbf{H}	CO_2Me	H	

Ē

KEY FOR TABLE 11

Group	\mathbb{R}^1
a b	$ ext{CF}_3$ $ ext{OCF}_3$
c	Cl
d	Br

10

15

20

Compounds of Table 11 wherein R^2 , R^{21} and q are as set out therein can be prepared having the recited values of Groups a through d. That is, for each value of R^2 , R^{21} , and q in the Table, R^1 can have the values recited in Groups a through d above. All of said compounds are of this invention.

TABLE 11

This Table contains a large number of compounds of Formula I in a format adopted to avoid mechanical reproduction of substituent values that do not vary. For example, the first table entry (where R² is 4-F, q is 3 and R²¹ is H) actually specifies 4 separate and distinct compounds because the key for Table 11 identifies additional substituent values of R¹, namely Groups a through d for each Table 11 entry.

q is 3; q i	s 3; q is 4;
q.50, q.	
<u>R</u> ²¹ R	21 <u>R</u> 21
H	н н
Me N	Me Me
iPr i	Pr iPr
Ph F	Ph Ph
4-F-Ph 4-F	F-Ph 4-F-Ph
R^2 is 5-F R^2 is	5-CF ₃ R ² is 5-F
q is 3;	s 3; q is 4;
\mathbb{R}^{21} R	21 <u>R</u> 21
	H H
Me N	Me Me
iPr i	Pr iPr
Ph F	Ph Ph
4-F-Ph 4-F	r-Ph 4-F-Ph

\mathbb{R}^2 is 5-Cl;	\mathbb{R}^2 is 4-F;	\mathbb{R}^2 is 5-Cl;
q is 4;	q is 5;	q is 5;
\mathbb{R}^{21}	<u>R</u> 21	\mathbb{R}^{21}
н	Н	Н
Me	Me	Me
i Pr	iPr	i Pr
Ph	Ph	\mathbf{Ph}
4-F-Ph	4-F-Ph	4-F-Ph
\mathbb{R}^2 is 5-CF3	\mathbb{R}^2 is 5-F;	\mathbb{R}^2 is 5-CF3;
q is 4;	q is 5;	q is 5;
\mathbb{R}^{21}	\mathbb{R}^{21}	\mathbb{R}^{21}
H	H	H
Me	Me	Me
iPr	iPr	iP r
Ph	Ph	Ph
4-F-Ph	4-F-Ph	4-F-Ph

KEY FOR TABLE 12

X Y || | Q-C-N-G

1

10	Table	Group	Q
	12 ^(a)	а	Q-1 (A=CH ₂ , X=O)
	12 ^(a)	b	$Q-1 (A=CH_2CH_2, X=O)$
	12 ^(a)	c	Q-1 (A=O, X=O)
15	12(a)	d	$Q-1 (A=OCH_2, X=O)$
	12 ^(a)	e	Q-1 (A=S, X=O)
	12(a)	${f f}$	$Q-1 (A=SCH_2, X=O)$
	12 ^(a)	g	Q-1 (A=NH, X=O)
	12 ^(a)	h	Q-1 (A=NHCH ₂ , X=O)
20	12 ^(a)	i	Q-1 ($A=N(Me)CH_2, X=O$)
	12(a)	j	Q-1 (A=NMe, X=O)
	12 ^(a)	k	$Q-1 (A=NCH_2Ph, X=O)$
	12 ^(a)	1	Q-1 (A=N(CH ₂ Ph)CH ₂ , X=0)
	12 ^(a)	m	$Q-1 (A=OCH_2, X=S)$
2 5	12 ^(a)	n	$Q-1 (A=CH_2CH_2, X=S)$
	12 ^(a)	o	$Q-2 (A=CH_2, X=O)$
	12 ^(a)	p	Q-2 (A=CH ₂ CH ₂ , X=O)
	12 ^(a)	q	Q-2 (A=O, X=O)
	12 ^(a)	r	Q-2 (A=OCH ₂ , X=O)
30	12 ^(a)	8	Q-2 (A=S, X=O)
	12 ^(a)	t	Q-2 (A=SCH ₂ , X=O)
	12 ^(a)	u	Q-2 (A=NH, X=O)

5	<u>Table</u>	Group	Q
	12 ^(a)	v	Q-2 (A=NHCH ₂ , X=O)
	12 ^(a)	w	$Q-2 (A=N(Me)CH_2, X=O)$
	12 ^(a)	x	Q-2 (A=NMe, X=O)
10	12 ^(a)	у	$Q-2 (A=NHC_2Ph, X=O)$
	12 ^(a)	z	Q-2 (A=N(CH2Ph)CH2, X=O)
	12 ^(a)	aa	$Q-2 (A=OCH_2, X=S)$
	12 ^(a)	ab	$Q-2 (A=CH_2CH_2, X=S)$
	12 ^(a)	ac	Q-2 (A=CHMe, X=O)
15	12 ^(a)	ad	$Q-2 (A=CH(Me)CH_2, X=O)$
	12 ^(a)	ae	Q-2 (A=CH ₂ CHCl, X=O)
	12 ^(a)	af	Q-1 (A=CHMe, X=O)
	12(a)	ag	$Q-1 (A=CH(Me)CH_2, X=O)$
	12 ^(a)	ah	Q-1 (A=CH ₂ CHMe, X=O)
20	12 ^(a)	ai	Q-1 (A=CH ₂ CHCl, X=O)
	12 ^(a)	aj	Q-2 (A=CH ₂ CHMe, X=O)
	12 ^(a)	ak	Q-3 (A=CH ₂ , R^4 =H, X=O, Y^1 =H)
	12 ^(a)	al	Q-3 (A=CH ₂ CH ₂ , R ⁴ =H, X=O, Y ¹ =H)
	12(a)	am	Q-3 (A=CHMe, R^4 =H, X=O, Y^1 =H)
2 5	12 ^(a)	an	Q-3 (A=CH ₂ CHMe, R^4 =H, X=O, Y ¹ =H)
	12 ^(a)	ao	Q-3 (A=CH(Me)CH ₃ , R^4 =H, X=O, Y^1 =H)
	12 ^(a)	ар	Q-3 (A=O, R^4 =H, X=O, Y^1 =H)
	12 ^(a)	aq	Q-3 (A=OCH ₂ , R ⁴ =H, X=O, Y ¹ =H)
	12 ^(a)	ar	Q-3 (A=S, R^4 =H, X=O, Y^1 =H)
30	12 ^(a)	as	Q-3 (A=SCH ₂ , R ⁴ =H, X=O, Y ¹ =H)
	12 ^(a)	at	Q-3 (A=NH, R^4 =H, X=O, Y^1 =H)
	12 ^(a)	au	Q-3 (A=NMe, R^4 =H, X=O, Y^1 =H)
	12 ^(a)	av	Q-3 (A=NHCH ₂ , R^4 =H, X=O, Y^1 =H)
	12 ^(a)	aw	Q-3 (A=N(Me)CH ₂ , R ⁴ =H, X=O, Y ¹ =H)

5	<u>Table</u>	Group	Q
	12 ^(a)	ax	Q-3 (A=NCH ₂ Ph, R^4 =H, X=O, Y^1 =H)
	12 ^(a)	ay	Q-3 (A=N(CH ₂ Ph)CH ₂ , R^4 =H, X=O, Y^1 =H)
	12 ^(a)	az	Q-3 (A=CH ₂ , R^4 =Me, X=O, Y^1 =H)
10	12(a)	ba	Q-3 (A=O, R^4 =Me, X=O, Y^1 =H)
	12 ^(a)	bb	Q-3 (A=O, R^4 =H, X=S, Y^1 =H)
	12 ^(a)	bc	Q-3 (A=CH ₂ , R ⁴ =H, X=S, Y ¹ =H)
	12(a)	bd	Q-3 (A=O, R^4 =H, X=O, Y^1 =Me)
	12(a)	be	Q-3 (A=CH ₂ , R^4 =H, X=O, Y^1 =Me)
15	12(a)	bf	Q-3 (A=O, R^4 =Me, X=O, Y^1 =Me)
	12(a)	bg	Q-3 (A=CH ₂ , R^4 =Me, X=O, Y^1 =Me)
	12 ^(a)	bh	$Q-4 (R^4=H, X=O)$
	12(a)	bi	$Q-4 (R^4=H, X=S)$
	12(a)	bj	Q-4 (R4=Me, X=O)
20	12 ^(a)	bk	Q-4 (R4=Et, X=O)
	12 ^(a)	bl	$Q-5 (R^{18}=H, V=CH_2, X=O)$
	12(a)	\mathbf{bm}	$Q-5 (R^{18}=H, V=O, X=O)$
	12 ^(a)	bx	$Q-5 (R^{18}=CO_2Me, V=NMe, X=O)$
	12(a)	by	Q-5 (R^{18} = SO_2Me , V = CH_2 , X = O)
25	12(a)	bz	Q-5 (R^{18} = SO_2Me , V=O, X=O)
	12(a)	ca	Q-5 (R^{18} =SO ₂ Me, V=NMe, X=O)
	12 ^(a)	cb	Q-5 (R^{18} =Me, V=CH ₂ , X=O)
	12 ^(a)	cc	$Q-5 (R^{18}=Me, V=O, X=O)$
	12(a)	cd	Q-5 (R ¹⁸ =Me, V=NMe, X=O)
30	12 ^(a)	ce	$Q-6 (R^{18}=H, V=CH_2, X=O)$
	12 ^(a)	cf	$Q-6 (R^{18}=H, V=O, X=O)$
	12(a)	cg	Q-6 (R ¹⁸ =H, V=NH, X=O)
	12 ^(a)	ch	Q-6 (R ¹⁸ =H, V=NMe, X=O)
	12 ^(a)	ci	$Q-6 (R^{18}=H, V=NCH_2Ph, X=O)$
35	12 ^(a)	cj	$Q-6 (R^{18}=H, V=S, X=O)$

5	<u>Table</u>	Group	Q
	12 ^(a)	ck	Q-6 (R ¹⁸ =H, V=CHCH ₃ , X=O)
	12 ^(a)	cl	Q-6 (R ¹⁸ =COMe, V=CH ₂ , X=O)
	12(a)		Q-6 (R ¹⁸ =COMe, V=O, X=O)
10	12(a)	cm cn	Q-6 (R ¹⁸ =COMe, V=NMe, X=O)
10	12(a)	co	$Q-6 (R^{18}=CO_2Me, V=CH_2, X=O)$
	12(a)	. cp	Q-6 (R^{18} = CO_2 Me, V=O, X=O)
	12(a)	cq	Q-6 (R^{18} = CO_2Me , V= NMe , X= O)
	12(a)	cr	Q-6 (R^{18} = CO_2 Me, V=S, X=O)
15	12 ^(a)	cs	Q-5 (R ¹⁸ =COMe, V=S, X=O)
10	12(a)	ct	Q-5 (R^{18} = CO_2Me , V=S, X=O)
	12 ^(a)	cu	Q-5 (R^{18} = SO_2Me , V= S , X= O)
	12(a)	cv	Q-5 (R^{18} =Me, V=S, X=O)
	12(a)	cw	Q-6 (R ¹⁸ =COMe, V=S, X=O)
20	12(a)	CX.	Q-6 (R^{18} = SO_2Me , V= CH_2 , X= O)
_,	12(a)	cy	Q-6 (R^{18} = SO_2Me , V=O, X=O)
	12(a)	cz	Q-6 (R^{18} =SO ₂ Me, V=NMe, X=O)
	12(a)	da	$Q-6 (R^{18}=SO_2Me, V=S, X=O)$
	12 ^(a)	db	Q-6 (R^{18} =Me, V=CH ₂ , X=O)
25	12 ^(a)	dc	Q-6 (R ¹⁸ =Me, V=O, X=O)
	12 ^(a)	dd	$Q-6 (R^{18}=Me, V=NMe, X=O)$
	12 ^(a)	de	Q-6 (R^{18} =Me, V=S, X=O)
	12 ^(a)	df	Q-7 (A=CH ₂ , R ⁴ =H, R ¹⁸ =H, X=O)
	12 ^(a)	dg	$Q-7 (A=0, R^4=H, R^{18}=H, X=0)$
30	12(a)	dh	Q-7 (A=CHCH ₃ , R^4 =H, R^{18} =H, X=O)
	12 ^(a)	di	$Q-7 (A=CH_2CH_2, R^4=H, R^{18}=H, X=O)$
	12 ^(a)	dj	Q-7 (A=OCH ₂ , R^4 =H, R^{18} =H, X=O)
	12 ^(a)	dk	Q-7 (A=CH(Me)CH ₂ , R^4 =H, R^{18} =H, X=O)
	12 ^(a)	dl	Q-7 (A=CH ₂ CH(Me), R^4 =H, R^{18} =H, X=O)

5	' <u>Table</u>	Group	Q
	12 ^(a)	dm	Q-7 (A=S, R ⁴ =H, R ¹⁸ =H, X=O)
	12(a) 12(a)	dn	Q-7 (A=SCH ₂ , R ⁴ =H, R ¹⁸ =H, X=O)
	12(a)		Q-7 (A=NH, R^4 =H, R^{18} =H, X=O)
10	12(a)	do	Q-7 (A=NMe, R^4 =H, R^{18} =H, X=O)
10	12(a)	dp dq	Q-7 (A=NMe, R^{1} =H, R^{1} 8=H, X =O)
	12(a)		
		dr -	Q-7 (A=N(Me)CH ₂ , R ⁴ =H, R ¹⁸ =H, X=O)
	12(a)	ds	Q-7 (A=NCH ₂ Ph, R ⁴ =H, R ¹⁸ =H, X=O)
	12 ^(a)	\mathbf{dt}	Q-7 (A=N(CH ₂ Ph)CH ₂ , R^4 =H, R^{18} =H, X=O)
15	12(a)	du	Q-7 (A=CH ₂ , R^4 =Me, R^{18} =H, X=O)
	12 ^(a)	$\mathbf{d}\mathbf{v}$	Q-7 (A=O, R^4 =Me, R^{18} =H, X=O)
	12 ^(a)	$\mathbf{d}\mathbf{w}$	Q-7 (A=CH ₂ , R^4 =Me, R^{18} =Me, X=O)
	12(a)	dx	Q-7 (A=O, R^4 =Me, R^{18} =Me, X=O)
	12(a)	dy	Q-7 (A=CH ₂ , R^4 =Me, R^{18} =COMe, X=O)
20	12 ^(a)	dz	Q-7 (A=O, R^4 =Me, R^{18} =COMe, X=O)
	12 ^(a)	ea	Q-7 (A=CH ₂ , R^4 =Me, R^{18} =CO ₂ Me, X=O)
	12 ^(a)	eb	Q-7 (A=O, R^4 =Me, R^{18} =CO ₂ Me, X=O)
	12 ^(a)	ec ·	Q-7 (A=CH ₂ , R^4 =Me, R^{18} =SO ₂ Me, X=O)
	12 ^(a)	ed	Q-7 (A=O, R^4 =Me, R^{18} =SO ₂ Me, X=O)
25	12 ^(a)	ee	Q-8 (R^4 =H, R^{18} =H, X=O)
	12 ^(a)	ef	$Q-8 (R^4=H, R^{18}=H, X=S)$
	12 ^(a)	eg	Q-8 (R^4 =Me, R^{18} =H, X=O)
	12 ^(a)	eh	Q-8 (R^4 =Et, R^{18} =H, X=O)
	12(a)	ei	$Q-8 (R^4=H, R^{18}=Me, X=O)$
30	12(a)	ej	Q-8 (R^4 =H, R^{18} =COMe, X=O)
	12 ^(a)	ek	Q-8 (R^4 =H, R^{18} = CO_2 Me, X=O)
	12 ^(a)	el	Q-8 (R^4 =H, R^{18} =SO ₂ Me, X=O)
	12(a)	em	Q-8 (R^4 =Me, R^{18} =Me, X=O)
	12(a)	en	Q-8 (R^4 =Me, R^{18} =COMe, X=O)
35	12 ^(a)	eo	Q-8 (R^4 =Me, R^{18} =CO ₂ Me, X=O)

5	<u>Table</u>	Group	Q
	12 ^(a)	ер	Q-8 (R^4 =Me, R^{18} =SO ₂ Me, X=O)
	12 ^(a)	eq	Q-8 (R^4 =Et, R^{18} =Me, X=O)
	12(a)	er	Q-8 (R ⁴ =Et, R ¹⁸ =COMe, X=O)
10	12(a)	es	Q-8 (R^4 =Et, R^{18} =CO ₂ Me, X=O)
	12 ^(a)	et	Q-8 (R^4 =Et, R^{18} =SO ₂ Me, X=O)
	12(a)	eu	$Q-9 (R^4=H, R^{18}=H, X=O)$
	12 ^(a)	ev	$Q-9 (R^4=H, R^{18}=H, X=S)$
	12(a)	ew	$Q-9 (R^4=Me, R^{18}=H, X=O)$
15	12 ^(a)	ex	Q-9 (R^4 =Et, R^{18} =H, X=O)
	12(a)	ey	$Q-9 (R^4=H, R^{18}=Me, X=O)$
	12 ^(a)	ez	Q-9 (R^4 =H, R^{18} =COMe, X=O)
	12(a)	fb	Q-9 (R^4 =H, R^{18} =SO ₂ Me, X=O)
	12(a)	fc	Q-9 (R^4 =Me, R^{18} =Me, X=O)
20	12 ^(a)	fd	Q-9 (R^4 =Me, R^{18} =COMe, X=O)
	12 ^(a)	fe	Q-9 (R^4 =Me, R^{18} =CO ₂ Me, X=O)
	12 ^(a)	ff	Q-9 (R^4 =Me, R^{18} =SO ₂ Me, X=O)
	12 ^(a)	fg	Q-9 (R^4 =Et, R^{18} =SO ₂ Me, X=O)
	12 ^(a)	fh	Q-9 (R^4 =Et, R^{18} =Me, X=O)
2 5	12 ^(a)	fi	Q-9 (R^4 =Et, R^{18} =COMe, X=O)
	12(a)	fj	Q-9 (R^4 =Et, R^{18} = CO_2 Me, X=O)

KEY FOR TABLE 13



Table Q Group 10 13(b) Q-1 ($A=CH_2$) а 13(b) b $Q-1 (A=CH_2CH_2)$ 13(þ) Q-1 (A=CHMe) c 13(b) $Q-1 (A=CH(Me)CH_2)$ d 13(b) $Q-1 (A=CH_2CHCI)$ 15 e 13(b) f Q-1 (A=CH₂CHMe) 13(b) Q-1 (A=O) g 13(b) h $Q-1 (A=OCH_2)$ 13(b) i Q-1 (A=S)13(b) 20 $Q-1 (A=SCH_2)$ j 13(b) k Q-1 (A=NH) 13(b) 1 Q-1 (A=NMe) 13(b) $Q-1 (A=N(Me)CH_2)$ \mathbf{m} 13(b) Q-1 ($A=NHCH_2$) \mathbf{n} 13(b) 25 $Q-1 (A=NCH_2Ph)$ 0 13(b) $Q-1 (A=N(CH_2Ph)CH_2)$ p 13(b) $Q-2 (A=CH_2)$ q 13(b) $Q-2 (A=CH_2CH_2)$ 13(b) Q-2 (A=CHMe) 13(b) 30 t Q-2 (A=CH(Me)CH₂) 13(b) Q-2 (A=CH₂CHCl) 13(b) Q-2 (A=CH₂CHMe) \mathbf{v} 13(b) w Q-2 (A=0)

5	<u>Table</u>	Group	Q
	13(b)	x	Q-2 (A=OCH ₂)
	13(b)	у	Q-2 (A=S)
	13 ^(b)	z	Q-2 (A=SCH ₂)
10	13 ^(b)	aa	Q-2 (A=NH)
	13(b)	ab	Q-2 (A=NMe)
	13 ^(b)	ac	$Q-2 (A=N(Me)CH_2)$
	13 ^(b)	ad	Q-2 (A=NHCH ₂)
	13(b)	ae	Q-2 (A=NCHPh)
15	13(b)	af	$Q-2 (A=N(CH_2Ph)CH_2)$
	13(b)	ag	Q-3 (A=CH(Me)CH ₂ , R^4 =H, Y^1 =H)
	13(b)	ah	Q-3 (A=CH ₂ CH ₂ , R ⁴ =H, Y ¹ =H)
	13(b)	ai	Q-3 (A=CHMe, R^4 =H, Y^1 =H)
	13(b)	aj	Q-3 (A=CH ₂ CHMe, R^4 =H, Y^1 =H)
20	13 ^(b)	ak	Q-3 (A=CH(Me)CH ₂ , R^4 =H, Y^1 =H)
	13(b)	al	Q-3 (A=O, R^4 =H, Y^1 =H)
	13 ^(b)	am	Q-3 (A=OCH ₂ , R^4 =H, Y^1 =H)
	13(b)	an	Q-3 (A=S, R^4 =H, Y^1 =H)
	13 ^(b)	ао	Q-3 ($A=SCH_2$, $R^4=H$, $Y^1=H$)
2 5	13(b)	ap	Q-3 (A=NH, R^4 =H, Y^1 =H)
	13(b)	aq	Q-3 (A=NMe, R^4 =H, Y^1 =H)
	13(b)	ar	Q-3 (A=NHCH ₂ , R^4 =H, Y^1 =H)
	13 ^(b)	as	Q-3 (A=N(Me)CH ₂ , R^4 =H, Y^1 =H)
	13 ^(b)	at	Q-3 (A=NCH ₂ Ph, R^4 =H, Y^1 =H)
30	13 ^(b)	au	Q-3 (A=N(CH ₂ Ph)CH ₂ , R ⁴ =H, Y ¹ =H)
	13 ^(b)	av	Q-3 (A=CH ₂ , R^4 =Me, Y^1 =H)
	13(b)	aw	Q-3 (A=O, R^4 =Me, Y^1 =H)
	13(b)	ax	Q-3 (A=CH ₂ , R ⁴ =H, Y ¹ =Me)
	13(b)	ay	Q-3 (A=O, R^4 =H, Y^1 =Me)
3 5	13 ^(b)	az	Q-3 (A=CH ₂ , R ⁴ =Me, Y ¹ =Me)

5	Table	Group	Q
	13(b)	ba	Q-3 (A=O, R^4 =Me, Y^1 =Me)
	13 ^(b)	bb	Q-4 (R4=H)
	13(b)	bc	$Q-4 (R^4=Me)$
10	13(b)	bd	$Q-4 (R^4=Et)$
	13(b)	be	Q-5 (R^{18} =H, V = CH_2)
	13(b)	bf	$Q-5 (R^{18}=H, V=O)$
	13(b)	bg	Q-5 (R ¹⁸ =H, V=NH)
	13(b)	bh	Q-5 (R ¹⁸ =H, V=NMe)
15	13 ^(b)	bi	$Q-5 (R^{18}=H, V=NCH_2Ph)$
	13(p)	bj	Q-5 (R ¹⁸ =H, V=S)
	13 ^(b)	bk	Q-5 (R^{18} =H, V=CHCH ₃)
	13(b)	bl	Q-5 (R^{18} =COMe, V =CH ₂)
	13(b)	\mathbf{bm}	Q-5 (R^{18} =COMe, V=O)
20	13(b)	bn	Q-5 (R^{18} =COMe, V=NMe)
	13 ^(b)	bo	Q-5 (R^{18} = CO_2Me , V = CH_2)
	13(b)	bp	$Q-5 (R^{18}=CO_2Me, V=O)$
	13(b)	bq	Q-5 (R ¹⁸ =CO ₂ Me, V=NMe)
	13(b)	br	Q-5 (R^{18} = SO_2Me , V = CH_2)
25	13(b)	bs	$Q-5 (R^{18}-SO_2Me, V=0)$
	13 ^(b)	bt	Q-5 (R^{18} =SO ₂ Me, V=NMe)
	13 ^(b)	bu	Q-5 (R^{18} =Me, V =CH ₂)
	13(b)	bv	$Q-5 (R^{18}=Me, V=O)$
	13 ^(b)	bw	Q-5 (R ¹⁸ =Me, V=NMe)
30	13 ^(b)	bx	Q-6 (R^{18} =H, V = CH_2)
	13(b)	by	Q-6 ($R^{18}=H, V=O$)
	13(b)	bz	$Q-6 (R^{18}=H, V=NH)$
	13(b)	ca	Q-6 (R ¹⁸ =H, V=NMe)
	13(b)	cb	Q-6 (R^{18} =H, V =NCH ₂ Ph)
35	13 ^(b)	cc	Q-6 (R^{18} =H, V =S)

5	<u>Table</u>	Group	Q
	13(b)	cd	Q-6 (R^{18} =H, V=CHCH ₃)
	13(b)	ce	Q-6 (R^{18} =COMe, V=CH ₂)
	13(b)	cf	Q-6 (R ¹⁸ =COMe, V=O)
10	13 ^(b)	cg	Q-6 (R ¹⁸ =COMe, V=NMe)
	13 ^(b)	ch	$Q-6 (R^{18}=CO_2Me, V=CH_2)$
	13(b)	ci	Q-6 (R^{18} = CO_2Me , V=O)
	13(b)	cj	Q-6 (R^{18} = CO_2 Me, V=NMe)
	13(b)	ck	Q-6 (R^{18} = CO_2Me , V= S)
15	13(b)	cl	Q-6 (R ¹⁸ =COMe, V=S)
	13(b)	cm	$Q-6 (R^{18}=SO_2Me, V=CH_2)$
	13 ^(b)	cn	Q-6 (R^{18} = SO_2Me , V=O)
	13 ^(b)	со	Q-6 ($R^{18}=SO_2Me, V=S$)
	13 ^(b)	ср	Q-6 (R^{18} =SO ₂ Me, V=NMe)
20	13 ^(b)	cq	Q-6 (R^{18} =Me, V =CH ₂)
	13(b)	cr	Q-6 (R^{18} =Me, V=O)
	13(b)	cs	$Q-6 (R^{18}=Me, V=S)$
	13(b)	ct	Q-6 (R^{18} =Me, V=NMe)
	13(b)	cu	$Q-5$ (R^{18} =COMe, $V=S$)
25	13(b)	cv	Q-5 (R^{18} = CO_2Me , V = S)
	13 ^(b)	cw	Q-5 (R18=SO2Me, V=S)
	13 ^(b)	сж	$Q-5 (R^{18}=Me, V=S)$
	13 ^(b)	сy	Q-7 (A=CH ₂ , R^4 =H, R^{18} =H)
	13 ^(b)	cz	Q-7 (A=O, R^4 =H, R^{18} =H)
30	13(b)	da	Q-7 (A=CH ₂ CH ₂ , R^4 =H, R^{18} =H)
	13 ^(b)	d b	Q-7 (A=CHMe, R^4 =H, R^{18} =H)
	13(b)	dc	Q-7 (A=O, R^4 =Et, R^{18} =H)
	13(b)	dd	Q-7 (A=CH ₂ , R ⁴ =Et, R ¹⁸ =H)
	13 ^(b)	de	Q-7 (A=OCH ₂ , R^4 =H, R^{18} =H)
35	13(b)	df	Q-7 (A=CH(Me)CH ₂ , R^4 =H, R^{18} =H)

5	Table	Group	<u>Ω</u>
	-		
	13(b)	$\mathbf{d}\mathbf{g}$	Q-7 (A=CH ₂ CH(Me), R^4 =H, R^{18} =H)
	13 ^(b)	dh	Q-7 (A=S, R^4 =H, R^{18} =H)
	13(b)	di	Q-7 (A=SCH ₂ , R^4 =H, R^{18} =H)
10	13(b)	dj	Q-7 (A=NH, R^4 =H, R^{18} =H)
	13(b)	dk	Q-7 (A=NMe, R^4 =H, R^{18} =H)
	13 ^(b)	dl	Q-7 (A=NHCH ₂ , R^4 =H, R^{18} =H)
	13(b)	\mathbf{dm}	Q-7 (A=N(Me)CH ₂ , R^4 =H, R^{18} =H)
	13(b)	\mathbf{dn}	Q-7 (A=NCH ₂ Ph, R^4 =H, R^{18} =H)
15	13(b)	do	Q-7 (A=N(CH ₂ Ph)CH ₂ , R^4 =H, R^{18} =H)
	13(b)	$\mathbf{d}\mathbf{p}$	Q-7 (A=CH ₂ , R^4 =Me, R^{18} =H)
	13(b)	dq	Q-7 (A=O, R^4 =Me, R^{18} =H)
	13(b)	dr	Q-7 (A=CH ₂ , R^4 =Me, R^{18} =Me)
	13(b)	ds	Q-7 (A=O, R^4 =Me, R^{18} =Me)
20	13(b)	dt	Q-7 (A=CH ₂ , R^4 =Me, R^{18} =COMe)
	13(b)	du	Q-7 (A=O, R^4 =Me, R^{18} =COMe)
	13(b)	$\mathbf{d}\mathbf{v}$	Q-7 (A=CH ₂ , R^4 =Me, R^{18} =CO ₂ Me)
	13 ^(b)	$\mathbf{d}\mathbf{w}$	Q-7 (A=O, R^4 =Me, R^{18} =CO ₂ Me)
	13 ^(b)	dx	Q-7 (A=CH ₂ , R^4 =Me, R^{18} =SO ₂ Me)
25	13(b)	dy	Q-7 (A=O, R^4 =Me, R^{18} =SO ₂ Me)
	13(b)	dz	$Q-8 (R^4=H, R^{18}=H)$
	13 ^(b)	ea	$Q-8 (R^4=Me, R^{18}=H)$
	13(b)	eb	Q-8 (R^4 =Et, R^{18} =H)
	13(b)	ec	$Q-8 (R^4=H, R^{18}=Me)$
30	13(b)	ed	Q-8 (R^4 =H, R^{18} =COMe)
	13(b)	ee	Q-8 (R^4 =H, R^{18} =CO ₂ Me)
	13(b)	ef	Q-8 (R^4 =H, R^{18} =SO ₂ Me)
	13(b)	eg	$Q-8 (R^4=Me, R^{18}=Me)$
	13(b)	eh	Q-8 (R^4 =Me, R^{18} =COMe)
35	13 ^(b)	ei	Q-8 (R^4 =Me, R^{18} = CO_2 Me)

5	<u>Table</u>	Group	Q
	13(b)	ej	Q-8 (R^4 =Me, R^{18} =SO ₂ Me)
	13(b)	ek	$Q-8 (R^4=Et, R^{18}=Me)$
	13(b)	el	Q-8 (R^4 =Et, R^{18} =COMe)
10	13(b)	\mathbf{em}	Q-8 (R^4 =Et, R^{18} = CO_2Me)
	13(b)	en	Q-8 (R^4 =Et, R^{18} =SO ₂ Me)
	13(b)	eo	$Q-9 (R^4=H, R^{18}=H)$
	13(b)	ер	Q-9 (R^4 =Me, R^{18} =H)
	13(b)	eq	Q-9 (R^4 =Et, R^{18} =H)
1 5	13(b)	er	Q-9 (R^4 =H, R^{18} =Me)
	13(b)	es	Q-9 (R^4 =H, R^{18} =COMe)
	13(b)	et	Q-9 (R^4 =H, R^{18} = CO_2 Me)
	13 ^(b)	eu	Q-9 (R^4 =H, R^{18} =SO ₂ Me)
	13(b)	ev	$Q-9 (R^4=Me, R^{18}=Me)$
20	13(b)	ew	$Q-9 (R^4=Me, R^{18}=COMe)$
	13 ^(b)	ex	Q-9 (R^4 =Me, R^{18} = CO_2 Me)
	13(b)	ey	Q-9 (R^4 =Me, R^{18} =SO ₂ Me)
	13(b)	ez	Q-9 (R^4 =Et, R^{18} =Me)
	13 ^(b)	fa	Q-9 (R^4 =Et, R^{18} =COMe)
2 5	13 ^(b)	fb	Q-9 (R^4 =Et, R^{18} = CO_2 Me)
	13(b)	fc	Q-9 (R^4 =Et, R^{18} =SO ₂ Me)

- (a) Compounds of Table 12 wherein G, R^1 , R_a^2 , R_b^2 ,
- R³ and Y are as set out therein can be prepared according to the teachings of this invention having the recited values of Groups a through fi in the Key for Table 12. Each of these compounds is specifically included within the scope of this invention.

5 (b) Compounds of Table 13 wherein G, R^1 , R_a^2 , R_b^2 ,

 \mathbb{R}^3 and \mathbb{X}^1 are as set out therein can be prepared according to the teachings of this invention having the recited values of Groups a through fc in the Key for Table 13. Each of these compounds is specifically included within the scope of this invention.

10

15

Compounds of Tables 12 and 13 wherein Q is as set out therein contain the following substitution patterns for R_a^2 , R_b^2 :

Q-1

Q-2

$$\begin{array}{c|c}
R_a^2 \\
\hline
R_b^2 \\
\hline
A R^3 \\
\hline
N-N \\
v^1
\end{array}$$

Q-3

$$R_a^2$$
 R_b^2
 R^3
 R^4

Q-4

$$R_a^2$$
 R_a^2
 R_a^2
 R_a^3
 R_a^{18}

Q-5

Q-7

Q-9

Each compound for which R_a^2 , R_b^2 , R^3 , R^4 , R^{18} , A, Y^1 and V are

recited is a compound of this invention which can be made according to the preparatory procedures disclosed herein. Each of such compounds is, therefore, a specific embodiment of this invention.

10 TABLE 12

15

This Table contains a large number of compounds of Formula I. A format has been adopted for their recitation that avoids the mechanical reproduction of substituent values that do not vary. For example, the first table entry (where G is G-1, R^1 is Cl and $R_a{}^2 = R_b{}^2 = R^3 = Y = H$) actually specifies 166 separate and distinct compounds because the Key for Table 12 identifies additional substituent values a through fi for each Table 12 entry.

$$R_a^2 = R_b^2 = R^3 = Y = H$$

		a	D	
20				
	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
	G-1	Cl	G-5	CN
	G-1	\mathbf{Br}	$\mathrm{CH_2Ph}$	Cl
2 5	G-1	CF ₃	$\mathrm{CH_2Ph}$	\mathbf{Br}
	G-1	OCF ₃	$\mathtt{CH_2Ph}$	CF_3
	G-1	CN	$\mathrm{CH_2Ph}$	OCF3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
3 0	G-2	CF ₃	G-7	Br
	G-2	OCF ₃	G-7	CF_3
•	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN

Cl

 \mathbf{Br}

CF₃ OCF3

G-5 G-5

G-5

G-5

$$R_a^2 = R_b^2 = Y = H, R^3 = Me$$

	<u>G</u>	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl .	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	${ m CH_2Ph}$	${f Br}$
	G-1	OCF3	$\mathrm{CH_2Ph}$	CF_3
	G-1	CN	CH_2Ph	OCF ₃
15	G-2	Cl	${ m CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	$\mathbf{CF_3}$
	G-2	CN	G-7	OCF_3
20	G-3	CI	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	Br
	G-3	OCF ₃	G-8	CF_3
	G-3	CN	G-8	OCF3
25	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	${\tt Br}$
	G-4	OCF ₃	G-10	CF_3
•	G-4	CN	G-10	OCF_3
30	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF3		
	G-5	OCF ₃		

$$R_a^2 = R_b^2 = Y = H, R^3 = Et$$

	<u>G</u>	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	C1
	G-1	CF3	$\mathrm{CH_2Ph}$	\mathbf{Br}
	G-1	OCF_3	$\mathtt{CH_2Ph}$	$\mathbf{CF_3}$
	G-1	CN	${ m CH_2Ph}$	OCF_3
15	G-2	Cl	$\mathtt{CH_2Ph}$	CN
	G-2	${f Br}$	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
	G-2	CN	G-7	OCF_3
20	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF_3	G-8	Br
	G-3	OCF_3	G-8	CF_3
	G-3	CN	G-8	OCF_3
2 5	G-4	Cl	G-8	CN
	G-4	${\tt Br}$	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
	G-4	OCF_3	G-10	$\mathbf{CF_3}$
	G-4	CN	G-10	OCF_3
30	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	$\mathbf{CF_3}$		
	G-5	OCF ₃		

$$R_a^2 = R_b^2 = Y = H, R^3 = i-Pr$$

	G	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
20	G-1	Br	CH_2Ph	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	\mathbf{Br}
	G-1	OCF ₃	$\mathrm{CH_2Ph}$	CF_3
	G-1	CN	$\mathrm{CH_2Ph}$	OCF ₃
15	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	${f Br}$
	G-2	OCF ₃	G-7	$\mathbf{CF_3}$
	G-2	CN	G-7	OCF ₃
20	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
	G-3	CN	G-8	OCF_3
25	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	$\mathbf{CF_3}$
	G-4	CN	G-10	OCF_3
30	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

$$R_a^2 = R_b^2 = Y = H, R^3 = CO_2Me$$

	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	C1
	G-1	CF ₃	$\mathtt{CH_2Ph}$	Br
	G-1	OCF ₃	$\mathrm{CH_2Ph}$	$\mathbf{CF_3}$
	G-1	CN	$\mathrm{CH_2Ph}$	OCF3
15	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	${\tt Br}$
	G-2	OCF_3	G-7	$\mathbf{CF_3}$
	G-2	CN	G-7	OCF_3
20	G-3	Cl	G-7	CN
	G-3	\mathbf{Br}	G-8	Cl
	G-3	CF ₃	G-8	Br
	G-3	OCF ₃	G-8	CF_3
	G-3	CN	G-8	OCF3
25	G-4	Cl	G-8	CN
	G-4	${f Br}$	G-10	Cl
	G-4	$\mathbf{CF_3}$	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
	G-4	CN	G-10	OCF_3
3 0	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF_3		
	G-5	OCF ₃		

$$R_a^2 = R_b^2 = Y = H, R^3 = Ph$$

	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	\mathbf{Br}
	G-1	OCF ₃	$\mathrm{CH_2Ph}$	CF_3
-	G-1	CN	CH_2Ph	OCF ₃
15	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	C1
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
	G-2	CN	G-7	OCF_3
20	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	$\mathbf{CF_3}$	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
	G-3	CN	G-8	OCF ₃
25	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
	G-4	CN	G-10	OCF ₃
3 0	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF_3		
	G-5	OCF ₃		

$R_a^2 = R_b^2 = Y = H, R^3 = 4 - F - Ph$

	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF3	$\mathrm{CH_2Ph}$	\mathbf{Br}
	G-1	OCF ₃	$\mathrm{CH_2Ph}$	CF_3
	G-1	CN	$\mathrm{CH_2Ph}$	OCF_3
15	G-2	C1	CH_2Ph	CN
	G-2	Br	G-7	C1
	G-2	CF ₃	G-7	Br
	G-2	OCF3	G-7	CF_3
	G-2	CN	G-7	OCF_3
20	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
	G-3	CN	G-8	OCF_3
25	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF3	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
	G-4	CN	G-10	ocf ₃
30	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

WO 92/03421 PCT/US91/05334

5

$R_a^2 = R_b^2 = Y = H, R^3 = 4$ -Cl-Ph

10	G	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF3	CH ₂ Ph	\mathbf{Br}
	G-1	OCF ₃	CH_2Ph	CF3
15	G-1	CN	CH ₂ Ph	OCF ₃
	G-2	Cl	CH ₂ Ph	CN
	G-2	Br	G-7	·Cl
	G-2 G-2	CF ₃	G-7	Br
	G-2	OCF ₃	G-7	CF3
20	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF ₃
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	${f Br}$
	G-4	OCF ₃	G-10	$\mathbf{CF_3}$
3 0	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF_3		
	G-5	OCF ₃		

$$R_b^2 = Y = H, R_a^2 = Cl, R^3 = H$$

10	<u>G</u>	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	CH_2Ph	Cl
	G-1	CF ₃	CH_2Ph	Br
	G-1	OCF ₃	$\mathrm{CH_2Ph}$	CF_3
15	G-1	CN	$ ext{CH}_2 ext{Ph}$	OCF3
10	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2 G-2	CF ₃	G-7	Br
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF3	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF3	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF3		

$$R_b^2 = Y = H, R_a^2 = Cl, R^3 = Me$$

10	G	<u>R</u> 1	<u>G</u>	<u>R</u> 1
	G-1	Cl	G-5	CN
	G-1	Br	CH_2Ph	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	\mathbf{Br}
	G-1	OCF3	$\mathrm{CH_2Ph}$	CF ₃
15	G-1	CN	$\mathtt{CH_2Ph}$	OCF ₃
	G-2	CI	CH ₂ Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	Br
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	${f Br}$
	G-3	OCF3	G-8	CF_3
25	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	${f Br}$
	G-4	OCF ₃	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

ŝ

R_b^2 =Y=H, R_a^2 =Cl, R^3 =Et

10	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	CH_2Ph	\mathbf{Br}
	G-1	OCF3	$\mathrm{CH_2Ph}$	$\mathbf{CF_3}$
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	\mathbf{Br}	G-7	Cl
	G-2	CF ₃	G-7	$\operatorname{\mathtt{Br}}$
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF_3	G-8	\mathbf{Br}
	G-3	OCF_3	G-8	CF_3
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	${f Br}$
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF_3		
	G-5	OCF ₃	•	

$$R_b^2$$
=Y=H, R_a^2 =Cl, R^3 =i-Pr

	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
10	G-1	C1	G-5	CN
	G-1	Br	CH_2Ph	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	\mathbf{Br}
	G-1	OCF3	$\mathrm{CH_2Ph}$	CF_3
	G-1	CN	$\mathrm{CH_2Ph}$	OCF ₃
15	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF3	G-7	${f Br}$
	G-2	OCF ₃	G-7	$\mathbf{CF_3}$
	G-2	CN	G-7	OCF3
20	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G- 8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
	G-3	CN	G-8	OCF_3
25	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	${\tt Br}$
	G-4	OCF ₃	G-10	$\mathbf{CF_3}$
	G-4	CN	G-10	ocf_3
30	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF_3		
	G-5	OCF3		

$R_b^2 = Y = H, R_a^2 = Cl, R^3 = CO_2Me$

	<u>G</u>	<u>R</u> 1	<u>G</u>	<u>R</u> 1
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	\mathbf{Br}
	G-1	OCF ₃	$\mathrm{CH_2Ph}$	CF_3
	G-1	CN	$\mathrm{CH_2Ph}$	OCF3
15	G-2	Cl	CH ₂ Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	Br
	G-2	OCF3	G-7	CF_3
	G-2	CN	G-7	OCF ₃
20	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	Br
	G-3	OCF ₃	G-8	CF_3
	G-3	CN	G-8	OCF_3
2 5	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	$\mathbf{CF_3}$
	G-4	CN	G-10	OCF3
30	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF3		
	G-5	OCF ₃		

$$R_b^2$$
=Y=H, R_a^2 =Cl, R^3 =Ph

10	<u>G</u>	<u>R</u> 1	$\underline{\mathbf{G}}$	<u>R</u> 1
	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	\mathbf{Br}
	G-1	OCF3	$\mathrm{CH_2Ph}$	CF_3
15	G-1	CN	CH_2Ph	OCF ₃
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	Br
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF3	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
3 0	G-4	CN	G-10	OCF ₃
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

ş

3.

$$R_b^2 = Y = H, R_a^2 = Cl, R^3 = 4-F-Ph$$

10	<u>G</u>	\mathbb{R}^1	<u>G</u>	<u>R</u> 1
10	G-1	Cl	G-5	CN
	G-1 G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	Br
		_	CH ₂ Ph	CF ₃
	G-1	OCF ₃		OCF ₃
1 5	G-1	CN	CH ₂ Ph	_
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	Br
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
	G-4	OCF3	G-10	CF_3
3 0	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	. CN
	G-5	Br		
	G-5	CF3		
	G-5	OCF3		

$$R_b^2 = Y = H, R_a^2 = Cl, R^3 = 4 - Cl - Ph$$

10	G	<u>R</u> 1	<u>G</u>	<u>R</u> 1
	G-1	Cl	G-5	CN
	G-1	Br	CH_2Ph	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	\mathbf{Br}
	G-1	OCF ₃	$\mathrm{CH_2Ph}$	$\mathbf{CF_3}$
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF ₃
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	C-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF3	G-10	CF ₃
3 0	G-4	CN	G-10	OCF_3
	G-5	Cl .	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

$$R_b^2 = Y = H, R_a^2 = CF_3, R^3 = H$$

10	<u>G</u>	\mathbb{R}^1	\mathbf{G}	\mathbb{R}^1
10	G-1	Cl	G- 5	CN
	G-1	Br	CH ₂ Ph	Cl
		CF ₃	CH_2Ph	Br
	G-1		_	CF ₃
	G-1	OCF ₃	CH ₂ Ph	_
1 5	G-1	CN	CH ₂ Ph	OCF3
	G-2	C1	$\mathtt{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF_3	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF ₃
20	G-2	CN .	G-7	OCF_3
	G-3	C1	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF_3	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	$\mathtt{CF_3}$	G-10	Br
	G-4	OCF3	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	\mathbf{Br}		
	G-5	$\mathbf{CF_3}$		
	G-5	OCF3		

R_b^2 =Y=H, R_a^2 =CF₃, R^3 =Me

10	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	CH_2Ph	\mathbf{Br}
	G-1	OCF ₃	$\mathrm{CH_2Ph}$	$\mathbf{CF_3}$
15	G-1	CN	${ m CH_2Ph}$	OCF_3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	Br
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF3	G-8	CF_3
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	CI
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

$R_b^2 = Y = H, R_a^2 = CF_3, R^3 = Et$

10	<u>G</u>	\mathbb{R}^1	<u>G</u>	<u>R</u> 1
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathtt{CH_2Ph}$	Cl
	G-1	CF ₃	CH ₂ Ph	\mathbf{Br}
	G-1	OCF3	CH ₂ Ph	CF3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF ₃
	G-2	Cl	${ m CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	Br
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	CI
	G-3	CF_3	G-8	Br
	G-3	OCF ₃	G-8	CF ₃
25	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	C1	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

$R_b^2 = Y = H, R_a^2 = CF_3, R^3 = i - Pr$

10	<u>G</u>	<u>R</u> 1	G	<u>R</u> 1
	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	\mathbf{Br}
	G-1	OCF ₃	CH ₂ Ph	CF_3
15	G-1	CN	CH_2Ph	OCF_3
	G-2	Cl	CH_2Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF ₃
30	G-4	CN	G-10	OCF ₃
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

$R_b^2 = Y = H, R_a^2 = CF_3, R^3 = CO_2Me$

10	<u>G</u>	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathtt{CH_2Ph}$	Cl
	G-1	CF_3	$\mathrm{CH_2Ph}$	${f Br}$
	G-1	OCF ₃	CH ₂ Ph	CF_3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF3
	G-2	Cl	CH_2Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF_3	G-8	\mathbf{Br}
	G-3	OCF3	G-8	$\mathbf{CF_3}$
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF_3		
	G-5	OCF ₃		

$R_b^2 = Y = H, R_a^2 = CF_3, R^3 = Ph$

10	<u>G</u>	<u>R</u> 1	<u>G</u>	<u>R</u> 1
	G-1	Cl	G-5	CN
	G-1	Br	CH_2Ph	C1
	G-1	CF ₃	$\mathrm{CH_2Ph}$	\mathbf{Br}
	G-1	OCF3	CH ₂ Ph	CF_3
15	G-1	CN	CH ₂ Ph	OCF3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	$\mathbf{CF_3}$
20	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF_3	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
2 5	G-3	CN	G-8	OCF ₃
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
3 0	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

$$R_b^2 = Y = H, R_a^2 = CF_3, R^3 = 4-F-Ph$$

10	<u>G</u>	<u>R</u> 1	<u>G</u>	<u>R</u> 1
10	G-1	Cl	G-5	CN
	G-1	Br	CH_2Ph	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	\mathbf{Br}
	G-1	OCF ₃	CH ₂ Ph	CF_3
15	G-1	CN	$\mathtt{CH_2Ph}$	OCF ₃
10	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	$\mathbf{CF_3}$
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF3	G-8	CF_3
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃ .	G-10	${f Br}$
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

R_b^2 =Y=H, R_a^2 =CF₃, R^3 =4-Cl-Ph

10	<u>G</u>	<u>R</u> 1	<u>G</u>	<u>R</u> 1
	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	\mathbf{Br}
	G-1	OCF ₃	$\mathrm{CH_2Ph}$	CF_3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF ₃
	G-2	Cl	$ ext{CH}_2 ext{Ph}$	CN
	G-2	Br	G-7	.C1
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	$\mathbf{CF_3}$
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	${f Br}$
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF3	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF ₃
	G-5	Cl	G-10	CN
	G-5	\mathbf{Br}		
	G-5	CF ₃		
	G-5	OCF3		

$$R_b^2$$
=Y=H, R_a^2 =Br, R^3 =H

10	<u>G</u>	R ¹	<u>G</u>	R ¹
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	Br
	G-1	OCF3	$\mathrm{CH_2Ph}$	$\mathbf{CF_3}$
15	G-1	CN	CH_2Ph	OCF3
	G-2	Cl	CH_2Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF3	G-8	CF_3
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

$$R_b^2 = Y = H, R_a^2 = Br, R^3 = Me$$

10	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
	G-1	Cl	G-5	CN
	G-1	Br	CH_2Ph	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	Br
	G-1	OCF ₃	CH ₂ Ph	CF ₃
15	G-1	CN	CH ₂ Ph	OCF ₃
	G-2	Cl	CH ₂ Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	${f Br}$
	G-3	OCF ₃	G-8	CF_3
2 5	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
	G-4	OCF_3	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF_3		
	G-5	OCF ₃		

₹

$$R_b^2$$
=Y=H, R_a^2 =Br, R^3 =Et

10	G	\mathbb{R}^1	<u>G</u>	<u>R</u> 1
10	G-1	Cl	G- 5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	\mathbf{Br}
	G-1	OCF ₃	$\mathrm{CH_2Ph}$	CF3
15		CN	CH ₂ Ph	OCF3
15	G-1		_	•
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	${f Br}$	G-7	Cl
	G-2	$\mathbf{CF_3}$	G-7	Br
	G-2	OCF_3	G-7	$\mathbf{CF_3}$
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	C1
	G-3	$\mathbf{CF_3}$	G-8	Br
	G-3	OCF3	G-8	CF_3
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	${\tt Br}$	G-10	Cl
	G-4	CF ₃	G-10	Br
	G-4	OCF ₃	G-10	CF_3
3 0	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	${\tt Br}$		
	G-5	CF_3		
	G-5	OCF3		

$$R_b^2 = Y = H, R_a^2 = Br, R^3 = i - Pr$$

10	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	${ m CH_2Ph}$	\mathbf{Br}
	G-1	OCF3	$\mathrm{CH_2Ph}$	CF_3
15	G-1	CN	CH ₂ Ph	OCF ₃
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF3	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	$\mathbf{CF_3}$
2 5	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	${\tt Br}$
	G-4	OCF ₃	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

$$R_b^2 = Y = H, R_a^2 = Br, R^3 = CO_2Me$$

10	<u>G</u>	\mathbb{R}^1	<u>G</u>	<u>R</u> 1
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathtt{CH_2Ph}$	C1
	G-1	$\mathbf{CF_3}$	CH ₂ Ph	Br
	G-1	OCF3	$\mathrm{CH_2Ph}$	CF_3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF3
	G-2	Cl	$\mathtt{CH_2Ph}$	CN
	G-2	Br	G-7	C1
	G-2	CF_3	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	\mathbf{Br}	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br	•	
	G-5	CF_3		
	G-5	OCF3		

$$R_b^2 = Y = H, R_a^2 = Br, R^3 = 4 - F - Ph$$

10	<u>G</u>	<u>R</u> 1	$\underline{\mathbf{G}}$	<u>R</u> 1
	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	\mathbf{Br}
	G-1	OCF ₃	$\mathrm{CH_2Ph}$	CF ₃
15	G-1	CN	CH ₂ Ph	OCF ₃
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	$\operatorname{\mathtt{Br}}$
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF ₃
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF_3		

$$R_b^2 = Y = H, R_a^2 = Br, R^3 = Ph$$

10	<u>G</u>	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1 G-1	Br	CH_2Ph	Cl
	G-1	CF ₃	CH_2Ph	Br
	G-1	OCF ₃	CH ₂ Ph	CF3
4 5		CN	CH ₂ Ph	OCF ₃
15	G-1		_	•
	G-2	Cl	CH ₂ Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF_3	G-7	CF ₃
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	${\tt Br}$	G-8	C1
	G-3	CF_3	G-8	Br
	G-3	OCF_3	G-8	CF_3
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	Br
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF_3		
	G-5	OCF ₃		

$$R_b^2$$
=Y=H, R_a^2 =Br, R^3 =4-Cl-Ph

10	<u>G</u>	R ¹	<u>G</u>	$\underline{\mathbf{R}^1}$
10	C 1	CI	0.5	CINT
	G-1 G-1	Cl Br	G-5	CN
			CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	Br
	G-1	OCF ₃	CH_2Ph	$\mathbf{CF_3}$
15	G-1	CN	CH ₂ Ph	OCF3
	G-2	Cl	CH ₂ Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
2 0	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

Ł

R_b^2 =Y=H, R_a^2 =OCHF₂, R^3 =H

	<u>G</u>	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10		~ 1	0.5	CINT
	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	C1
	G-1	$\mathbf{CF_3}$	$\mathtt{CH_2Ph}$	\mathbf{Br}
	G-1	OCF3	$\mathrm{CH_2Ph}$	$\mathbf{CF_3}$
15	G-1	CN	$\mathrm{CH_2Ph}$	ocf_3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF_3	G-7	CF_3
2 0	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF_3	G-8	\mathbf{Br}
	G-3	OCF_3	G-8	$\mathbf{CF_3}$
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	${f Br}$	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
•	G-4	OCF ₃	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF3
	G-5	Cl	G-10	CN
	G-5	\mathbf{Br}		
	G-5	CF_3		
	G-5	OCF ₃		

R_b^2 =Y=H, R_a^2 =OCHF₂, R^3 =Me

10	<u>G</u>	<u>R</u> ¹	<u>G</u>	\mathbb{R}^1
	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	${\tt Br}$
	G-1	OCF3	CH ₂ Ph	CF_3
15	G-1	CN	CH_2Ph	OCF3
	G-2	Cl	CH ₂ Ph	CN
	G-2	\mathbf{Br}	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF3		
	G-5	OCF ₃		

-

R_b^2 =Y=H, R_a^2 =OCHF₂, R^3 =Et

10	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	${ m CH_2Ph}$	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	Br
	G-1	OCF3	$\mathrm{CH_2Ph}$	CF3
15	G-1	CN	$\mathtt{CH_2Ph}$	OCF ₃
	G-2	Cl	CH_2Ph	CN
	G-2	Br	G-7	Cl
	G-2	$\mathbf{CF_3}$	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	${f Br}$	G-8	C1
	G-3	$\mathbf{CF_3}$	G-8	\mathbf{Br}
	G-3	OCF3	G-8	CF_3
25	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	CN
	G-4	${f Br}$	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF3		

R_b^2 =Y=H, R_a^2 =OCHF₂, R^3 =i-Pr

G-1 Cl G-5	CN Cl
C1 D. CU.D.	
$G-1$ Br CH_2Ph	
$G-1$ CF_3 CH_2Ph	${\tt Br}$
G-1 OCF ₃ CH ₂ Ph	CF_3
15 G-1 CN CH ₂ Ph	OCF ₃
G-2 Cl CH ₂ Ph	CN
G-2 Br G-7	Cl
G-2 CF ₃ G-7	${f Br}$
G-2 OCF ₃ G-7	CF_3
20 G-2 CN G-7	OCF ₃
G-3 Cl G-7	CN
G-3 Br G-8	Cl
G-3 CF ₃ G-8	Br
G-3 OCF ₃ G-8	$\mathbf{CF_3}$
25 G-3 CN G-8	OCF ₃
G-4 Cl G-8	CN
G-4 Br G-10	Cl
$G-4$ CF_3 $G-10$	\mathbf{Br}
G-4 OCF ₃ G-10	CF_3
30 G-4 CN G-10	OCF ₃
G-5 Cl G-10	CN
G-5 Br	
G-5 CF ₃	
G-5 OCF ₃	

$R_b^2 = Y = H, R_a^2 = OCHF_2, R^3 = CO_2Me$

10	\mathbf{G}	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF3	$\mathtt{CH_2Ph}$	\mathbf{Br}
	G-1	OCF3	$\mathrm{CH_2Ph}$	CF_3
15	G-1	CN	CH_2Ph	OCF3
	G-2	Cl	CH_2Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	Br
	G-2	OCF3	G-7	CF3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF_3	G-8	$\mathbf{CF_3}$
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
	G-4	OCF3	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	$\mathbf{CF_3}$		
	G-5	OCF ₃		

R_b^2 =Y=H, R_a^2 =OCHF₂, R^3 =Ph

10	<u>G</u>	\mathbb{R}^1	G	\mathbb{R}^1
	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	CH_2Ph	\mathbf{Br}
	G-1	OCF3	CH ₂ Ph	$\mathbf{CF_3}$
1 5	G-1	CN	CH_2Ph	OCF ₃
	G-2	Cl	CH_2Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	${\tt Br}$
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF_3	G-8	${f Br}$
	G-3	OCF3	G-8	CF_3
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
	G-4	OCF_3	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF_3		
	G-5	OCF ₃		

$$R_b^2$$
=Y=H, R_a^2 =OCHF₂, R^3 =4-Cl-Ph

10	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF_3	$\mathrm{CH_2Ph}$	Br
	G-1	OCF3	${ m CH_2Ph}$	CF ₃
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF3
	G-2	Cl	CH_2Ph	CN
	G-2	${\tt Br}$	G-7	Cl
	G-2	CF ₃	G-7	Br
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G -8	Cl
	G-3	CF_3	G-8	Br
	G-3	OCF_3	G-8	CF_3
25	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF3	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF_3		
	G-5	OCF_3		

$$R_b^2$$
=Y=H, R_a^2 =OCHF₂, R^3 =4-F-Ph

<u>G</u>	R ¹	$\underline{\mathbf{G}}$	<u>R</u> 1
0 1	CI.	~ *	~
			CN
			Cl
G-1	$\mathbf{CF_3}$	CH_2Ph	\mathbf{Br}
G-1	OCF ₃	$\mathrm{CH_2Ph}$	CF_3
G-1	CN	CH ₂ Ph	OCF_3
G-2	Cl	$\mathrm{CH_2Ph}$	CN
G-2	Br	G-7	Cl
G-2	CF ₃	G-7	${f Br}$
G-2	OCF ₃	G-7	CF_3
G-2	CN	G-7	OCF_3
G-3	Cl	G-7	CN
G-3	Br	G-8	Cl
G-3	CF ₃	G-8	\mathbf{Br}
G-3	ocf ₃	G-8	CF_3
G-3	CN	G-8	OCF_3
G-4	Cl	G-8	CN
G-4	Br	G-10	C1
G-4	CF ₃	G-10	\mathbf{Br}
G-4	OCF ₃	G-10	CF_3
G-4	CN	G-10	OCF_3
G-5	Cl	G-10	CN
G-5	Br		
G-5	CF ₃		
G-5	OCF ₃		
	G-1 G-1 G-1 G-1 G-2 G-2 G-2 G-2 G-3 G-3 G-3 G-3 G-3 G-4 G-4 G-4 G-4 G-4 G-4 G-5 G-5 G-5	G-1 Cl G-1 Br G-1 CF3 G-1 OCF3 G-1 CN G-2 Cl G-2 Br G-2 CF3 G-2 OCF3 G-2 CN G-3 Cl G-3 Br G-3 CF3 G-3 CF3 G-3 CF3 G-4 CN G-4 Cl G-4 Br G-4 CF3 G-4 CF3 G-4 CN G-5 Cl G-5 Br G-5 CF3	G-1 Cl G-5 G-1 Br CH ₂ Ph G-1 CF ₃ CH ₂ Ph G-1 OCF ₃ CH ₂ Ph G-1 CN CH ₂ Ph G-1 CN CH ₂ Ph G-2 Cl CH ₂ Ph G-2 G-7 G-2 CF ₃ G-7 G-2 CN G-7 G-2 CN G-7 G-3 Cl G-7 G-3 Cl G-7 G-3 G-8 G-3 CF ₃ G-8 G-3 CF ₃ G-8 G-3 CF ₃ G-8 G-3 CF ₃ G-8 G-4 Cl G-8 G-4 CF ₃ G-10 G-4 CF ₃ G-10 G-5 Cl G-10 G-5 Cl G-10 G-5 CF ₃ G-10

$$R_b^2$$
=Y=H, R_a^2 =OCH₂CF₃, R^3 =H

10	G	<u>R</u> 1	\mathbf{G}	<u>R</u> 1
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	CH_2Ph	Br
	G-1	OCF3	$\mathrm{CH_2Ph}$	CF ₃
15	G-1	CN	$\operatorname{CH_2Ph}$	OCF3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	$\mathbf{CF_3}$	G-7	Br
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	${f Br}$	G-8	Cl
	G-3	CF_3	G-8	\mathbf{Br}
	G-3	OCF3	G-8	CF_3
2 5	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	$\mathbf{CF_3}$	G-10	\mathtt{Br}
	G-4	OCF ₃	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	\mathbf{Br}		
	G-5	CF_3		
	G-5	OCF ₃		

$$R_b^2 = Y = H, R_a^2 = OCH_2CF_3, R^3 = Me$$

10	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	\mathbf{Br}
15	G-1	OCF3	CH_2Ph	$\mathbf{CF_3}$
	G-1	CN	$\overline{\mathrm{CH_2Ph}}$	OCF3
	G-2	Cl	CH_2Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF_3	G-7	\mathbf{Br}
20	G-2	OCF3	G-7	$\mathbf{CF_3}$
	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
2 5	G-3	OCF ₃	G-8	$\mathbf{CF_3}$
	G-3	CN	G-8	OCF ₃
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	${f Br}$
30	G-4	OCF ₃	G-10	CF_3
	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
35	G-5	OCF ₃		

$$\mathtt{R}_{b}^{2}\mathtt{=Y=H},\,\mathtt{R}_{a}^{2}\mathtt{=OCH_{2}CF_{3}},\,\mathtt{R}^{3}\mathtt{=Et}$$

10	<u>G</u>	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	CH_2Ph	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	\mathbf{Br}
	G-1	OCF ₃	CH_2Ph	CF_3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF ₃
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	CI
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF3	G-8	CF_3
25	G-3	CN	G-8	OCF ₃
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF3	G-10	$\mathbf{CF_3}$
3 0	G-4	CN	G-10	OCF_3
	G-5	C1	G-10	CN
	G-5	Br		
	G-5	CF3		
	G-5	OCF ₃		

R_b^2 =Y=H, R_a^2 =OCH₂CF₃, R^3 =i-Pr

10	<u>G</u>	R ¹	<u>G</u>	\mathbb{R}^1
	G-1	Cl	G-5	CN
	G-1	Br	CH_2Ph	Cl
	G-1	CF ₃	CH ₂ Ph	\mathbf{Br}
	G-1	OCF ₃	$\mathrm{CH_2Ph}$	CF ₃
15	G-1	CN	CH_2Ph	OCF3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	Br
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF ₃
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	Br
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF3
	G-5	Cl	G-10	CN
	G-5	Br		
	G -5	CF ₃		
	G-5	OCF ₃		

$$R_b^2$$
=Y=H, R_a^2 =OCH₂CF₃, R^3 =CO₂Me

10	G	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	CH_2Ph	Cl
	G-1	CF_3	CH_2Ph	${f Br}$
	G-1	OCF3	CH_2Ph	CF_3
15	G-1	CN .	$\mathrm{CH_2Ph}$	OCF3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF ₃
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF_3	G-8	CF_3
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF_3	G-10	${f Br}$
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	$\mathbf{CF_3}$		
	G-5	OCF3		

$$R_b^2 = Y = H, R_a^2 = OCH_2CF_3, R^3 = Ph$$

	\mathbf{G}	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10		•		
	G-1	Cl	G-5	CN
	G-1	Br	CH_2Ph	Cl
	G-1	CF ₃	CH_2Ph	\mathbf{Br}
	G-1	OCF3	CH ₂ Ph	CF_3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF_3
	G-2	Cl	CH ₂ Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF ₃
	G-4	Cl	G-8	CN
	G-4	Br	G-10	C1
	G-4	CF_3	G-10	${\tt Br}$
	G-4	OCF_3	G-10	$\mathbf{CF_3}$
3 0	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	\mathbf{Br}		
	G-5	CF ₃		
	G-5	OCF ₃		

${\scriptstyle R_b^2=Y=H,\;R_a^2=OCH_2CF_3,\;R^3=4\text{-}Cl\text{-}Ph}$

10	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1 G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH_2Ph	Br
	G-1	OCF ₃	CH ₂ Ph	CF ₃
4 -			CH ₂ Ph	OCF ₃
15	G-1	CN	-	CN
	G-2	Cl	CH ₂ Ph	
	G-2	${\tt Br}$	G-7	CI
	G-2	$\mathbf{CF_3}$	G-7	Br
	G-2	OCF ₃	G-7	$\mathbf{CF}_{\mathfrak{F}}$
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl -
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF3	G-8	$\mathbf{CF_3}$
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	\mathbf{Br}	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF3	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	$\mathbf{CF_3}$		
	G-5	OCF3		

WO 92/03421

5

R_b^2 =Y=H, R_a^2 =OCH₂CF₃, R^3 =4-F-Ph

<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
G-1	Cl	G-5	CN
G-1	Br	CH ₂ Ph	Cl
G-1	CF3	$\mathrm{CH_2Ph}$	Br
G-1	OCF3	CH ₂ Ph	CF_3
G-1	CN	CH ₂ Ph	OCF ₃
G-2	Cl	CH_2Ph	CN
G-2	Br	G-7	Cl
G-2	$\mathbf{CF_3}$	G-7	\mathbf{Br}
G-2	OCF ₃	G-7	CF_3
G-2	CN	G-7	OCF ₃
G-3	Cl	G-7	CN
	Br		Cl
G-3	CF_3	G-8	Br
G-3	OCF ₃	G-8	CF_3
G-3	CN	G-8	OCF3
G-4	Cl	G-8	CN
G-4	\mathbf{Br}	G-10	Cl
G-4	CF_3	G-10	\mathbf{Br}
G-4	OCF ₃	G-10	CF_3
G-4	CN	G-10	OCF ₃
G-5	Cl	G-10	CN
	\mathbf{Br}		
G-5	CF_3		
G-5	OCF ₃		
	G-1 G-1 G-1 G-1 G-1 G-2 G-2 G-2 G-2 G-3 G-3 G-3 G-3 G-3 G-4 G-4 G-4 G-4 G-4 G-4 G-5 G-5 G-5	G-1 Cl G-1 Br G-1 CF3 G-1 OCF3 G-1 CN G-2 Cl G-2 Br G-2 CF3 G-2 OCF3 G-2 CN G-3 Cl G-3 Br G-3 CF3 G-3 CF3 G-3 CF3 G-4 Cl G-4 Br G-4 Cl G-4 Br G-4 CF3 G-4 CF3 G-4 CN G-5 Cl G-5 Br G-5 CF3	G-1 Cl G-5 G-1 Br CH ₂ Ph G-1 CF ₃ CH ₂ Ph G-1 OCF ₃ CH ₂ Ph G-1 CN CH ₂ Ph G-1 CN CH ₂ Ph G-2 Cl CH ₂ Ph G-2 Cl CH ₂ Ph G-2 CF ₃ G-7 G-2 CF ₃ G-7 G-2 CN G-7 G-3 Cl G-7 G-3 Cl G-7 G-3 G-8 G-3 CF ₃ G-8 G-3 CF ₃ G-8 G-3 CF ₃ G-8 G-3 CN G-8 G-4 Cl G-10 G-5 G-4 CN G-10 G-5 G-5 Cl G-10 G-5 G-5 G-5 G-5 G-5 CF ₃ G-10

$$R_a^2 = Y = H, R_b^2 = F, R^3 = H$$

10	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
	G-1	Cl	G-5	CN
	G-1	\mathbf{Br}	$\mathtt{CH_2Ph}$	Cl
	G-1	CF ₃	${ m CH_2Ph}$	\mathbf{Br}
	G-1	OCF ₃	${ m CH_2Ph}$	CF ₃
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF_3
	G-2	Cl	${ m CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF_3	G-7	$\mathbf{CF_3}$
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	${f Br}$	G-8	Cl
	G-3	CF3	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	$\mathbf{CF_3}$
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	$\operatorname{\mathtt{Br}}$	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF3	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
•	G-5	Br		
	G-5	$\mathbf{CF_3}$		
	G-5	OCF ₃		

$$R_a^2 = Y = H, R_b^2 = F, R^3 = Me$$

	<u>G</u>	<u>R</u> 1	\mathbf{G}	<u>R</u> 1
10				
	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	CH_2Ph	\mathbf{Br}
	G-1	OCF ₃	CH ₂ Ph	CF ₃
15	G-1	CN	CH_2Ph	OCF_3
	G-2	Cl	CH_2Ph	CN
	G-2	Br	G-7	Cl
	G-2	$\mathbf{CF_3}$	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl-
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF3	G-8	CF_3
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF ₃
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃	•	
	G-5	OCF ₃		

$$R_a^2 = Y = H, R_b^2 = F, R^3 = Et$$

	G	R ¹	\mathbf{G}	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1 G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	Br
	G-1	OCF ₃	CH ₂ Ph	CF3
15		CN	CH ₂ Ph	OCF ₃
15	G-1		CH ₂ Ph	CN
	G-2	C1	G-7	Cl
	G-2	Br CF3	G-7	Br
	G-2	_	G-7	CF ₃
	G-2	OCF3		
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	C1
	G-3	CF_3	G-8	Br
	G-3	OCF3	G-8	CF_3
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	${f Br}$	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
	G-4	OCF3	G-10	CF_3
30	G-4	CN	G-10	OCF3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF3		
		-		

WO 92/03421 PCT/US91/05334 170

$$R_a^2 = Y = H, R_b^2 = F, R^3 = i - Pr$$

10	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	Br
	G-1	OCF ₃	CH_2Ph	CF ₃
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	C1
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	Br
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF ₃
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF3		

$$R_a^2 = Y = H, R_b^2 = F, R^3 = CO_2Me$$

10	<u>G</u>	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH_2Ph	\mathbf{Br}
	G-1	OCF ₃	CH_2Ph	CF_3
15	G-1	CN	$\mathtt{CH_2Ph}$	OCF ₃
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	Br
	G-3	OCF3	G-8	$\mathbf{CF_3}$
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	ocf_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

$$R_a^2 = Y = H, R_b^2 = F, R^3 = Ph$$

10	<u>G</u>	\mathbb{R}^1	\mathbf{G}	\mathbb{R}^1
10	0.1	CI.	0.5	CDT
	G-1 G-1	Cl Br	G-5	CN Cl
			CH ₂ Ph	
	G-1	CF ₃	CH ₂ Ph	Br
	G-1	OCF ₃	$\mathrm{CH_2Ph}$	$\mathbf{CF_3}$
1 5	G-1	CN	CH ₂ Ph	OCF_3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF_3	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
3 0	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

$$R_a^2$$
=Y=H, R_b^2 =F, R^3 =4-Cl-Ph

10	<u>G</u>	R ¹	<u>G</u>	R ¹
	G-1	Cl	G-5	CN
	G-1	Br	CH_2Ph	Cl
	G-1	CF ₃	CH_2Ph	\mathbf{Br}
	G-1	OCF3	$\mathrm{CH_2Ph}$	$\mathbf{CF_3}$
15	G-1	CN	CH_2Ph	ocf_3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF3	G-7	Br
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	\mathbf{Br}		
	G-5	CF ₃		
	G-5	OCF_3		

$$R_a^2$$
=Y=H, R_b^2 =F, R^3 =4-F-Ph

10	<u>G</u>	\mathbb{R}^1	<u>G</u>	<u>R</u> 1
10	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	Br
	G-1	OCF ₃	$\mathrm{CH_2Ph}$	CF_3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF3
	G-2	CI	CH_2Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
2 5	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF_3	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
-	G-5	Br		
	G-5	CF ₃		
	G-5	OCF_3		

$$R_a^2 = R_b^2 = H$$
, Y=COMe, $R^3 = Me$

10	<u>G</u>	\mathbb{R}^1	$\underline{\mathbf{G}}$	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	${ m CH_2Ph}$	Br
	G-1	OCF3	$\mathrm{CH_2Ph}$	CF_3
15	G-1	CN	$\mathtt{CH_2Ph}$	OCF_3
	G-2	Cl	$\mathtt{CH_2Ph}$	CN
	G-2	\mathtt{Br}	G-7	Cl
	G-2	$\mathtt{CF_3}$	G-7	\mathbf{Br}
	G-2	OCF_3	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	${f Br}$	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF_3	G-8	$\mathbf{CF_3}$
2 5	G-3	CN	G-8	ocf_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	$\mathbf{CF_3}$	G-10	${f Br}$
	G-4	OCF_3	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF ₃
	G-5	Cl	G-10	CN
	G-5	${\tt Br}$		
	G-5	CF_3		
	G-5	OCF ₃		

 $R_a^2 = R_b^2 = H$, Y=COMe, $R^3 = i$ -Pr

10	<u>G</u>	\mathbb{R}^1	\mathbf{G}	\mathbb{R}^1
10	- G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	\mathbf{Br}
	G-1	OCF ₃	CH ₂ Ph	CF_3
15	G-1	CN	${ m CH_2Ph}$	OCF ₃
	G-2	Cl	${ m CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF_3	G-8	Br
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	- G-8	OCF_3
	G-4	C1	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
3 0	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

 R_a^2 = R_b^2 =H, Y=COMe, R^3 =CO₂Me

10	<u>G</u>	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	Br
	G-1	OCF ₃	CH ₂ Ph	CF3
15	G-1	CN	CH ₂ Ph	OCF ₃
	G-2	Cl	CH_2Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	$\mathbf{CF_3}$
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF3	G-8	\mathbf{Br}
	G-3	OCF3	G-8	CF_3
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	Br
	G-4	OCF ₃	G-10	$\mathbf{CF_3}$
3 0	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF3		

$R_a^2=R_b^2=H$, Y=COMe, $R^3=4$ -F-Ph

	<u>G</u>	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10				
	G-1	C1	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	$\mathbf{CF_3}$	$\mathtt{CH_2Ph}$	\mathbf{Br}
	G-1	OCF ₃	$\mathrm{CH_2Ph}$	CF_3
15	G-1	CN	CH_2Ph	OCF_3
	G-2	Cl	${ m CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF_3	G-7	${f Br}$
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	CN
	G-4	\mathbf{Br}	G-10	Cl
	G-4	$\mathbf{CF_3}$	G-10	${f Br}$
	G-4	OCF ₃	G-10	CF3
30	G-4	CN	G-10	OCF3
	G-5	Cl	G-10	CN
	G- 5	\mathbf{Br}		
	G-5	CF_3		
	G-5	OCF_3		

$$R_a^2 = R_b^2 = H$$
, Y=CO₂Me, R³=Me

<u>G</u>	<u>R</u> 1	<u>G</u>	<u>R</u> 1
G-1	Cl	G-5	CN
		$\mathrm{CH_2Ph}$	Cl
	CF ₃	CH ₂ Ph	\mathbf{Br}
		CH ₂ Ph	$\mathbf{CF_3}$
	<u> </u>	CH ₂ Ph	OCF3
		CH ₂ Ph	CN
		G-7	Cl
		G-7	\mathbf{Br}
	_	G-7	$\mathbf{CF_3}$
		G-7	OCF ₃
		G-7	CN
G-3	Br	G-8	Cl
G-3	CF ₃	G-8	\mathbf{Br}
G-3	OCF3	G-8	CF_3
G-3	CN	G-8	OCF ₃
G-4	Cl	G-8	CN
G-4	Br	G-10	Cl
G-4	CF_3	G-10	\mathbf{Br}
G-4	OCF3	G-10	CF_3
G-4	CN	G-10	OCF3
G-5	Cl	G-10	CN
G-5	Br		
G-5	CF ₃		
G-5	OCF ₃		
	G-1 G-1 G-1 G-1 G-1 G-2 G-2 G-2 G-2 G-2 G-3 G-3 G-3 G-3 G-3 G-4 G-4 G-4 G-4 G-4 G-4 G-5 G-5 G-5	G-1 Cl G-1 Br G-1 CF3 G-1 OCF3 G-1 CN G-2 Cl G-2 Br G-2 CF3 G-2 CF3 G-2 CN G-3 Cl G-3 Br G-3 CF3 G-3 CF3 G-3 CF3 G-4 Cl G-4 Cl G-4 Br G-4 CF3 G-4 CF3 G-4 CF3 G-4 CF3 G-4 CN G-5 Cl G-5 Br G-5 CF3	G-1 Cl G-5 G-1 Br CH ₂ Ph G-1 CF ₃ CH ₂ Ph G-1 OCF ₃ CH ₂ Ph G-1 CN CH ₂ Ph G-1 CN CH ₂ Ph G-2 Cl CH ₂ Ph G-2 CF ₃ G-7 G-2 CF ₃ G-7 G-2 CF ₃ G-7 G-2 CN G-7 G-3 Cl G-7 G-3 Cl G-7 G-3 Cl G-7 G-3 G-8 G-3 CF ₃ G-8 G-3 CF ₃ G-8 G-3 CF ₃ G-8 G-3 CN G-8 G-4 Cl G-8 G-4 CF ₃ G-10 G-4 CF ₃ G-10 G-5 Cl G-10 G-5 Cl G-5 G-5 CF ₃

$$R_a^2 = R_b^2 = H$$
, Y=CO₂Me, $R^3 = i$ -Pr

10	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	CH ₂ Ph	Br
	G-1	OCF3	CH ₂ Ph	CF_3
1 5	G-1	CN	CH ₂ Ph	OCF3
	G-2	Cl	CH ₂ Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	C1
	G-3	CF3	G-8	${f Br}$
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF ₃
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF_3		
	G-5	OCF ₃		

_. 5

$$R_a^2 = R_b^2 = H$$
, Y=CO₂Me, R³=CO₂Me

10	$\underline{\mathbf{G}}$	<u>R</u> 1	<u>G</u>	<u>R</u> 1
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	Br
	G-1	OCF3	$\mathrm{CH_2Ph}$	CF_3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF ₃
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	\mathbf{Br}	G-8	Cl
	G-3	CF_3	G-8	${f Br}$
	G-3	OCF3	G-8	CF_3
25	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	$\mathtt{CF_3}$	G-10	${\tt Br}$
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	${\tt Br}$		
	G-5	CF ₃		
	G-5	OCF_3		

WO 92/03421 PCT/US91/05334 182

$$R_a^2 = R_b^2 = H$$
, Y=CO₂Me, R³=4-F-Ph

10	<u>G</u>	<u>R</u> 1	<u>G</u>	R ¹
	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	\mathbf{Br}
	G-1	OCF ₃	CH ₂ Ph	CF3
15	G-1	CN	CH ₂ Ph	OCF ₃
	G-2	C1	CH ₂ Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
2 0	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G- 3	CF_3	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
3 0	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

R_a^2 =Cl, R_b^2 =H, Y=COMe, R^3 =Me

10	<u>G</u>	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	CH_2Ph	Cl
	G-1	CF ₃	CH_2Ph	Br
	G-1	OCF3	CH_2Ph	CF_3
15	G-1	CN	$\mathtt{CH_2Ph}$	OCF ₃
10	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	Br
	G-2	OCF ₃	G-7	CF3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF_3	G-8	Br
	G-3	OCF ₃	G-8	$\mathbf{CF_3}$
25	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	. CN
	G-4	Br	G-10	Cl
	G-4	CF3	G-10	\mathbf{Br}
	G-4	OCF_3	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF_2		

R_a^2 =Cl, R_b^2 =H, Y=COMe, R^3 =i-Pr

10	<u>G</u>	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10	· G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	Br
	G-1	OCF ₃	CH ₂ Ph	CF ₃
4 5		_	-	
15	G-1	CN	CH ₂ Ph	OCF ₃
	G-2	Cl	CH ₂ Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	Br
	G-2	OCF_3	G-7	CF_3
2 0	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	$\mathbf{CF_3}$
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	Br
	G-4	OCF ₃	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

G-5

OCF₃

5

R_a^2 =Cl, R_b^2 =H, Y=COMe, R^3 =CO₂Me

10	<u>G</u>	<u>R</u> 1	<u>G</u>	R ¹
10	G-1	Cl	G-5	CN
	G-1	Br	CH_2Ph	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	Br
	G-1	OCF ₃	$\mathrm{CH_2Ph}$	CF_3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF_3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF3	G-8	CF ₃
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	Br
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		

R_a^2 =Cl, R_b^2 =H, Y=COMe, R^3 =4-F-Ph

	\mathbf{G}	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10				
	G-1	C1	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH_2Ph	\mathbf{Br}
	G-1	OCF3	CH_2Ph	CF_3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF ₃
	G-2	Cl	CH_2Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF_3	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	C1
	G-3	CF_3	G-8	${f Br}$
	G-3	OCF_3	G-8	CF_3
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	$\mathbf{CF_3}$	G-10	\mathbf{Br}
	G-4	OCF_3	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	\mathbf{Br}		
	G-5	CF ₃		
	G-5	OCF ₃		

.

$$R_a^2$$
=Cl, R_b^2 =H, Y=CO₂Me, R^3 =Me

•				
10	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
10	0.1	Cl	G-5	CN
	G-1 G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	\mathbf{Br}
	G-1 G-1	OCF ₃	CH_2Ph	CF ₃
1.5		CN	CH ₂ Ph	OCF ₃
15	G-1		CH ₂ Ph	CN
	G-2	Cl	_	
	G-2	Br	G-7	Cl
	G-2	$\mathbf{CF_3}$	G-7	\mathbf{Br}
	G-2	OCF_3	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	C1
	G-3	CF_3	G-8	\mathbf{Br}
	G-3	OCF3	G-8	CF3
2 5	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
	G-4	OCF3	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	\mathbf{Br}		
	G-5	CF_3		
	G-5	OCF ₃		

 R_a^2 =Cl, R_b^2 =H, Y=CO₂Me, R^3 =i-Pr

10	<u>G</u>	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	\mathbf{Br}
	G-1	OCF ₃	CH ₂ Ph	CF ₃
15	G-1	CN	CH ₂ Ph	OCF3
	G-2	Cl	CH ₂ Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	Br
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF3	G-8	CF_3
2 5	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	CN
	G-4	\mathtt{Br}	G-10	C1
	G-4	CF3	G-10	${\tt Br}$
	G-4	OCF3	G-10	CF_3
30	G-4	CN	G-10	OCF ₃
	G-5	Cl	G-10	CN
	G-5 .	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

ş

R_a^2 =Cl, R_b^2 =H, Y=CO₂Me, R^3 =CO₂Me

10	<u>G</u>	\mathbb{R}^1	\mathbf{G}	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF3	CH_2Ph	\mathbf{Br}
	G-1	OCF_3	CH_2Ph	CF_3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	\mathbf{Br}	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	$\mathbf{CF_3}$	G-8	${\tt Br}$
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF3	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF_3		
	G-5	OCF ₂		

R_a^2 =Cl, R_b^2 =H, Y=CO₂Me, R^3 =4-F-Ph

10	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	\mathbf{Br}
	G-1	OCF ₃	${ m CH_2Ph}$	CF_3
1 5	G-1	CN	CH ₂ Ph	OCF ₃
	G-2	Cl	$\mathtt{CH_2Ph}$	CN
	G-2	${f Br}$	G-7	Cl
	G-2	CF ₃	G-7	${f Br}$
	G-2	OCF ₃	G-7	$\mathbf{CF_3}$
2 0	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	C1
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
2 5	G-3	CN	G-8	ocf_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	C1
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF3
	G-5	Cl	G-10	CN
	G-5	\mathbf{Br}		
	G-5	CF_3		
	G-5	OCF ₃		

ā

$$R_a^2$$
=CF₃, R_b^2 =H, Y=COMe, R^3 =Me

	G	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10		~ 1	0.5	CINT
	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	$\mathbf{CF_3}$	$\mathtt{CH_2Ph}$	\mathbf{Br}
	G-1	OCF_3	$\mathrm{CH_2Ph}$	$\mathbf{CF_3}$
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF_3
	G-2	Cl	${ m CH_2Ph}$	CN
	G-2	${f Br}$	G-7	Cl
	G-2	CF_3	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF_3	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	$\mathbf{CF_3}$
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	$\mathbf{CF_3}$	G-10	${\tt Br}$
	G-4	OCF_3	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G -5	CF_3		
	G-5	OCF ₃		

R_a^2 =CF₃, R_b^2 =H, Y=COMe, R^3 =i-Pr

10	<u>G</u>	R ¹	<u>G</u>	<u>R</u> 1
10	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	Br
	G-1	OCF3	CH_2Ph	CF_3
1 5	G-1	CN	$\mathrm{CH_2Ph}$	OCF3
	G-2	Cl	CH ₂ Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF3	G-7	Br
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	Br
	G-3	OCF ₃	G-8	CF_3
2 5	G-3	CN	G-8 .	OCF ₃
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF_3	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

G-5

 OCF_3

5

${\scriptstyle R_a^2=CF_3,\;R_b^2=H,\;Y=COMe,\;R^3=CO_2Me}$

10	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
10	G-1 G-1	Cl Br	G-5 CH ₂ Ph	CN Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	Br
	G-1	OCF3	CH_2Ph	CF3
15	G-1	CN	CH_2Ph	OCF3
	G-2	C1	CH_2Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF_3	G-8	\mathbf{Br}
	G-3	OCF3	G-8	$\mathbf{CF_3}$
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4'	CF ₃	G-10	\mathbf{Br}
	G-4	OCF3	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	C1	G-10	CN
	G-5	Br		
	G-5	CF ₃		

R_a^2 =CF₃, R_b^2 =H, Y=COMe, R³=4-F-Ph

10	<u>G</u>	<u>R</u> 1	<u>G</u>	R ¹
10	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	Br
		-	_	
	G-1	OCF ₃	CH ₂ Ph	CF ₃
15	G-1	CN	CH ₂ Ph	OCF3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	CI
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	$\mathbf{CF_3}$
20	G-2	CN	G-7	OCF ₃
	G-3	CI	G-7	CN
	G-3	Br	G-8	Cl
	G-3 (CF_3	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF ₃
25	G-3	CN	G-8	OCF_3
	G-4	CI .	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF_3	G-10	CF_3
30	G-4	CN	G-10	OCF ₃
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

$R_a^2 = CF_3$, $R_b^2 = H$, $Y = CO_2Me$, $R^3 = Me$

10	G	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	\mathbf{Br}
•	G-1	OCF3	$\mathrm{CH_2Ph}$	CF_3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF3	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	$\mathbf{CF_3}$
20	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	$\operatorname{\mathbf{Br}}$	G-8	Cl
	G-3	CF_3	G- 8	\mathbf{Br}
	G-3	OCF_3	G-8	CF_3
2 5	G-3	CN	G-8	OCF ₃
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
	G-4	ocf_3	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF_3		

R_a^2 =CF₃, R_b^2 =H, Y=CO₂Me, R^3 =i-Pr

10	<u>G</u>	R ¹	<u>G</u>	<u>R</u> 1
	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	${f Br}$
	G-1	OCF ₃	CH ₂ Ph	CF3
15	G-1	CN	CH ₂ Ph	OCF_3
	G-2	Cl	CH ₂ Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF_3
2 0	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

${\scriptstyle R_a^2=CF_3,\ R_b^2=H,\ Y=CO_2Me,\ R^3=CO_2Me}$

10	<u>G</u>	R ¹	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	Br
	G-1	OCF3	$\mathrm{CH_2Ph}$	CF_3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF3
	G-2	Cl	CH_2Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF3	G-8	CF_3
25	G-3	CN	G-8	OCF3
	G-4	C1	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

$$R_a^2$$
=CF₃, R_b^2 =H, Y=CO₂Me, R³=4-F-Ph

10	<u>G</u>	R ¹	<u>G</u>	\mathbb{R}^1
	G-1	C1	G-5	CN
	G-1	Br	CH_2Ph	Cl
	G-1	CF ₃	CH_2Ph	\mathbf{Br}
15	G-1	OCF3	CH ₂ Ph	CF3
	G-1	CN	CH_2Ph	OCF3
	G-2	Cl	CH_2Ph	CN
	G-2	\mathbf{Br}	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
20	G-2	OCF ₃	G-7	CF_3
	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
25	G-3	OCF3	G-8	CF_3
	G-3	CN	G-8	OCF ₃
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
30	G-4	OCF ₃	G-10	CF_3
	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
35	G-5	OCF_3		

•

${\scriptstyle R_a^2=Cl,\;R_b^2=Cl,\;Y=H,\;R^3=CO_2Me}$

10	<u>G</u>	<u>R</u> 1	<u>G</u>	R ¹
10	G-1	Cl	G-5	CN Cl
	G-1	Br	CH ₂ Ph	
	G-1	CF ₃	$\mathrm{CH_2Ph}$	Br
	G-1	OCF_3	$\mathrm{CH_2Ph}$	CF_3
15	G-1	CN	CH_2Ph	OCF_3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF_3	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF3	G-8	CF_3
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF_3	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF_3		
	G-5	OCF3		

R_a^2 =Cl, R_b^2 =Cl, Y=H, R^3 =i-Pr

10	<u>G</u>	<u>R</u> 1	G	<u>R</u> 1
	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	CH_2Ph	Br
	G-1	OCF ₃	$\mathrm{CH_2Ph}$	CF_3
15	G-1	CN	CH ₂ Ph	OCF ₃
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF_3	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
2 0	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF ₃
	G-4	CI	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	${f Br}$
	G-4	OCF ₃	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF ₃
	G-5	C1	G-10	CN
	G-5	Br		
	G-5	CF ₃		•
	G-5	OCF ₃		

è

R_a^2 =Cl, R_b^2 =F, Y=H, R^3 =CO₂Me

	<u>G</u>	$\underline{\mathbf{R}^1}$	G	\mathbb{R}^1
10	0.1	C1	G-5	CN
	G-1	Cl Br	CH ₂ Ph	Cl
	G-1		CH ₂ Ph	Br
	G-1	CF ₃	-	
	G-1	OCF_3	CH ₂ Ph	CF ₃
15	G-1	CN	$\mathtt{CH_2Ph}$	ocf_3
	G-2	C1	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	${\tt Br}$
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	C1	G-7	CN
	G-3	\mathbf{Br}	G-8	Cl
	G-3	CF3	G-8	\mathbf{Br}
	G-3	OCF3	G-8	CF_3
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl .	G-8	CN
	G-4	\mathbf{Br}	G-10	Cl
	G-4	$\mathbf{CF_3}$	G-10	${f Br}$
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	\mathbf{Br}		
	G-5	CF_3		
	G-5	OCF3		

$$R_a^2$$
=Cl, R_b^2 =F, Y=H, R^3 =i-Pr

10	<u>G</u>	<u>R</u> 1	<u>G</u>	<u>R</u> 1
	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	Br
	G-1	OCF ₃	$\mathrm{CH_2Ph}$	CF_3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF ₃
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
2 0	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF_3	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
2 5	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

$R_a^2 = R_b^2 = H$, Y=Me, $R^3 = Me$

	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	Br
			CH ₂ Ph	CF3
	G-1	OCF ₃	_	-
15	G-1	CN	CH ₂ Ph	OCF ₃
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF_3		
	G-5	OCF3		

WO 92/03421

5

$$R_a^2 = R_b^2 = H$$
, Y=Me, $R^3 = i$ -Pr

	\mathbf{G}	\mathbb{R}^1	\mathbf{G}	\mathbb{R}^1
10				
	G-1	Cl	G-5	CN
	G-1	Br	CH_2Ph	Cl
	G-1	CF ₃	CH_2Ph	\mathbf{Br}
	G-1	OCF ₃	CH ₂ Ph	$\mathbf{CF_3}$
15	G-1	CN	CH_2Ph	OCF_3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	Br
	G-2	OCF ₃	G-7	$\mathbf{CF_3}$
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	\mathbf{Br}	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF ₃
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

$$R_a^2 = R_b^2 = H$$
, Y=Me, $R^3 = CO_2$ Me

10	<u>G</u>	R ¹	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	Br
	G-1	OCF ₃	CH ₂ Ph	CF_3
15	G-1	CN	CH ₂ Ph	OCF ₃
20	G-2	Cl	CH ₂ Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	$\operatorname{\mathtt{Br}}$
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF3	G-8	\mathbf{Br}
	G-3	OCF3	G-8	CF ₃
2 5	G-3	CN	G-8	OCF ₃
•	G-4	Cl	G-8	CN
	G-4	Br	G-10	C1
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF3	G-10	CF_3
30	G-4	CN	G-10	OCF ₃
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF3		

WO 92/03421

5

$$R_a^2 = R_b^2 = H$$
, Y=Me, $R^3 = 4$ -F-Ph

10	<u>G</u>	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	CH_2Ph	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	Br
	G-1	OCF3	CH_2Ph	CF_3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Ci	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF3
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	\mathbf{Br}	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

 R_a^2 =Cl, R_b^2 =H, Y=Me, R^3 =Me

	G	\mathbb{R}^1	\mathbf{G}	\mathbb{R}^1
10		C 1	G-5	CN
	G-1	C1		Cl
	G-1	Br	CH ₂ Ph	
	G-1	CF_3	CH ₂ Ph	Br
	G-1	OCF_3	CH_2Ph	$\mathbf{CF_3}$
15	G-1	CN	$\mathrm{CH_2Ph}$	ocf3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF3	G-7	\mathbf{Br}
	G-2	OCF3	G-7	$\mathbf{CF_3}$
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF3	G-10	CF_3
3 0	G-4	CN	G-10	OCF3
	G-5	Cl	G-10	CN
	G-5	$\operatorname{\mathtt{Br}}$		
	G-5	CF ₃		
	G-5	OCF ₃		
35				

WO 92/03421

PCT/US91/05334

208

$$R_a^2$$
=Cl, R_b^2 =H, Y=Me, R^3 =i-Pr

	<u>G</u>	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10				
	G-1	Cl	G-5	CN
•	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	\mathbf{Br}
	G-1	OCF3	CH_2Ph	CF_3
15	G-1	CN	CH_2Ph	OCF_3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	$\mathbf{CF_3}$
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF_3	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	$\mathbf{CF_3}$
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	\mathbf{Br}	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G -5	CF_3		
	G-5	OCF ₃		
35				

5

$$R_a^2$$
=Cl, R_b^2 =H, Y=Me, R^3 =CO₂Me

10	<u>G</u>	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	CH_2Ph	\mathbf{Br}
	G-1	OCF ₃	CH_2Ph	CF_3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	${f Br}$	G-7	Cl
	G-2	CF3	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	\mathbf{Br}	G-8	Cl
	G-3	$\mathbf{CF_3}$	G-8	Br
	G-3	OCF ₃	G-8	$\mathbf{CF_3}$
25	G-3	CN	G-8	OCF_3
	G-4	C1	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
	G-4	OCF3	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	\mathbf{Br}		
	G-5	CF_3		
	G-5	OCF ₃		

WO 92/03421

5

$$R_a^2$$
=Cl, R_b^2 =H, Y=Me, R^3 =4-F-Ph

10	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
-0	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	Br
	G-1	OCF ₃	CH ₂ Ph	CF3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	Br
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF3	G-8	CF_3
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	Br
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF ₃
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF3		
0.5		-		

$$R_a^2 = CF_3$$
, $R_b^2 = H$, Y=Me, $R^3 = Me$

10	<u>G</u>	R ¹	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF_3	$\mathrm{CH_2Ph}$	Br
	G-1	OCF ₃	CH_2Ph	CF ₃
15	G-1	CN	CH_2Ph	OCF ₃
	G-2	Cl	$\mathtt{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	$\mathbf{CF_3}$	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	C1	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF3	G-8	\mathbf{Br}
	G-3	OCF3	G-8	CF_3
2 5	G-3	CN	G-8	ocf_3
	G-4	Cl	G-8	CN
	G-4	${f Br}$	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF_3	G-10	CF_3
30	G-4	CN	G-10	OCF3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF3		
	G-5	OCF_3		

$$R_a^2$$
=CF₃, R_b^2 =H, Y=Me, R^3 =i-Pr

10	<u>G</u>	<u>R</u> 1	<u>G</u>	<u>R</u> 1
10	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	Br
	G-1	OCF ₃	CH ₂ Ph	CF ₃
15	G-1	CN	$\mathtt{CH_2Ph}$	OCF_3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	$\mathbf{CF_3}$	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	$\mathbf{CF_3}$
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF_3	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	$\mathbf{CF_3}$	G-10	\mathbf{Br}
	G-4	OCF_3	G-10	CF_3
3 0	G-4	CN	G-10	ocf_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	$\mathbf{CF_3}$		
	G-5	OCF3		
~~				

R_a^2 =CF₃, R_b^2 =H, Y=Me, R^3 =CO₂Me

10	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1 G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	Br
	G-1	OCF ₃	CH ₂ Ph	CF_3
15	G-1	CN	CH_2Ph	OCF ₃
10	G-2	Cl	CH ₂ Ph	CN
	G-2	Br	G-7	Cl
	G-2 G-2	CF ₃	G-7	Br
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	Br
	G-3	OCF ₃	G-8	CF ₃
2 5	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF ₃
30	G-4	CN	G-10	ocf ₃
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF3		

WO 92/03421

5

 R_a^2 =CF₃, R_b^2 =H, Y=Me, R³=4-F-Ph

10	<u>G</u>	\mathbb{R}^1	$\underline{\mathbf{G}}$	<u>R</u> 1
10	C 1	C)	0.5	CNT
	G-1	Cl R-	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	$\mathtt{CH_2Ph}$	\mathbf{Br}
	G-1	OCF ₃	CH ₂ Ph	CF_3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF_3
	G-2	Cl	CH ₂ Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	${f Br}$
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	Br
	G-3	OCF ₃	G-8	CF_3
2 5	G-3	CN	G-8	OCF ₃
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl.
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		
0.5		_		

$$R_a^2 = R_b^2 = H$$
, Y=Me, $R^3 = 4$ -Cl-Ph

	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
10	0.1	CI	G-5	CN
	G-1 G-1	Cl Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	Br
			CH ₂ Ph	CF ₃
	G-1	OCF ₃	CH ₂ Ph	OCF3
15	G-1	CN		CN
	G-2	Cl	CH ₂ Ph	
	G-2	Br	G-7	Cl
	G-2	CF3	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	$\mathbf{CF_3}$
20	G-2	CN .	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF_3	G-8	$\mathbf{CF_3}$
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	$\operatorname{\mathtt{Br}}$	G-10	Cl
	G-4	CF3	G-10	Br
	G-4	OCF ₃	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF_3		
	G-5	OCF3		
35				

WO 92/03421

5

 R_a^2 =Cl, R_b^2 =H, Y=Me, R^3 =4-Cl-Ph

	\mathbf{G}	\mathbb{R}^1	\mathbf{G}	\mathbb{R}^1
10				
	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	\mathbf{Br}
	G-1	OCF_3	$\mathrm{CH_2Ph}$	$\mathbf{CF_3}$
1 5	G-1	CN	CH_2Ph	OCF ₃
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	Br
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF ₃
	G-3	C1	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
2 5	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	C1	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

 R_a^2 =CF₃, R_b^2 =H, Y=Me, R^3 =4-Cl-Ph

10	<u>G</u>	<u>R</u> 1	<u>G</u>	<u>R</u> 1
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	CH_2Ph	\mathbf{Br}
	G-1	OCF ₃	CH ₂ Ph	CF ₃
15	G-1	CN	CH ₂ Ph	OCF3
	G-2	Cl	${ m CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF3	G-7	\mathbf{Br}
	G-2	OCF3	G-7	$\mathbf{CF_3}$
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF3	G-8	CF_3
25	G-3	CN	G-8	OCF ₃
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	$\mathbf{CF_3}$	G-10	\mathbf{Br}
	G-4	OCF3	G-10	CF3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF3		
	G-5	OCF3		

WO 92/03421 PCT/US91/05334

218

5 TABLE 13

10

This Table contains a large number of compounds of Formula II. The first entry (where G is G-1, R^1 is Cl, $R_a{}^2$ = Cl, $R_b{}^2$ = H, X^1 = Cl and R^3 = i-Pr) actually specifies 159 separate and distinct compounds because the Key for Table 13 identifies additional substituent values a through fc for each Table 13 entry.

$${\scriptstyle R_a^2=Cl,\;R_b^2=H,\;X^1=Cl,\;R^3=i\text{-}Pr}$$

*

15	<u>G</u>	<u>R</u> 1	<u>G</u>	<u>R</u> 1
	G-1	Cl	G-5	OCF3
	G-1	Br	G-5	CN
	G-1	CF ₃	$\mathrm{CH_2Ph}$	C1
20	G-1	OCF ₃	CH ₂ Ph	\mathbf{Br}
	G-1	CN	CH ₂ Ph	CF_3
	G-2	Cl	CH_2Ph	OCF ₃
	G-2	Br	CH ₂ Ph	CN
	G-2	CF_3	G-7	Cl
2 5	G-2	OCF ₃	G-7	\mathbf{Br}
	G-2	CN	G-7	CF_3
	G-3	Cl	G-7	OCF ₃
	G-3	Br	G-7	CN
	G-3	CF ₃	G-8	Cl
30	G-3	OCF3	G-8	${f Br}$
	G-3	CN	G-8	CF_3
	G-4	Cl	G-8	OCF3
	G-4	Br	G-8	CN
	G-4	CF ₃	G-10	Cl
35	G-4	OCF ₃	G-10	\mathtt{Br}

WO 92/03421 PCT/US91/05334

219 \mathbb{R}^1 \mathbb{R}^1 G 5 \mathbf{G} CF₃ CN G-10 G-4 G-10 OCF3 G-5 Cl G-10 CN Br CF3 G-5 G-5 10

$$R_a^2$$
=Cl, R_b^2 =H, X^1 =Cl, R^3 =CO₂Me

10	\mathbf{G}	<u>R</u> 1	\mathbf{G}	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1 G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	Br
	G-1	OCF ₃	CH_2Ph	CF_3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	C1
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF3	G-8	Br
	G-3	OCF3	G-8	CF ₃
2 5	G-3	CN	G-8	OCF ₃
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃	•	
	G-5	OCF ₃		

 R_a^2 =Cl, R_b^2 =H, X^1 =Cl, R^3 =4-Cl-Ph

10	<u>G</u>	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1 G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH_2Ph	Br
		-	CH ₂ Ph	CF ₃
	G-1	OCF3		_
15	G-1	CN	CH ₂ Ph	OCF3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	$\mathbf{CF_3}$
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	Br
	G-3	OCF ₃	G-8	CF3
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	Br
	G-4	OCF ₃	G-10	CF ₃
30	G-4	CN	G-10	ocf ₃
	G-5	C1	G-10	CN
	G-5	Br		•
	G-5	CF ₃		
	G-5	OCF3		

WO 92/03421 PCT/US91/05334

.

$$R_a^2$$
=Cl, R_b^2 =H, X^1 =Cl, R^3 =4-F-Ph

10	<u>G</u>	<u>R</u> 1	<u>G</u>	<u>R</u> 1
	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	\mathbf{Br}
	G-1	OCF ₃	CH ₂ Ph	$\mathbf{CF_3}$
15	G-1	CN	CH ₂ Ph	OCF_3
	G-2	Cl	CH ₂ Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF3	G-7	Br
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF ₃
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF_3		
	G-5	OCF ₃		

$$R_a^2$$
=Cl, R_b^2 =H, X^1 =OAc, R^3 =i-Pr

	<u>G</u>	R ¹ .	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	. C1
	G-1	CF ₃	CH_2Ph	\mathbf{Br}
			CH ₂ Ph	CF ₃
	G-1	OCF ₃	_	OCF ₃
15	G-1	CN	CH ₂ Ph	
	G-2	Cl	CH ₂ Ph	CN
	G-2	${f Br}$	G-7	C1
	G-2	CF_3	G-7	\mathbf{Br}
	G-2	OCF_3	G-7	$\mathbf{CF_3}$
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF_3	G-8	\mathbf{Br}
	G-3	OCF3	G-8	$\mathbf{CF_3}$
25	G-3	CN	G-8	OCF3
	G-4	C1	G-8	CN
	G-4	\mathbf{Br}	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
	G-4	OCF_3	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	$\mathbf{CF_3}$		
	G-5	OCF ₃		

WO 92/03421 PCT/US91/05334

224

5

 R_a^2 =Cl, R_b^2 =H, X^1 =OAc, R^3 =CO₂Me

	G	<u>R</u> 1	\mathbf{G}	\mathbb{R}^1
10				
	G-1	Cl	G-5	CN
	G-1	Br	CH_2Ph	Cl
	G-1	CF_3	CH_2Ph	Br
	G-1	OCF ₃	CH_2Ph	CF_3
15	G-1	CN	CH_2Ph	OCF_3
	G-2	Cl	CH_2Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF3	G-8	Br
	G-3	OCF3	G-8	$\mathbf{CF_3}$
2 5	G-3	CN	G-8	OCF ₃
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF_3		
	G-5	OCF ₃		

$$R_a^2$$
=Cl, R_b^2 =H, X^1 =OAc, R^3 =4-Cl-Ph

10	<u>G</u>	R ¹	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	CH_2Ph	Cl
	G-1	CF ₃	CH_2Ph	\mathbf{Br}
	G-1	OCF3	$\mathrm{CH_2Ph}$	CF_3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF ₃
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	Br
	G-3	OCF3	G-8	CF_3
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	${f Br}$
	G-4	OCF3	G-10	CF ₃
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

WO 92/03421 PCT/US91/05334

$$R_a^2$$
=Cl, R_b^2 =H, X^1 =OAc, R^3 =4-F-Ph

10	<u>G</u>	<u>R</u> 1	<u>G</u>	<u>R</u> 1
	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	CH ₂ Ph	\mathbf{Br}
	G-1	OCF ₃	CH ₂ Ph	CF_3
15	G-1	CN	CH ₂ Ph	OCF3
	G-2	Cl	CH ₂ Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF_3	G-7	Br
	G-2	OCF3	G-7	CF_3
2 0	G-2	CN	G-7	ocf_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	Br
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	${f Br}$
	G-4	OCF3	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

$$R_a^2$$
=CF₃, R_b^2 =H, X^1 =Cl, R^3 =i-Pr

10	<u>G</u>	\mathbb{R}^1	<u>G</u>	R ¹
10	G-1	Cl	G-5	CN
	G-1	Br	CH_2Ph	Cl
	G-1	CF ₃	CH_2Ph	\mathbf{Br}
	G-1	OCF3	CH_2Ph	CF_3
15	G-1	CN	CH_2Ph	OCF ₃
	G-2	Cl	CH_2Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF3	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF3	G-10	CF_3
30	G-4	CN	G-10	OCF ₃
	G-5	Cl	G-10	CN
	G-5	Br	•	
	G-5	CF3		
	G-5	OCF3		

$$R_a^2$$
=CF₃, R_b^2 =H, X¹=Cl, R³=CO₂Me

	<u>G</u>	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10				
	G-1	Cl	G-5	CN
	G-1	Br	CH_2Ph	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	\mathbf{Br}
	G-1	OCF3	$\mathtt{CH_2Ph}$	$\mathbf{CF_3}$
1 5	G-1	CN	CH_2Ph	OCF_3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	C1
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G- 8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF_3		
	G-5	OCF_3		

$$R_a^2$$
=CF₃, R_b^2 =H, X^1 =Cl, R^3 =4-Cl-Ph

	G	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10				
	G-1	Cl	G -5	CN
	G-1	Br	${ m CH_2Ph}$	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	\mathbf{Br}
	G-1	OCF ₃	$\mathrm{CH_2Ph}$	CF_3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF_3
	G-2	C1	$\mathrm{CH_2Ph}$	CN
	G-2	${f Br}$	G-7	Cl
	G-2	CF_3	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF_3	G-8	\mathbf{Br}
	G-3	OCF_3	G-8	$\mathbf{CF_3}$
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	$\mathbf{CF_3}$	G-10	\mathbf{Br}
	G-4	OCF_3	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF ₃
	G-5	Cl	G-10	CN
	G-5	\mathbf{Br}		
	G-5	CF_3	•	
	G-5	OCF_3		

WO 92/03421 PCT/US91/05334

$$R_a^2$$
=CF₃, R_b^2 =H, X^1 =Cl, R^3 =4-F-Ph

10	<u>G</u>	<u>R</u> ¹	<u>G</u>	\mathbb{R}^1
	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	\mathbf{Br}
	G-1	OCF ₃	CH ₂ Ph	CF_3
15	G-1	CN	CH ₂ Ph	OCF3
	G-2	Cl	CH ₂ Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF_3
2 0	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	Br
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	ocf_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF3	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF_3		
	G-5	OCF ₃		

$$R_a^2$$
=CF₃, R_b^2 =H, X¹=OAc, R³=4-F-Ph

10	<u>G</u>	\mathbb{R}^1	G	R ¹
10	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	Br
	G-1	OCF ₃	CH ₂ Ph	CF_3
15	G-1	CN	CH ₂ Ph	OCF ₃
	G-2	Cl	CH ₂ Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	CI
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G- 5	OCF3		

WO 92/03421

$$R_a^2$$
=CF₃, R_b^2 =H, X¹=OAc, R³=4-Cl-Ph

	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
10	0.1	C)	0.5	CINT
	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH_2Ph	Br
	G-1	OCF ₃	CH_2Ph	$\mathbf{CF_3}$
15	G-1	CN	CH ₂ Ph	OCF_3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	$\mathbf{CF_3}$
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF_3		
	G-5	OCF ₃		

$$R_a^2$$
=CF₃, R_b^2 =H, X¹=OAc, R^3 =CO₂Me

10	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	Cl
	G-1	CF ₃	CH ₂ Ph	Br
	G-1	OCF ₃	CH ₂ Ph	CF_3
1 5	G-1	CN	CH ₂ Ph	OCF_3
	G-2	Cl	CH_2Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	${\tt Br}$
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF_3	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	$\mathbf{CF_3}$
25	G-3	CN	G-8	OCF_3
	G-4	C1	G-8	CN
	G-4	Br	G-10	C1
	G-4	CF ₃	G-10	Br
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	ocf_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF3		
	G-5	OCF ₃		

$$R_a^2$$
=CF₃, R_b^2 =H, X¹=OAc, R³=i-Pr

10	\mathbf{G}	<u>R</u> 1	<u>G</u>	<u>R</u> 1
10	G-1	Cl	CE	CN
	G-1 G-1	Br	G-5 CHaPh	Cl
			CH ₂ Ph	
	G-1	CF ₃	CH ₂ Ph	Br
	G-1	OCF_3	CH_2Ph	CF_3
15	G-1	CN	CH ₂ Ph	OCF_3
	G-2	Cl	${ m CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	$\mathbf{CF_3}$
2 0	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

$$R_a^2$$
=H, R_b^2 =F, X^1 =Cl, R^3 =i-Pr

	\mathbf{G}	\mathbb{R}^1	\mathbf{G}	\mathbb{R}^1
10	~ 1	G1	O 5	CN
	G-1	Cl Por	G-5 CH ₂ Ph	Cl
	G-1	Br		
	G-1	$\mathbf{CF_3}$	CH ₂ Ph	Br
	G-1	OCF_3	$\mathrm{CH_2Ph}$	CF_3
15	G-1	CN	$\mathtt{CH_2Ph}$	ocf_3
	G-2	Cl	${ m CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	${\tt Br}$
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF3	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	$\mathbf{CF_3}$
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	\mathbf{Br}	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
	G-4	OCF_3	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	\mathbf{Br}		
	G-5	$\mathbf{CF_3}$		
	G-5	OCF_3		

PCT/US91/05334

5

R_a^2 =H, R_b^2 =F, X^1 =Cl, R^3 =CO₂Me

40	G	\mathbb{R}^1	G	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	Br
	G-1	OCF3	CH ₂ Ph	$\mathbf{CF_3}$
15	G-1	CN	CH ₂ Ph	OCF ₃
	G-2	Cl	CH_2Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	CF_3
20	G-2	CN	G-7	OCF ₃
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF ₃
2 5	G-3	CN	G-8	OCF ₃
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF ₃
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

$$R_a^2$$
=H, R_b^2 =F, X^1 =Cl, R^3 =4-Cl-Ph

10	<u>G</u>	<u>R</u> 1	<u>G</u>	<u>R</u> 1
10	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF3	CH ₂ Ph	Br
	G-1	OCF3	CH ₂ Ph	CF_3
15	G-1	CN	CH ₂ Ph	OCF3
	G-2	Cl	CH ₂ Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF3	G-7	$\mathbf{CF_3}$
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	\mathbf{CF}_{3}
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF3	G-10	$\mathbf{CF}_{\mathfrak{F}}$
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G -5	OCF3		

WO 92/03421

$$R_a^2$$
=H, R_b^2 =F, X^1 =Cl, R^3 =4-F-Ph

10	<u>G</u>	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1 G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	CH ₂ Ph	Br
			_	
	G-1	OCF ₃	CH ₂ Ph	CF ₃
15	G-1	CN	CH ₂ Ph	OCF_3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF_3	G-7	CF_3
2 0	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	$\mathbf{CF_3}$
2 5	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
3 0	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF3		
	G-5	OCF3	,	

$$R_a^2$$
=H, R_b^2 =F, X^1 =OAc, R^3 =4-F-Ph

10	. G	\mathbb{R}^1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	CH_2Ph	Cl
	G-1	CF ₃	CH ₂ Ph	Br
	G-1	OCF3	CH_2Ph	CF ₃
15	G-1	CN	CH ₂ Ph	OCF3
	G-2	Cl	${ m CH_2Ph}$	CN
	G-2	Br	G-7	C1
	G-2	CF ₃	G-7	Br
	G-2	OCF ₃	G-7	$\mathbf{CF_3}$
20	G-2	CN	G-7	OCF_3
	G-3	Cl ·	G-7	CN
	G-3	${f Br}$	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF ₃
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	\mathbf{Br}	G-10	Cl
	G-4	$\mathbf{CF_3}$	G-10	${f Br}$
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF_3
	G-5	Cl	G-10	CN
	G-5	\mathbf{Br}		
	G-5	$\mathbf{CF_3}$		
	G-5	OCF_3		

$$R_a^2$$
=H, R_b^2 =F, X^1 =OAc, R^3 =4-Cl-Ph

	<u>G</u>	R ¹	<u>G</u>	\mathbb{R}^1
10	0.1	CI.	0.5	CINT
	G-1	Cl Pro	G-5	CN Cl
	G-1	Br	CH ₂ Ph	
	G-1	CF ₃	CH ₂ Ph	\mathbf{Br}
	G-1	OCF_3	CH ₂ Ph	CF_3
15	G-1	CN	CH ₂ Ph	OCF_3
	G-2	Cl	CH_2Ph	CN
	G-2	Br	G-7	Cl
	G-2	CF ₃	G-7	\mathbf{Br}
	G-2	OCF ₃	G-7	CF ₃
20	G-2	CN	G-7	OCF_3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF_3
	G-4	Cl	G-8	CN
	G-4	\mathbf{Br}	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	$\mathbf{CF_3}$
30	G-4	CN	G-10	OCF ₃
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

 ${\scriptstyle R_a^2=H,\;R_b^2=F,\;X^1=OAc,\;R^3=CO_2Me}$

10	<u>G</u>	<u>R</u> 1	<u>G</u>	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	$\mathrm{CH_2Ph}$	CI
	G-1	CF3	$\mathrm{CH_2Ph}$	${\tt Br}$
	G-1	OCF3	$\mathrm{CH_2Ph}$	CF3
15	G-1	CN	$\mathrm{CH_2Ph}$	OCF3
	G-2	Cl	$\mathrm{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF3	G-7	Br
	G-2	OCF3	G-7	$\mathbf{CF_3}$
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	Br	G-8	Cl
	G-3	CF ₃	G-8	\mathbf{Br}
	G-3	OCF3	G-8	CF_3
25	G-3	CN	G-8	OCF3
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF ₃	G-10	\mathbf{Br}
	G-4	OCF3	G-10	CF_3
30	G-4	CN	G-10	OCF3
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF_3		
	G-5	OCF3		

$$R_a^2$$
=H, R_b^2 =F, X^1 =OAc, R^3 =i-Pr

10	<u>G</u>	\mathbb{R}^1	<u>G</u> .	\mathbb{R}^1
10	G-1	Cl	G-5	CN
	G-1	Br	CH ₂ Ph	Cl
	G-1	CF ₃	$\mathrm{CH_2Ph}$	\mathbf{Br}
	G-1	OCF3	$\mathtt{CH_2Ph}$	$\mathbf{CF_3}$
15	G-1	CN	$\mathtt{CH_2Ph}$	OCF ₃
	G-2	Cl	$\mathtt{CH_2Ph}$	CN
	G-2	Br	G-7	Cl
	G-2	CF3	G-7	Br
	G-2	OCF ₃	G-7	CF3
20	G-2	CN	G-7	OCF3
	G-3	Cl	G-7	CN
	G-3	\mathbf{Br}	G-8	Cl
	G-3	CF_3	G-8	\mathbf{Br}
	G-3	OCF ₃	G-8	CF_3
25	G-3	CN	G-8	OCF ₃
	G-4	Cl	G-8	CN
	G-4	Br	G-10	Cl
	G-4	CF_3	G-10	\mathbf{Br}
	G-4	OCF ₃	G-10	CF_3
30	G-4	CN	G-10	OCF ₃
	G-5	Cl	G-10	CN
	G-5	Br		
	G-5	CF ₃		
	G-5	OCF ₃		

Key for Table 14

Group	<u>R</u> 1	
a	CF3	
ь	OCF ₃	
c	Cl	
d	Br	

Compounds of Table 14 wherein A, R², and R³ are as set out therein can be prepared having the recited values of Groups a through d. That is, for each value of A, R², and R³ in Table 1, R¹ can be CF₃, OCF₃, Cl or Br. All of said compounds are specifically included within the scope of this invention.

15

20

10

TABLE 14

This Table contains a large number of compounds of Formula I in a format adopted to avoid mechanical reproduction of substituent values that do not vary. For example, the first table entry (where A is CH₂CH₂, R² is 4-F and R³ is CN) actually specifies 4 separate and distinct compounds because the key for Table 14 identifies additional substituent values of R¹, namely Group a through d for each Table 14 entry.

A is CH₂CH₂; A is CH₂CH₂; A is CH₂CH₂; R^2 is 4-F \mathbb{R}^2 is 5-Cl \mathbb{R}^2 is 5-F \mathbb{R}^3 \mathbb{R}^3 \mathbb{R}^3 CN CN CN NO_2 NO_2 NO_2 N_3 N_3 N_3 \mathbf{Br} \mathbf{Br} \mathbf{Br} NMe₂ NMe₂ NMe₂ NHCO₂Me NHCO₂Me NHCO₂Me epoxyethyl epoxyethyl epoxyethyl epoxyisopropenyl epoxyisopropenyl epoxyisopropenyl SiMea SiMe3 SiMe₃ SMe SMe SMe S(O)Me S(O)Me S(O)Me SO₂Me SO₂Me SO₂Me $P(O)(OMe)_2$ $P(O)(OMe)_2$ $P(O)(OMe)_2$ CH₂C(O)NMe₂ CH₂C(O)NMe₂ CH₂C(O)NMe₂ CH_2SMe CH₂SMe CH₂SMe CH2OCH2CF3 CH2OCH2CF3 CH2OCH2CF3 CH₂SiMe₃ CH₂SiMe₃ CH₂SiMe₃ CH₂NMe₂ CH₂NMe₂ CH2NMe2 CH₂-2-pyridyl CH₂-2-pyridyl CH₂-2-pyridyl 2-pyridyl 2-pyridyl 2-pyridyl C(H)=NOCH₃ C(H)=NOCH₃ $C(H)=NOCH_3$ 3-pyridyl 3-pyridyl 3-pyridyl 2-thienyl 2-thienyl 2-thienyl 3-thienyl 3-thienyl 3-thienyl 2-oxazolinyl 2-oxazolinyl 2-oxazolinyl 2-(1,3-dioxolanyl) 2-(1,3-dioxolanyl) 2-(1,3-dioxolanyl)

5			
	A is CH_2CH_2 ;	A is OCH_2 ;	A is OCH_2 ;
	\mathbb{R}^2 is 5-CF ₃ ;	\mathbb{R}^2 is 4-F;	$ m R^2$ is 5-F
	R ³	\mathbb{R}^3	<u>R</u> 3
	CN	CN	CN
	NO_2	NO_2	NO_2
	N ₃	N_3	N_3
	Br NMe ₂	Br NMe ₂	Br NMe ₂
	NHCO ₂ Me	NHCO ₂ Me	NHCO ₂ Me
	epoxyethyl	epoxyethyl	epoxyethyl
	epoxyisopropenyl SiMe3	epoxyisopropenyl SiMe3	epoxyisopropenyl SiMe3
	SMe	SMe	SMe
	S(O)Me	S(O)Me	S(O)Me
	SO ₂ Me	SO_2Me	SO ₂ Me
	$P(O)(OMe)_2$	$P(O)(OMe)_2$	$P(O)(OMe)_2$
	CH ₂ C(O)NMe ₂	$CH_2C(O)NMe_2$	$CH_2C(O)NMe_2$
	CH ₂ SMe	$\mathrm{CH}_2\mathrm{SMe}$	CH ₂ SMe
	CH ₂ OCH ₂ CF ₃	$\mathrm{CH_{2}OCH_{2}CF_{3}}$	$CH_2OCH_2CF_3$
	CH ₂ SiMe ₃	$ m CH_2SiMe_3$	CH ₂ SiMe ₃
	CH ₂ NMe ₂	$\mathrm{CH_2NMe_2}$	${ m CH_2NMe_2}$
	CH ₂ -2-pyridyl	CH ₂ -2-pyridyl	$ ext{CH}_2 ext{-2-pyridyl}$
	2-pyridyl 3-pyridyl	2-pyridyl	2-pyridyl
		3-pyridyl	3-pyridyl
	2-thienyl	2-thienyl	2-thienyl
	3-thienyl	3-thienyl	3-thienyl
	2-oxazolinyl	2-oxazolinyl	2-oxazolinyl
	2-(1,3-dioxolanyl)	2-(1,3-dioxolanyl)	2-(1,3-dioxolanyl)

ś

5

 $A \ is \ OCH_2;$

 \mathbb{R}^2 is 5-Cl;

A is OCH₂;

 \mathbb{R}^2 is 5-CF3;

A is NHCH2;

 \mathbb{R}^2 is 4-F

R³ CN

CN NO₂

Ng Br

NMe₂ NHCO₂Me epoxyethyl

epoxyisopropenyl

SiMe₃
SMe
S(O)Me
SO₂Me
P(O)(OMe)₂

CH₂C(O)NMe₂

CH₂SMe

R³ CN

NO₂

Br NMe₂

NHCO₂Me epoxyethyl

epoxyisopropenyl

SiMe₃ SMe S(O)Me SO₂Me

P(O)(OMe)₂ CH₂C(O)NMe₂

CH₂SMe

R³ CN

NO₂ NMe₃

NHCO₂Me epoxyethyl SiMe₃

SMe

 $\begin{array}{c} P(O)(OMe)_2 \\ CH_2SMe \end{array}$

CH₂OCH₂CF₃ CH₂SiMe₃ 2-pyridyl 2-thienyl

A is OCH_2 ; R^2 is 5-Cl;

 \mathbb{R}^3

CH₂OCH₂CF₃ CH₂SiMe₃ CH₂NMe₂ CH₂-2-pyridyl

2-pyridyl 3-pyridyl 2-thienyl

3-thienyl 2-oxazolinyl

2-(1,3-dioxolanyl) C(H)=NOCH₃

C(CH₃)=NOCH₃ C(Cl)=NOCH₃

C(CN)=NOCH₃

C(OCH₃)=NOCH₃

A is OCH₂;

 R^2 is 5-CF3;

 \mathbb{R}^3

CH₂OCH₂CF₃ CH₂SiMe₃ CH₂NMe₂ CH₂-2-pyridyl

2-pyridyl 3-pyridyl 2-thienyl

3-thienyl 2-oxazolinyl

2-(1,3-dioxolanyl) C(SCH₃)=N(OCH₃)

C(H)=NOCH₃ C(H)=NOCH₂CH₃

C(CH₃)=NOCH₃

C(H)=NOH

A is NHCH₂;

R₂ is 5-F;

 \mathbb{R}^3 $\mathbb{C}\mathbb{N}$ $\mathbb{N}\mathbb{O}_2$ $\mathbb{N}\mathbb{M}\mathbb{O}_2$

NHCO₂Me

epoxyethyl SiMe3

SMe

P(O)(OMe)₂ CH₂SMe

CH₂OCH₂CF₃ CH₂SiMe₃

2-pyridyl 2-thienyl 5 A is NHCH₂; A is N(Me)CH₂; A is N(Me)CH₂; R^2 is 5-Cl; R^2 is 4-F; R^2 is 5-Cl

 \mathbb{R}^3 \mathbb{R}^3 \mathbb{R}^3 CN CN CN NO_2 NO_2 NO_2 NMe₂ NMe₂ NMe_2 NHCO₂Me NHCO₂Me NHCO₂Me epoxyethyl epoxyethyl epoxyethyl SiMe₃ SiMe₃ SiMe₃ SMe SMe SMe $P(O)(OMe)_2$ $P(O)(OMe)_2$ $P(O)(OMe)_2$ CH₂SMe CH_2SMe CH₂SMe CH2OCH2CF3 CH₂OCH₂CF₃ CH₂OCH₂CF₃ CH₂SiMe₃ CH₂SiMe₃ CH₂SiMe₃ 2-pyridyl 2-pyridyl 2-pyridyl 2-thienyl 2-thienyl 2-thienyl

5 A is N(Me)CH2; A is $N(Me)CH_2$; A is NHCH2; \mathbb{R}^2 is 5-CF3; \mathbb{R}^2 is 5-CF3; \mathbb{R}^2 is 5-F; \mathbb{R}^3 \mathbb{R}^3 \mathbb{R}^3 CN CN CN NO_2 NO_2 NO_2 NMe₂ NMe₂ NMe₂ ${
m NHCO_2Me}$ NHCO₂Me NHCO₂Me epoxyethyl epoxyethyl epoxyethyl SiMe₃ SiMe3 SiMe3 SMe SMe SMe $P(O)(OMe)_2$ $P(O)(OMe)_2$ $P(O)(OMe)_2$ CH₂SMe CH₂SMe CH₂SMe CH₂OCH₂CF₃ CH2OCH2CF3 CH2OCH2CF3 CH₂SiMe₃ CH₂SiMe₃ CH₂SiMe₃ 2-pyridyl 2-pyridyl 2-pyridyl 2-thienyl 2-thienyl 2-thienyl C(CH₃)=NOCH₃ C(OCH₃)=NOCH₃ C(H)=NOCH₃

WO 92/03421 PCT/US91/05334

5 Formulation and Use

10

15

20

The compounds of this invention will generally be used in formulation with an agriculturally suitable carrier comprising a liquid or solid diluent or an organic solvent. Useful formulations of the compounds of Formulas I and II can be prepared in conventional ways. They include dusts, granules, baits, pellets, solutions, suspensions, emulsions, wettable powders, emulsifiable concentrates, dry flowables and the like. Many of these can be applied directly. Sprayable formulations can be extended in suitable media and used at spray volumes of from about one to several hundred liters per hectare. High strength compositions are primarily used as intermediates for further formulation. The formulations, broadly, contain from less than about 1% to 99% by weight of active ingredient(s) and at least one of a) about 0.1% to 20% surfactant(s) and b) about 5% to 99% solid or liquid diluent(s). More specifically, they will contain effective amounts of these ingredients in the following approximate proportions:

		Percent by Weight			
		Active		~ • · · · · · ·	
95		Ingredient	Diluent(s)	Surfactant(s)	
25	Wettable Powders	25-90	0-74	1-10	
30	Oil Suspensions, Emulsions, Solutions, (including Emulsifiable Concentrates)	5-50	40-95	0-15	
	Dusts	1-25	70-99	0-5	
35	Granules, Baits and Pellets	0.01-95	5-99 .	0-15	
40	High Strength Compositions	90-99	0-10	0-2	

10

15

20

25

30

Lower or higher levels of active ingredient can, of course, be present depending on the intended use and the physical properties of the compound. Higher ratios of surfactant to active ingredient are sometimes desirable, and are achieved by incorporation into the formulation or by tank mixing.

Typical solid diluents are described in Watkins, et al., "Handbook of Insecticide Dust Diluents and Carriers", 2nd Ed., Dorland Books, Caldwell, New Jersey. The more absorptive diluents are preferred for wettable powders and the denser ones for dusts. Typical liquid diluents and solvents are described in Marsden, "Solvents Guide," 2nd Ed., Interscience, New York, 1950. Solubility under 0.1% is preferred for suspension concentrates; solution concentrates are preferably stable against phase separation at 0°C. "McCutcheon's Detergents and Emulsifiers Annual", Allured Publ. Corp., Ridgewood, New Jersey, as well as Sisely and Wood, "Encyclopedia of Surface Active Agents", Chemical Publ. Co., Inc., New York, 1964, list surfactants and recommended uses. All formulations can contain minor amounts of additives to reduce foam, caking, corrosion, microbiological growth, etc. Preferably, ingredients should be approved by the U.S. Environmental Protection Agency for the use intended.

The methods of making such compositions are well known. Solutions are prepared by simply mixing the ingredients. Fine solid compositions are made by blending and, usually, grinding as in a hammer or fluid energy mill. Suspensions are prepared by wet milling (see, for example, U.S. 3,060,084). Granules and pellets can be made by spraying the active material upon preformed granular carriers or by agglomeration techniques. See J. E. Browning, "Agglomeration", Chemical Engineering, December 4, 1967, pages 147 and following, and "Perry's Chemical Engineer's Handbook", 4th Ed., McGraw-Hill, New York, 1963, pages 8 to 59 and following.

25

252

5	Example A				
	Emulsifiable Concentrate				
	7-chloro-,3a-cyano-3a,4-dihydro-N-				
	[4-(trifluoromethyl)phenyl][1]benzopyrano-				
	[4,3-C]pyrazole-2(3H)-carboxamide	20%			
10	blend of oil soluble sulfonates				
	and polyoxyethylene ethers	10%			
	isophorone	70%			
	The ingredients are combined and stir	red with gentle warming to			
	speed solution. A fine screen filter is included	l in packaging operation to			

Example B

insure the absence of any extraneous undissolved material in the product.

Wettable Powder

7-chloro-,3a-cyano-3a,4-dihydro-N-

20 [4-(trifluoromethyl)phenyl][1]benzopyrano-

[4,3-C]pyrazole-2(3H)-carboxamide30%sodium alkylnaphthalenesulfonate2%sodium ligninsulfonate2%synthetic amorphous silica3%kaolinite63%

The active ingredient is mixed with the inert materials in a blender. After grinding in a hammer-mill, the material is re-blended and sifted through a 50 mesh screen.

30 Example C

Dust

Wettable powder of Example B 10% pyrophyllite (powder) 90%

The wettable powder and the pyrophyllite diluent are thoroughly blended and then packaged. The product is suitable for use as a dust.

15

25

Example D

Granule

7-chloro-2,3,3a,4-tetrahydro-3a-oxiranyl-N-[4-(trifluoromethyl)phenyl][1]benzopyrano-

[4,3-C]pyrazole-2-carboxamide

10%

10 attapulgite granules (low volative

matter, 0.71/0.30 mm; U.S.S. No.

25-50 sieves)

90%

The active ingredient is dissolved in a volatile solvent such as acetone and sprayed upon dedusted and pre-warmed attapulgite granules in a double cone blender. The acetone is then driven off by heating. The granules are then allowed to cool and are packaged.

Example E

Granule

20 Wettable powder of Example B 15% gypsum 69% potassium sulfate 16%

The ingredients are blended in a rotating mixer and water sprayed on to accomplish granulation. When most of the material has reached the desired range of 0.1 to 0.42 mm (U.S.S. No. 18 to 40 sieves), the granules are removed, dried, and screened. Oversize material is crushed to produce additional material in the desired range. These granules contain 4.5% active ingredient.

59.0%

5	Example F	
	Solution	
	7-chloro-2,3,3a,4-tetrahydro-3a-oxiranyl-	
	N-[4-(trifluoromethyl)phenyl][1]benzopyrano-	
	[4,3-C]pyrazole-2-carboxamide	25%
10	N-methyl-pyrrolidone	75%
	The ingredients are combined and stirred t	o produce a solution
	suitable for direct, low volume application.	
	Example G	
15	Aqueous Suspension	
	3a-(5-chloro-2-thienyl)-3a,4-dihydro-F-	
	(trifluoromethyl)-N-[4-(trifluoromethyl)phenyl][1]b	enzopyrano-
	[4,3-C]pyrazole-2(3H)-carboxamide	40%
	polyacrylic acid thickener	0.3%
20	dodecyclophenol polyethylene glycol	
	ether	0.5%
	disodium phosphate	1.0%
	monosodium phosphate	0.5%
	polyvinyl alcohol	1.0%
2 5	water	56.7%
	The ingredients are blended and ground toge	
	produce particles substantially all under 5 microns	in size.
	Example H	
30	Oil Suspension	
	7-chloro-2,3,3a,4-tetrahydro-3a-oxiranyl-	
	N-[4-(trifluoromethyl)phenyl][1]benzopyrano-	
	[4,3-C]pyrazole-2-carboxamide	35.0%
	blend of polyalcohol carboxylic	6.0%
35	esters and oil soluble petroleum	
	•	

sulfonates

xylene range solvent

The ingredients are combined and ground together in a sand mill to produce particles substantially all below 5 microns. The product can be used directly, extended with oils, or emulsified in water.

10

20

25

Example I

Bait Granules

2-(2,3,3a,8a-tetrahydrocyclopent[a]inden-8(1H)-ylidene-N-[4-(trifluoromethyl)phenyl]hydrazine-carboxamide

25%

15 blend of polyethoxylated nonyl-

9.0%

phenols and sodium dodecyl-

benzene sulfonates

88.0%

ground up corn cobs

The active ingredient and surfactant blend are dissolved in a suitable solvent such as acetone and sprayed onto the ground corn cobs. The granules are then dried and packaged.

Compounds of Formulas I and II can also be mixed with one or more other insecticides, fungicides, nematocides, bactericides, acaricides, or other biologically active compounds to form a multi-component pesticide giving an even broader spectrum of effective agricultural protection. Examples of other agricultural protectants with which compounds of this invention can be formulated are:

Insecticides:

30 3-hydroxy-N-methylcrotonamide(dimethylphosphate)ester (monocrotophos)

methylcarbamic acid, ester with 2,3-dihydro-2,2-

dimethyl-7-benzofuranol (carbofuran)

 $O-[2,4,5-trichloro-\alpha-(chloromethyl)benzyl]$ phosphoric

acid, O',O'-dimethyl ester (tetrachlorvinphos)

2-mercaptosuccinic acid, diethyl ester, S-ester with thionophosphoric acid, dimethyl ester (malathion)

phosphorothioic acid, O,O-dimethyl, O-p-nitrophenyl 5 ester (methyl parathion) methylcarbamic acid, ester with α -naphthol (carbaryl) methyl O-(methylcarbamoyl)thiolacetohydroxamate (methomyl) 10. N'-(4-chloro-o-tolyl)-N,N-dimethylformamidine (chlordimeform) O,O-diethyl-O-(2-isopropyl-4-methyl-6-pyrimidylphosphorothioate (diazinon) octachlorocamphene (toxaphene) O-ethyl-O-p-nitrophenyl phenylphosphonothioate (EPN) 15 (S)- α -cyano- \underline{m} -phenoxybenzyl(1R,3R)-3-(2,2-dibromovinyl)-2,2dimethylcyclopropanecarboxylate (deltamethrin) Methyl-N', N'-dimethyl-N-[(methylcarbamoyl)oxy]-1thioox amimidate (oxamyl) 20 cyano(3-phenoxyphenyl)-methyl-4-chloro-α-(1-methylethyl)benzeneacetate (fenvalerate) (3-phenoxyphenyl)methyl(+)-cis,trans-3-(2,2-dichloro ethenyl)-2,2-dimethylcyclopropanecarboxylate (permethrin) 25 α-cyano-3-phenoxybenzyl 3-(2,2-dichlorovinyl)-2,2dimethylcyclopropane carboxylate (cypermethrin) O-ethyl-S-(p-chlorophenyl)ethylphosphonodithioate (profenofos)

O-ethyl-O-[4-(methylthio)-phenyl]-S-n-propyl ester

phosphorothiolothionic acid,

(sulprofos).

30

Additional insecticides are listed hereafter by their common names: triflumuron, diflubenzuron, methoprene, buprofezin, thiodicarb, acephate, azinphosmethyl, chlorpyrifos, dimethoate, fonophos, isofenphos, methidathion, methamidiphos, monocrotphos, phosmet, phosphamidon, phosalone, pirimicarb, phorate, terbufos, trichlorfon, methoxychlor,

bifenthrin, biphenate, cyfluthrin, fenpropathrin, fluvalinate, flucythrinate, tralomethrin, metal-dehyde and rotenone.

Fungicides:

methyl 2-benzimidazolecarbamate (carbendazim)

tetramethylthiuram disulfide (thiuram)
n-dodecylguanidine acetate (dodine)

manganese ethylenebisdithiocarbamate (maneb) 1,4-dichloro-2,5-dimethoxybenzene (chloroneb)

methyl 1-(butylcarbamoyl)-2-benzimidazolecarbamate

15 (benomyl)

1-[2-(2,4-dichlorophenyl)-4-propyl-1,3-dioxolan-2ylmethyl]-1H-1,2,4-triazole (propiconazole)

2-cyano-N-ethylcarbamoyl-2-methoxyiminoacetamide (cymoxanil)

20 1-(4-chlorophenoxy)-3,3-dimethyl-1-(1H-1,2,4-triazol-1-yl)-2-butanone (triadimefon)

N-(trichloromethylthio)tetrahydrophthalimide (captan)

N-(trichloromethylthio)phthalimide (folpet)

1-[[[bis(4-fluorophenyl)][methyl]silyl]methyl]-1H-

25 1,2,4-triazole.

Nematocides:

S-methyl 1-(dimethylcarbamoyl)-N-(methylcarbamoyloxy)thioformimidate

30 S-methyl 1-carbamoyl-N-(methylcarbamoyloxy)thioformimidate

N-isopropylphosphoramidic acid O-ethyl O'-[4-(methyl-thio)-m-tolyl]diester (fenamiphos)

35 Bactericides:

tribasic copper sulfate streptomycin sulfate WO 92/03421 PCT/US91/05334

258

ۇ

5

Acaricides:

senecioic acid, ester with 2-<u>sec</u>-butyl-4,6-dinitrophenol (binapacryl)

6-methyl-1,3-cithiolo[4,5-B]quinoxalin-2-one (oxythioquinox)

10 ethyl 4,4'-dichlorobenzilate (chlorobenzilate)

1,1-bis(p-chlorophenyl)-2,2,2-trichloroethanol (dicofol)

bis(pentachloro-2,4-cyclopentadien-1-yl) (dienochlor)

tricyclohexyltin hydroxide (cyhexatin)

trans-5-(4-chlorophenyl)-N-cyclohexyl-4-methyl-2-oxo-

15 thiazolidine-3-carboxamide (hexythiazox)

amitraz

propargite

fenbutatin-oxide

20 Biological

Bacillus thuringiensis

Avermectin B.

Utility

25

30

35

The compounds of this invention exhibit activity against a wide spectrum of foliar and soil inhabiting arthropods which are pests of growing and stored agronomic crops, forestry, greenhouse crops, ornamentals, nursery crops, stored food and fiber products, livestock, household, and public and animal health. Those skilled in the art will recognize that not all compounds are equally effective against all pests but the compounds of this invention display activity against economically important agronomic, forestry, greenhouse, ornamental food and fiber product, stored product, domestic structure, and nursery pests, such as:

larvae of the order <u>Lepidoptera</u> including fall and beet armyworm and other <u>Spodoptera spp.</u>, tobacco budworm, corn earworm and other <u>Heliothis spp.</u>, European corn borer, navel

10	orangeworm, stalk/stem borers and other pyralids, cabbage and soybean loopers and other loopers, codling moth, grape berry moth and other tortricids, black cutworm, spotted cutworm, other cutworms and other noctuids, diamondback moth, green cloverworm, velvetbean caterpillar, green cloverworm, pink bollworm, gypsy moth, and spruce budworm;
15	foliar feeding larvae and adults of the order <u>Coleoptera</u> including Colorado potato beetle, Mexican bean beetle, flea beetle, Japanese beetles, and other leaf beetles, boll weevil, rice water weevil, granary weevil, rice weevil and other weevil
-	pests, and soil inhabiting insects such as Western corn rootworm and other <u>Diabrotica spp.</u> , Japanese beetle, European chafer and other coleopteran grubs, and wireworms;
20	adults and larvae of the orders <u>Hemiptera</u> and <u>Homoptera</u> including tarnished plant bug and other plant bugs (<u>miridae</u>), aster leafhopper and other leafhoppers (<u>cicadellidae</u>), rice planthopper, brown planthopper, and other planthoppers (<u>fulgoroidea</u>), psylids, whiteflies (<u>aleurodidae</u>), aphids
25	(aphidae), scales (coccidae and diaspididae), lace bugs (tingidae), stink bugs (pentatomidae), cinch bugs and other seed bugs (lygaeidae), cicadas (cicadidae), spittlebugs (cercopids), squash bugs (coreidae), red bugs and cotton stainers (pyrrhocoridae);
30	adults and larvae of the order <u>acari</u> (mites) including
	European red mite, two spotted spider mite, rust mites, McDaniel mite, and foliar feeding mites;
35	adults and immatures of the order <u>Orthoptera</u> including grasshoppers;

5	adults and immatures of the order Diptera including
	leafminers, midges, fruit flies (<u>tephritidae</u>), and soil maggots;
	adults and immatures of the order Thysanoptera including
10	onion thrips and other foliar feeding thrips.
10	The compounds are also active against economically important
	livestock, household, public and animal health pests such as:
	insect pests of the order Hymenoptera including carpenter
15	ants, bees, hornets, and wasps;
	insect pests of the order <u>Diptera</u> including house flies, stable
	flies, face flies, horn flies, blow flies, and other muscoid fly
	pests, horse flies, deer flies and other Brachycera, mosquitoes,
20	black flies, biting midges, sand flies, sciarids, and other
	Nematocera;
	insect pests of the order Orthoptera including cockroaches and
25	crickets;
	insect pests of the order <u>Isoptera</u> including the Eastern
	subterranean termite and other termites;
	insect pests of the order Mallophaga and Anoplura including
30	the head louse, body louse, chicken head louse and other
	sucking and chewing parasitic lice that attack man and animals;
	insect pests of the order Siphonoptera including the cat flea,
35	dog flea and other fleas.
	•

10

15

20

25

30

35

3

The specific species for which control is exemplified are: fall armyworm, Spodoptera fruigiperda; tobacco budworm, Heliothis virescens; boll weevil, Anthonomus grandis; aster leafhopper, Macrosteles fascifrons; black bean aphid, (Aphis Fabae); southern corn rootworm, Diabrotica undecimpunctata. The pest control protection afforded by the compounds of the present invention is not limited, however, to these species. The compounds of this invention may also be utilized as rodenticides.

Application

Arthropod pests are controlled and protection of agronomic crops, animal and human health is achieved by applying one or more of the Formula I or II compounds of this invention, in an effective amount, to the environment of the pests including the agronomic and/or nonagronomic locus of infestation, to the area to be protected, or directly on the pests to be controlled. Because of the diversity of habitat and behavior of these arthropod pest species, many different methods of application are employed. A preferred method of application is by spraying with equipment that distributes the compound in the environment of the pests, on the foliage, animal, person, or premise, in the soil or animal, to the plant part that is infested or needs to be protected. Alternatively, granular formulations of these toxicant compounds can be applied to or incorporated into the soil. Other methods of application can also be employed including direct and residual sprays, aerial sprays, baits, eartags, boluses, foggers, aerosols, and many others. The compounds can be incorporated into baits that are consumed by the arthropods or in devices such as traps and the like which entice them to ingest or otherwise contact the compounds.

The compounds of this invention can be applied in their pure state, but most often application will be of a formulation comprising one or more compounds with suitable carriers, diluents, and surfactants and possibly in combination with a food depending on the contemplated end use. A preferred method of application involves spraying a water

WO 92/03421 PCT/US91/05334

262

ş

dispersion or refined oil solution of the compounds. Combinations with spray oils, spray oil concentrations, and synergists such as piperonyl butoxide often enhance the efficacy of the compounds of Formula I and II.

10

15

20

The rate of application of the Formula I and II compounds required for effective control will depend on such factors as the species of arthropod to be controlled, the pest's life cycle, life stage, its size, location, time of year, host crop or animal, feeding behavior, mating behavior, ambient moisture, temperature, etc. In general, application rates of 0.01 to 2 kg of active ingredient per hectare are sufficient to provide large-scale effective control of pests in agronomic ecosystems under normal circumstances, but as little as 0.001 kg/hectare or as much as 8 kg hectare may be required. For nonagronomic applications, effective use rates will range from about 1.0 to 50 mg/square meter but as little as about 0.1 mg/square meter or as much as 150 mg/square meter may be required.

The following Tests demonstrate the control efficacy of compounds of Formula I and II on specific pests; see Index Tables A through I for compound descriptions. Compounds not included in the tests had activities that gave <80% mortality or were not screened.

INDEX TABLE A

10	CMPD	Y	\mathbb{R}^1	R ²	<u>R</u> 3	<u>mp (°C)</u>
10	1	CH_2	CF ₃	5-Cl	CN	227-229
	2	CH_2	CF_3	CF ₃	CN	220-222
	3	CH_2	CF ₃	CF ₃	NHCO ₂ Me	271-272
	4	0	CF ₃	5-CF ₃	2-oxazolinyl	247-251
15	5	0	CF3	5-CF3	CN	265-269
	6	0	CF ₃	5-CF3	NHCO ₂ Me	>245
	7	0	CF ₃	5-CF3	2-Cl-thienyl	184-187
	8	0	CF_3	5-Cl	CH ₂ SiMe ₃	152-154
	9	0	CF3	5-Cl	epoxyethyl	187-190a
20					isomer 1	
	10	0	CF_3	5-Cl	epoxyethyl	79-81 ^a
					isomer 2	
	11	0	CF_3	H	epoxyethyl	159-162a
					isomer 1	
25	12	0	CF_3	H	epoxyethyl	135-138 ^a
					isomer 2	
	36	0	CF3	CF ₃	C(H)=NOCH	₃ 155-159

a. Two diastereoisomers were isolated after column chromatography of these compounds.

WO 92/03421 PCT/US91/05334

264

5

INDEX TABLE B

4	^
ı	61

10								
	CMPD	Y	<u>R</u> 1	<u>R</u> 2	<u>R</u> 19	<u>R</u> 32	Ω	mp (°C)
	13	O	CF_3	5-Cl	Н	H	3	208-209
	14	Ο	Cl	5-C1	\mathbf{H}	H	3	166-167
	15	0	CF_3	4-F	H	H	3	197-198
	16	Ο	Cl	4-F	H	H	3	175-177
	17	0	Cl	$5\text{-}\mathrm{CF}_3$	H	H	3	170-171
	18	0	CF_3	5-CF3	H	H	3	194-195
	19	0	$\mathbf{CF_3}$	5-C1	Me	H	3	69-71
•	20	Ο	$\mathbf{CF_3}$	5-C1	H	H	4	240-242
	21	Ο	\mathbf{Br}	5-C1	H	H	4	250-252
	22	Ο	$\mathbf{CF_3}$	5-C1	iPr	H	3	141-142
	23	Ο	CF_3	5-C1	${f Ph}$	H	3	semi-solid
	24	O	$\mathbf{CF_3}$	5-C1	H	${ m CO_2Me}$	3	196-199
	2 5	О	$\mathbf{CF_3}$	5 -CF $_3$	CO_2Me	H	3	190-192
	26	Ο	CF_3	5-C1	CO_2Me	H	3	114-117
	27	O	CF_3	5-CF3	Me	H	3	64-65
	28	Ο	$\mathbf{CF_3}$	$5\text{-}\mathrm{CF}_3$	i Pr	H	3	semi-solid
	29	O	CF_3	5-Cl	CH_2OH	H	3	115-117

PCT/US91/05334

265

5

INDEX TABLE C

$$R^{2} \xrightarrow{6} \xrightarrow{7} \xrightarrow{N} H$$

$$NHCNH \xrightarrow{\parallel} O$$

CMPD	<u>R</u> 1	\mathbb{R}^2	Q	<u>mp (°C)</u>
30	$\mathbf{CF_3}$	н	3	214-215
31	Br	H	3	221-223
32	Cl	H	3	230-233
33	CF ₃	5-C1	4	218-223
34	•	5-C1	4	232-243
35	Cl Br	5-C1	4	231-235

10

5

INDEX TABLE D

ā

7

ŧ

ò

267

5

INDEX TABLE E

 $R^1 = Cl$

CMPD	<u>G</u>	R_a^2	${\tt R}_{\tt b}^2$	\mathbb{R}^3	R ⁴	mp (°C)
53	G-2	Cl	H	Me	CO ₂ Me	174-177
54	G-2	Cl	H	4-F-Ph	H	191-193

10

268

5

INDEX TABLE F

<u>CMPD</u>	<u>G</u>	R ¹	R_a^2	R_b^2	R ³	mp (°C)
55	G-2	Cl	H	H	c-Hex	oil
56	G-2	Cl	Cl	H	c-Hex	oil
57	G-2	Cl	Cl	H	4-F-Ph	230-232
58	G-2	Cl	Cl	H	i-Pr	218-220
59	G-1	Cl	Cl ·	H	i-Pr	167-169
60	G-1	\mathbf{Br}	Cl	H	i-Pr	145-147
61	G-1	Cl	Cl	H	4-F-Ph	218-220
62	G-1	\mathbf{Br}	C1	H	4-F-Ph	210-212
63	G-1	CF_3	Cl	H	4-F-Ph	>250°C
64	G-2	Cl	Cl	H	4-Cl-Ph	>250°C
65	G-1	Cl	Cl	\mathbf{H}	4-Cl-Ph	214-217
66	G-1	CF ₃	Cl	H	4-Cl-Ph	250-252
67	G-2	C1	H	F	4-F-Ph	234-236

.

5

INDEX TABLE G

CMPD

 \mathbb{R}^1

mp (°C)

68

119-121

78

-s-()-F

194-196

10

5

INDEX TABLE H

<u>CMPD</u>	Y	R_a^2	R_{b}^{2}	\mathbb{R}^3	mp (°C)
69 70	COMe COMe	Cl Cl	H H	4-F-Ph 4-Cl-Ph	121-123 oil
71	COMe	H	\mathbf{F}	4-F-Ph	oil
72	$COCH_2CH_3$	Cl	H	4-F-Ph	oil
73	CO_2Me	Cl	H	4-F-Ph	oil

10

Ī

₹

ţ.

WO 92/03421 PCT/US91/05334

271

5

INDEX TABLE I

CMPD	R_a^2	$R_{\mathbf{b}}^{2}$	R3	<u>mp (°C)</u>
74	Cl	H	4-Cl-Ph	207-209
75	\mathbf{F}	H	4-F-Ph	201-204
76	Cl	H	n-Pr	218-220
77	H	${f F}$	4-F-Ph	224-226

10

7

Ē

ĵ

5 TEST A

Fall Armyworm

10

15

20

25

30

Test units, each consisting of an 8-ounce (230 mL) plastic cup containing a layer of wheat germ diet, approximately 0.5 cm thick, were prepared. Five third-instar larvae of fall armyworm (Spodoptera frugiperda) were placed into each cup. Solutions of each of the test compounds (acetone/distilled water 75/25 solvent) were sprayed into the cups. Spraying was accomplished by passing the cups, on a conveyer belt, directly beneath a flat fan hydraulic nozzle which discharged the spray at a rate of 0.5 pounds of active ingredient per acre (about 0.55 kg/ha) at 30 p.s.i. (207 kPa). The cups were then covered and held at 27°C and 50% relative humidity for 72 hours, after which time readings were taken. Of the compounds tested, the following gave mortality levels of 80% or higher: 1, 4, 5, 6, 7, 8, 10, 13, 14, 17, 18, 19, 20, 22, 23, 24, 26, 27, 28, 29, 30, 33, 36, 37, 38, 40, 51, 54, 55, 57, 58, 60, 61, 62, 63, 64, 65, 66, 67, 68, 69, 70, 72, 73, 74, 75, and 78.

TEST B

Tobacco Budworm

The test procedure of Test A was repeated for efficacy against third-instar larvae of the tobacco budworm (Heliothis virescens) except that mortality was assessed at 48 hours. Of the compounds tested, the following gave mortality level of 80% or higher: 1, 4, 5, 6, 7, 8, 9, 10, 13, 14, 15, 16, 17, 18, 19, 20, 22, 23, 24, 26, 27, 28, 33, 34, 36, 37, 38, 39, 40, 54, 57, 58, 59, 61, 62, 63, 64, 65, 66, 67, 68, 69, 70, 71, 72, 73, 74, 75 and 78.

5

10

15

>

TEST C

Southern Corn Rootworm

The units, each consisting of an 8-ounce (230 mL) plastic cup containing 1 sprouted corn seed, were prepared. Test units were sprayed as described in Test A with individual solutions of the test compounds. After the spray on the cups had dried, five third-instar larvae of the southern corn rootworm (Diabrotica undecimpunctata howardi) were placed into each cup. A moistened dental wick was inserted into each cup to prevent drying and the cups were then covered. The cups were then held at 27°C and 50% relative humidity for 48 hours, after which time mortality readings were taken. Of the compounds tested, the following gave mortality levels of 80% or higher: 1, 2, 4, 5, 6, 7, 8, 9, 11, 13, 14, 18, 19, 21, 22, 23, 28, 30, 31, 32, 33, 34, 35, 36, 37, 39, 54, 57, 58, 63, 64, 67, 68, 69, 71, 72, 73, 74, 75, 76, 77 and 78.

20

25

30

TEST D

Aster Leafhopper

Test units were prepared from a series of 12-ounce (350 mL) cups, each containing oat (Avena sativa) seedlings in a 1-inch (2.54 cm) layer of sterilized soil. The test units were sprayed as described in Test A with individual solutions of the below-listed compounds. After the oats had dried from the spraying, between 10 and 15 adult aster leafhoppers (Mascrosteles fascifrons) were aspirated into each of the covered cups. The cups were held at 27°C and 50% relative humidity for 48 hours, after which time mortality readings were taken. Of the compounds tested, the following gave mortality levels of 80% or higher: 1, 9, 10, 13, 14, 17, 18, 19, 20, 22, 23, 30, 31, 33, 36, 57, 58, 64, 67, 69, 70, 71, 72, 73 and 74.

1

Ø

274

5

TEST E

Boll Weevil

Five adult boll weevils (Anthonomus grandis grandis) were placed into each of a series of 9 ounce (260 mL) cups. The test procedure employed was then otherwise the same as in Test A. Mortality readings were taken 48 hours after treatment. Of the compounds tested, the following gave mortality levels of 80% or higher: 1, 2, 4, 5, 8, 9, 10, 13, 14, 16, 17, 19, 23, 26, 27, 30, 33, 36, 37, 38, 39, 57, 58, 64, 67, 69, 70, 71, 72, 73 and 74.

15

20

25

30

35

10

TEST F

Black Bean Aphid

Individual nasturtium leaves were infested with 5 to 10 aphids (all stages of <u>Aphis fabae</u>) and sprayed with their undersides facing up on a hydraulic sprayer as described in Test A. The leaves were then set in 1-inch diameter vials containing sugar water solution and covered with a clear plastic 1 oz - portion cup to prevent escape of aphids that drop from the leaves. The test units were held at 27°C and 50% relative humidity for 48 hours, after which time mortality readings were taken. Of the compounds tested, the following gave mortality levels of 80% or higher: 63 and 78.

TEST G

Two-Spotted Spider Mite

One inch squares of kidney bean leaves that have been infested on the undersides with 25 to 30 adult mites (<u>Tetranychus urtical</u>) were sprayed with their undersides facing up on a hydraulic sprayer as described in Test A. The leaf squares were placed underside up on a square of wet cotton in a petri dish and the perimeter of the leaf square was tamped down onto the cotton with forceps so that the mites cannot escape onto untreated leaf surface. The test units were held at 27°C and 50% relative humidity for 48 hours, after which time mortality readings

were taken. Of the compounds tested, Compounds 23 and 69 produced mortality levels of greater than 80%.

3

0

,

l

¥

į.

276

5

CLAIMS

What is claimed:

1. A compound selected from:

10



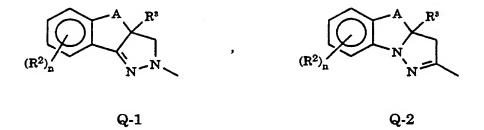
•

П

15 wherein:

Q is selected from the group

I



$$(\mathbb{R}^2)_n \xrightarrow{A \qquad \mathbb{R}^3} \mathbb{R}^3$$

$$\mathbb{N} - \mathbb{N}$$

$$\mathbb{Y}^1$$

$$(\mathbb{R}^2)_{\mathbf{n}} \longrightarrow \mathbb{N}$$

$$\mathbb{N}$$

$$\mathbb{N}$$

Q-4

$$(\mathbb{R}^2)_{\mathbf{n}} \xrightarrow{\mathbf{H}} \mathbb{N} - \mathbb{N}$$

Q-5

$$(\mathbb{R}^2)_n \xrightarrow[R^{18}]{V} \mathbb{R}^3$$

Q-6

$$(R^2)_n \xrightarrow{A \qquad R^3} R^3$$

$$N-N-$$

$$R^{18} \qquad H$$

Q-7

$$(\mathbb{R}^2)_n \xrightarrow{\mathbb{R}^3} \mathbb{R}^4$$

Q-8

5

١

١

Ţ

278

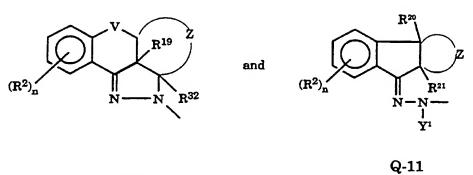
$$(\mathbb{R}^2)_n \xrightarrow[N]{\mathbb{R}^3} \mathbb{R}^4$$

5

Q-9

10

15



Q-10

A is selected from the group CH₂, CH₂CH₂, O, S, S(O), S(O)₂, N(R⁵), OCH₂, SCH₂, N(R⁵)CH₂, substituted CH₂, and substituted CH₂CH₂, the substituents independently selected from 1-2 halogen and methyl;

G is selected from the group

١

279

$$- \langle \bigcup_{N} - R^{1} \rangle - R^{1} \rangle - R^{1}$$

G-1 G-2 G-3

G-7 G-8 G-9

5 $-NH - R^{1} ;$ G-10

X is selected from the group O, S and N-X²; $X^{1} \text{ is selected from the group Cl, Br, OR}^{6}, SR^{6} \text{ and } N(R^{6})R^{7};$

 $\rm X^2$ is selected from the group $\rm R^6$, OH, OR⁶, CN, SO₂R⁶, SO₂Ph,

5	$OC(O)N(R^7)R^8$, $OC(O)OC_1-C_6$ alkyl, $N(R^7)R^8$ and phenyl
	optionally substituted with \mathbb{R}^9 ;
	Y and Y^1 are independently selected from the group H, C_1 - C_6
	alkyl, benzyl, C_2 - C_6 alkoxyalkyl, CHO, C_2 - C_6 alkylcarbonyl, C_2 - C_6 alkoxycarbonyl, C_2 - C_6 haloalkylcarbonyl, C_1 - C_6
10	alkylthio, C_1 - C_6 haloalkylthio, phenylthio, $R^{10}OC(O)N(R^{11})S$ -and $R^{12}(R^{13})NS$ -;
	${ m R^{1},R^{2}}$ and ${ m R^{14}}$ are independently selected from the group H,
	C ₁ -C ₆ alkyl, C ₁ -C ₆ haloalkyl, C ₂ -C ₆ alkenyl, C ₂ -C ₆
	haloalkenyl, C ₂ -C ₆ alkynyl, C ₃ -C ₆ haloalkynyl, C ₂ -C ₆
15	alkoxyalkyl, C ₂ -C ₆ alkylthioalkyl, C ₁ -C ₆ nitroalkyl, C ₂ -C ₆ cyanoalkyl, C ₃ -C ₈ alkoxycarbonylalkyl, C ₃ -C ₆ cycloalkyl, C ₃ -C ₆
	halocycloalkyl, halogen, CN, N ₃ , SCN, NO ₂ , OR ¹⁵ , SR ¹⁵ ,
	$S(O)R^{15}$, $S(O)_2R^{15}$, $OC(O)R^{15}$, $OS(O)_2R^{15}$, CO_2R^{15} , $C(O)R^{15}$,
	$C(O)N(R^{15})R^{16}$, $SO_2N(R^{15})R^{16}$, $N(R^{15})R^{16}$, $N(R^{16})C(O)R^{15}$,
20	$OC(O)NHR^{15}$, $N(R^{16})C(O)NHR^{15}$, $N(R^{16})SO_2R^{15}$, phenyl
	optionally substituted with 1 to 3 substituents independently selected from W and benzyl optionally substituted with 1 to 3 substituents independently selected from W; or when m, n or p is 2, $(R^1)_2$ can be taken together, or $(R^2)_2$ can be taken together
25	or $(R^{14})_2$ can be taken together as -OCH ₂ O, -OCF ₂ O,
20	OCH ₂ CH ₂ O, CF ₂ CF ₂ O, -CH ₂ C(CH ₃) ₂ O- or -OCF ₂ CF ₂ O- to
	form a cyclic bridge between adjacent atoms on the same ring;
	R^3 is selected from the group R^a , R^b and J ;
	R^a is selected from the group H, C_1 - C_6 alkyl, C_1 - C_6 haloalkyl,
30 .	C ₂ -C ₆ alkenyl, C ₂ -C ₆ haloalkenyl, C ₂ -C ₆ alkynyl, OR ³⁵ , C ₂ -C ₆ alkoxyalkyl, C ₂ -C ₆ cyanoalkyl, C ₃ -C ₈ alkoxycarbonylalkyl, CO ₂ R ¹⁵ , C(O)R ¹⁵ , C(O)N(R ¹⁵)R ¹⁶ , C(S)N(R ¹⁵)R ¹⁶ , C(S)R ¹⁵ , C(S)SR ¹⁵ , phenyl, phenyl substituted with $(R^{14})_p$, benzyl and

¥

,

j

5	benzyl substituted with 1 to 3 substituents independently selected from W;
	R^b is selected from the group CN, N3, NO2, halogen, $N(R^{22})R^{23}$,
10	$C(R^{33})=N-O-R^{34}$, C_2-C_6 epoxyalkyl optionally substituted with a group selected from C_1-C_3 alkyl, phenyl optionally substituted with W, $C(O)R^{24}$, $Si(R^{28})(R^{29})$ R^{30} , SR^{27} , $S(O)R^{27}$, SO_2R^{27} and $-P(O)(OR^{27})_2$, CO_2R^{24} ; or R^b is C_1-C_6 alkyl substituted with a group selected from $C(O)N(R^{25})R^{26}$, $C(O)R^{25}$, SR^{27} ,
	$S(O)R^{27}$, SO_2R^{27} , SCN , C_1 - C_2 haloalkoxy, $Si(R^{28})(R^{29})R^{30}$, $N(R^{22})R^{23}$, NO_2 , $OC(O)R^{25}$, $Si(R^{28})(R^{29})R^{30}$, - $P(O)(OR^{27})_2$
15	and J;
	J is selected from the group consisting of saturated, partially
20	saturated and aromatic 5- or 6-membered substituted or unsubstituted heterocyclic rings, bonded through carbon or nitrogen, containing 1-4 heteroatoms independently selected from the group consisting of 0-2 oxygen, 0-2 sulfur and 0-4 nitrogen and optionally containing one carbonyl wherein the substituent groups are selected from W;
I	R4 is selected from the group H, C1-C6 alkyl, C1-C6 haloalkyl,
2 5	C_2 - C_6 alkenyl, C_2 - C_6 haloalkenyl, C_2 - C_6 alkynyl, C_3 - C_6 haloalkynyl, C_2 - C_6 alkoxyalkyl, C_2 - C_6 cyanoalkyl, phenyl optionally substituted with $(R^{14})_p$ and benzyl optionally
	substituted with 1 to 3 substituents independently selected from W;
	$ m R^{5}$ is selected from the group H, C ₁ -C ₄ alkyl, C ₁ -C ₄ haloalkyl,
30	$\begin{array}{l} \text{C}_2\text{-C}_4 \text{ alkenyl, } \text{C}_2\text{-C}_4 \text{ haloalkenyl, } \text{SR}^{15}, \text{S(O)R}^{15}, \text{S(O)}_2\text{R}^{15}, \\ \text{C(O)R}^{15}, \text{CO}_2\text{R}^{15}, \text{C(O)N}(\text{R}^{15})\text{R}^{17}, \text{C(S)N}(\text{R}^{15})\text{R}^{17}, \text{C(S)R}^{15}, \\ \text{C(S)OR}^{15}, \text{P(O)}(\text{OR}^{15})_2, \text{P(S)}(\text{OR}^{15})_2, \text{P(O)}(\text{R}^{17})\text{OR}^{15}, \end{array}$
	$P(O)(R^{15})(SR^{17})$, and optionally substituted phenyl and

Б	selected from W;	
	${ m R}^6$ is selected from the group ${ m C}_1 ext{-}{ m C}_3$ alkyl, benzyl optionally	
10	substituted with R^9 , C_2 - C_4 alkenyl, C_2 - C_4 alkynyl, C_1 - C_3 haloalkyl, C_2 - C_4 haloalkenyl, C_3 - C_6 cycloalkyl and C_1 - C_3 alkyl substituted with OCH ₃ , OCH ₂ CH ₃ , NO ₂ , CN, CO ₂ CH ₃ , CO ₂ CH ₂ CH ₃ , SCH ₃ or SCH ₂ CH ₃ ;	*
	${ m R}^7$ is selected from the group H, C ₁ -C ₄ alkyl, C ₁ -C ₄ haloalkyl,	
	$C_2\text{-}C_4$ alkoxycarbonyl, optionally substituted phenyl and	
15	optionally substituted pyridinyl wherein the substituent(s) are selected from \mathbb{R}^9 ; or	
	${ m R}^6$ and ${ m R}^7$ when attached to the same atom can be taken together	
	to form $-(CH_2)_4$ -, $-(CH_2)_5$ - or $-CH_2CH_2OCH_2CH_2$ - each of which is optionally substituted with 1 or 2 CH ₃ groups;	
	${ m R}^{8}$ is selected from the group H and ${ m C}_{1} ext{-}{ m C}_{4}$ alkyl; or	
20	${ m R}^7$ and ${ m R}^8$ can be taken together to form -(CH2)4-, -(CH2)5- or	
	-CH ₂ CH ₂ OCH ₂ CH ₂ - each of which is optionally substituted with 1 or 2 CH ₃ groups;	
	${ m R}^{9}$ is selected from the group halogen, CN, ${ m C}_{1} ext{-}{ m C}_{3}$ haloalkyl and	
	C ₁ -C ₃ haloalkoxy;	
2 5	R^{10} is C_1 - C_6 alkyl;	
	R^{11} is C_1 - C_4 alkyl;	
	$ m R^{12}$ and $ m R^{13}$ are independently $ m C_1\text{-}C_4$ alkyl; or	
	${ m R}^{12}$ and ${ m R}^{13}$ when attached to the same atom, can be taken	
	together as -CH2CH2CH2CH2CH2- or -CH2CH2OCH2CH2-;	.4
30	R ¹⁵ is selected from the group H, C ₁ -C ₆ alkyl, C ₁ -C ₆ haloalkyl.	Ĉ

5	C2-C6 alkenyl, C2-C6 haloalkenyl, C2-C6 alkynyl, C3-C6 haloalkynyl, C2-C6 alkoxyalkyl, C2-C6 alkylthioalkyl, C1-C6 nitroalkyl, C2-C6 cyanoalkyl, C3-C8 alkoxycarbonylalkyl, C3-C6 cycloalkyl, C3-C6 halocycloalkyl, optionally substituted phenyl
10	and optionally substituted benzyl wherein the substituent(s) are 1 to 3 substituents independently selected from W; wherein R^{15} is other than H when: (i) R^1 , R^2 , or R^{14} is $S(O)R^{15}$, SO_2R^{15} , $OC(O)R^{15}$ or $OS(O)_2R^{15}$, (ii) R^a is $C(S)R^{15}$ or $C(S)SR^{15}$, or (iii) R^5 is not $C(O)R^{15}$, $C(O)N(R^{15})R^{17}$ or $C(S)N(R^{15})R^{17}$;
	$ m R^{16}$ is selected from the group H and $ m C_1\text{-}C_4$ alkyl; or
15	$ m R^{15}$ and $ m R^{16}$, when attached to the same atom, can be taken together as -(CH ₂) ₄ -, -(CH ₂) ₅ -, or -CH ₂ CH ₂ OCH ₂ CH ₂ -;
	$ m R^{17}$ is selected from the group H, $ m C_1\text{-}C_4$ alkyl, $ m C_2\text{-}C_4$ alkenyl and
	C ₂ -C ₄ alkynyl;
	R18 is selected from the group H, C1-C4 alkyl, C1-C4 haloalkyl,
20 25	C ₂ -C ₄ alkylcarbonyl, C ₂ -C ₄ haloalkylcarbonyl, C ₂ -C ₄ alkoxycarbonyl, C ₂ -C ₄ haloalkoxycarbonyl, C ₂ -C ₅ alkylaminocarbonyl, C ₃ -C ₆ cycloalkyl, C ₃ -C ₆ halocycloalkyl, C ₄ -C ₇ alkylcycloalkyl, C ₄ -C ₇ haloalkylcycloalkyl, C ₁ -C ₄ alkylsulfonyl, C ₁ -C ₄ haloalkylsulfonyl and SO ₂ Ph optionally substituted with Cl, Br or CH ₃ ;
	$ m R^{19}, R^{20}$ and $ m R^{21}$ are independently selected from the group H,
	C_1 - C_3 alkyl, CO_2R^{15} , phenyl and phenyl substituted by W;
	R^{22} is selected from the group H, C((O)C ₁ -C ₆ alkyl, CO ₂ C ₁ -C ₆
30	alkyl, optionally substituted C_1 - C_4 alkyl, optionally substituted C_2 - C_4 alkenyl, and optionally substituted C_2 - C_4 alkynyl, the substituents selected from C_1 - C_2 alkoxy, CN , $C(O)R^{31}$ and $C(O)_2R^{27}$;

5	\mathbb{R}^{23} is selected from the group H, C ₁ -C ₃ alkyl, phenyl, phenyl
	substituted with W, benzyl and benzyl substituted with W;
	$ m R^{24}$ is selected from the group H, $ m C_1$ - $ m C_4$ alkyl $ m C_2$ - $ m C_4$ alkenyl and
	C ₂ -C ₄ alkynyl;
	$ m R^{25}$ and $ m R^{26}$ are independently selected from the group H and
10	$\mathrm{C_{1}\text{-}C_{2}}$ alkyl;
	${ m R}^{27}$ is selected from the group ${ m C}_1\text{-}{ m C}_3$ alkyl, phenyl and phenyl
	substituted with W;
	\mathbb{R}^{28} is \mathbb{C}_1 - \mathbb{C}_3 alkyl;
	R^{29} is C_1 - C_3 alkyl;
15	${ m R}^{30}$ is selected from the group C1-C3 alkyl, phenyl and phenyl
	substituted with W;
	R^{31} is selected from the group H, $\mathrm{C}_1\text{-}\mathrm{C}_3$ alkyl, phenyl and phenyl
	substituted by W;
	$ m R^{32}$ is selected from the group H, CH $_3$ and CO $_2$ CH $_3$
20	$ m R^{33}$ is selected from the group H, Cl, C ₁ -C ₄ alkyl, C ₁ -C ₂ alkoxy,
	C ₁ -C ₂ thioalkyl and CN;
	$ m R^{34}$ is selected from the group H, $ m C_1\text{-}C_4$ alkyl, $ m C_2\text{-}C_3$
	alkylcarbonyl and C2-C3 alkoxycarbonyl;
	$ m R^{35}$ is selected from the group H, C ₁ -C ₄ alkyl, C ₂ -C ₄ alkenyl,
2 5	C2-C4 alkynyl, C2-C4 alkylcarbonyl, C2-C4 alkoxycarbonyl, C2-
	C_4 alkylaminocarbonyl, C_2 - C_4 dialkylaminocarbonyl, and C_1 - C_4 alkylsulfonyl;

١

)

V is selected from the group O, S, N(R⁵), CH₂ and CH₂ substituted with 1-2 CH₃;

W is selected from the group halogen, CN, NO₂, C₁-C₂ alkyl, C₁-C₂ haloalkyl, C₁-C₂ alkoxy, C₁-C₂ haloalkoxy, C₁-C₂ alkylthio, C₁-C₂ haloalkylthio, C₁-C₂ alkylsulfonyl and C₁-C₂ haloalkylsulfonyl;

Z is selected from the group (CH₂)_q, -CH₂OCH₂-, -CH₂SCH₂- and -CH₂NHCH₂-, where said groups can be optionally substituted with 1-2 CH₃;

m is 1 to 3;

15 n is 1 to 3;

p is 1 to 3;

q is 2 to 4;

u is 1 or 2;

provided that (i) when Q is Q-3, R^3 is other than CN, $N(R^{22})R^{23}$, SR^{27} , $S(O)R^{27}$ or SO_2R^{27} , (ii) when Q is Q-4, R^3 is other than CN or J, and (iii) when G is G-9, R^3 is other than R^a .

2. A compound according to Claim 1 wherein J is selected from the group:

25



J-1



J-2



J-3



J-5



J-4



J-7



$$\bigcup_{N}^{N}$$

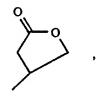


J-10





J-13



J-16

J-18

,

Ĩ

10

15

)

1

3. A compound according to Claim 2 wherein:

R¹ is selected from the group H, halogen, CN, SCN, NO₂, OR¹⁵, SR¹⁵, SO₂R¹⁵, CO₂R¹⁵, C(O)R¹⁵, C₁-C₆ alkyl, C₁-C₆ haloalkyl, C₂-C₆ alkenyl, C₂-C₆ haloalkenyl, C₂-C₆ alkynyl, C₃-C₆ haloalkynyl, C₂-C₆ alkoxyalkyl, C₂-C₆ alkylthioalkyl, C₁-C₆ nitroalkyl, C₂-C₆ cyanoalkyl, C₃-C₈ alkoxycarbonylalkyl, C₃-C₆ cycloalkyl, C₃-C₆ halocycloalkyl, phenyl optionally substituted with 1 to 3 substituents independently selected from W, and benzyl optionally substituted with 1 to 3 substituents independently selected from W;

R² and R¹⁴ are independently selected form the group H, halogen, CN, SCN, NO₂, OR¹⁵, SR¹⁵, S(O)₂R¹⁵, OC(O)R¹⁵, OS(O)₂R¹⁵, CO₂R¹⁵, C(O)R¹⁵, C(O)N(R¹⁵)R¹⁶, SO₂N(R¹⁵)R¹⁶, N(R¹⁵)R¹⁶, C₁-C₆ alkyl, C₁-C₆ haloalkyl, C₂-C₆ alkenyl, C₂-C₆ alkylthioalkyl, C₃-C₆ haloalkynyl, C₂-C₆ alkoxyalkyl, C₂-C₆ alkylthioalkyl, C₁-C₆ nitroalkyl, C₂-C₆ cyanoalkyl, C₃-C₈ alkoxycarbonylalkyl, C₃-C₆ cycloalkyl, C₃-C₆ halocycloalkyl, phenyl optionally substituted with 1 to 3 substituents independently selected from W, and benzyl optionally substituted with 1 to 3 substituents independently selected from W;

 \mathbb{R}^3 is selected from the group \mathbb{R}^a , \mathbb{R}^b and J;

Đ	alkoxycarbonylalkyl, CO_2R^{15} , $C(O)R^{15}$, phenyl and phenyl substituted by $(R^{14})_p$;
10	R^b is selected from the group CN, $C(R^{33})=NOR^{34}$, C_2 - C_6 epoxyalkyl and SR^{27} ; or R^b is C_1 - C_4 alkyl substituted with a group selected from $C(O)R^{25}$, SR^{27} , $S(O)R^{27}$, SO_2R^{27} , C_1 - C_2 haloalkoxy and $Si(R^{28})(R^{29})R^{30}$;
	$ m R^4$ is selected from the group H, C ₁ -C ₃ alkyl, C ₂ -C ₄ alkenyl and C ₂ -C ₄ alkynyl;
	R^{18} is H;
1 5	X is O;
	X^1 is selected from the group Cl, OR^6 , SR^6 and $N(CH_3)_2$;
	X^2 is selected from the group R^6 , OR^6 and $N(CH_3)_2$;
	A is selected from the group O, CH_2 , OCH_2 , S, CH_2CH_2 , SCH_2 , $N(R^5)CH_2$ and $N(R^5)$;
20	$ m R^{5}$ is selected from the group H, $ m C_{1}\text{-}C_{4}$ alkyl, $ m S(O)_{2}R^{15}$, $ m CO_{2}R^{15}$ and $ m CONHR^{15}$;
	$ m R^{15}$ is selected from the group C1-C4 alkyl, C1-C2 haloalkyl, C3-C4 alkenyl and propargyl;
	R^{16} is selected from H and CH ₃ ;
25	\mathbb{R}^{27} is $\mathbb{C}_{1}\text{-}\mathbb{C}_{3}$ alkyl;
	R^{33} is selected from the group H and CH_3 ;
	R^{34} is selected from the group H and C_1 - C_3 alkyl;
	V is selected from the group O, $N(R^5)$ and CH_2 ;
30	W is selected from the group Cl, F, Br, CN, CF ₃ , C ₁ -C ₂ alkyl, C ₁ -C ₂ alkoxy, OCF ₂ H, OCF ₃ and NO ₂ ;

Y is selected from the group H, CH₃, C(O)CH₃ and CO₂CH₃;
Z is -CH₂CH₂- or -CH₂CH₂-;

m is 1 or 2;

n is 1 or 2; and

p is 1.

10

30

1

- 4. A compound according to Claim 3 wherein: G is selected from the group G-2, G-3, G-7 and G-9.
- 5. A compound according to Claim 4 wherein:

 J is selected from the group J-1, J-2, J-8, J-9 and J-16.
 - 6. A compound according to Claim 1 of Formula I.
- 7. A compound according to Claim 6 wherein Q is Q-1 and A is selected from the group CH₂, OCH₂, CH₂CH₂, SCH₂ and N(R⁵)CH₂.
 - 8. A compound according to Claim 6 wherein Q is Q-2 and A is selected from the group CH_2 , OCH_2 , CH_2CH_2 , SCH_2 and $N(R^5)CH_2$.
- 9. A compound according to Claim 6 wherein Q is Q-3 and A is selected from the group CH₂, OCH₂, CH₂CH₂, SCH₂ and N(R⁵)CH₂.
 - 10. A compound according to Claim 6 wherein Q is Q-4 and A is selected from the group CH_2 , OCH_2 , CH_2CH_2 , SCH_2 and $N(R^5)CH_2$.
 - 11. An arthropodicidal composition comprising an arthropodicidally effective amount of a compound according to any one of Claims 1 to 10 and a carrier therefor.

5 12. A method for controlling arthropods comprising applying to them or to their environment an arthropodicidally effective amount of a compound according to any one of Claims 1 to 10.

* -